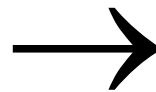
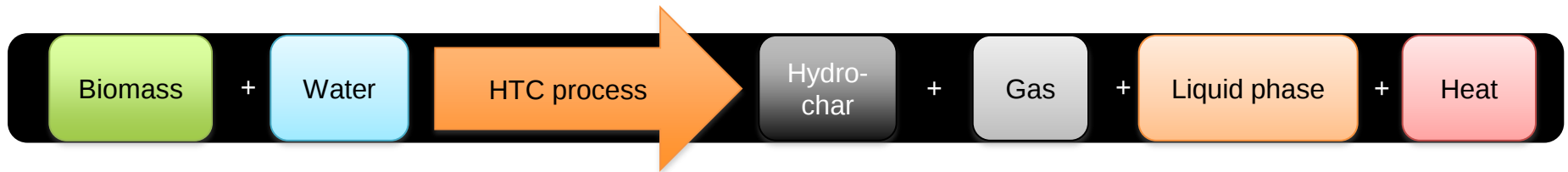


Porous hydrochars from hydrothermal carbonization of biomasses

Prof. Dr. Michael Wark

Dr. Hendrik Multhaupt, Julia Lüttmann, Manuel Nowotny, Frank Schröter, Tim Worriescheck -
Technische Chemie 1, Carl von Ossietzky Universität Oldenburg



T about 200 ° C
p ca. 20 bar



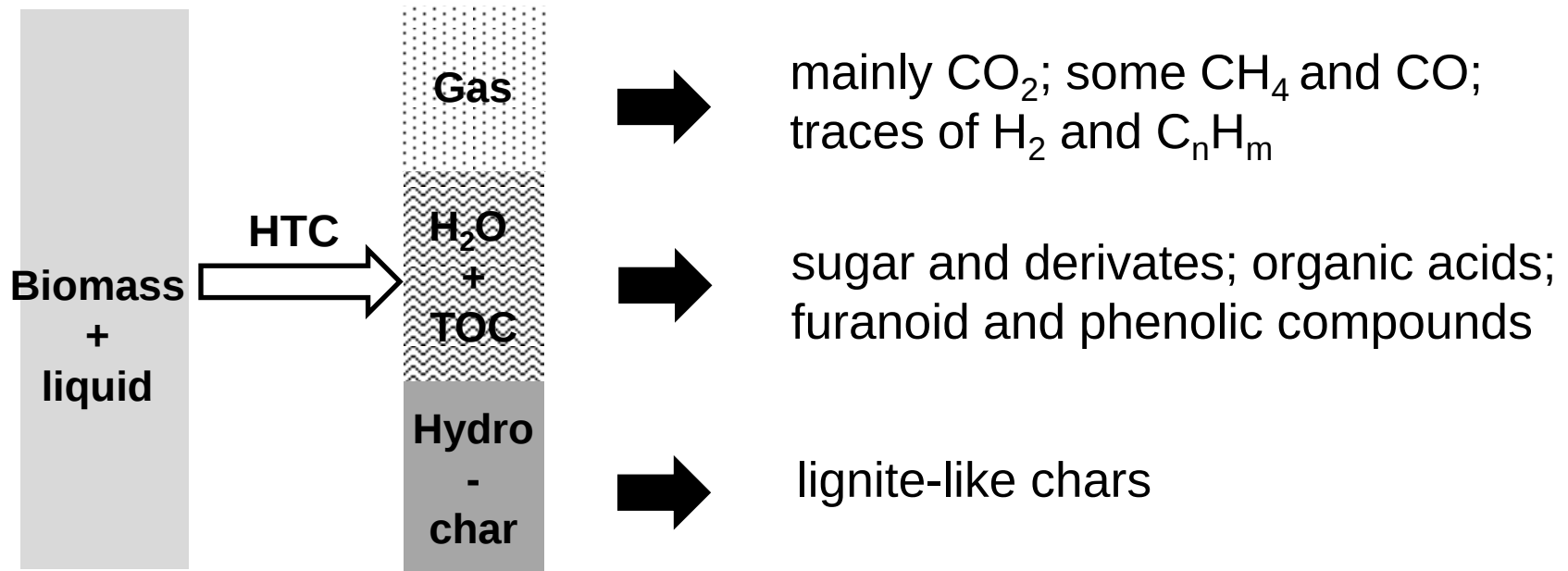
1 h

3 h

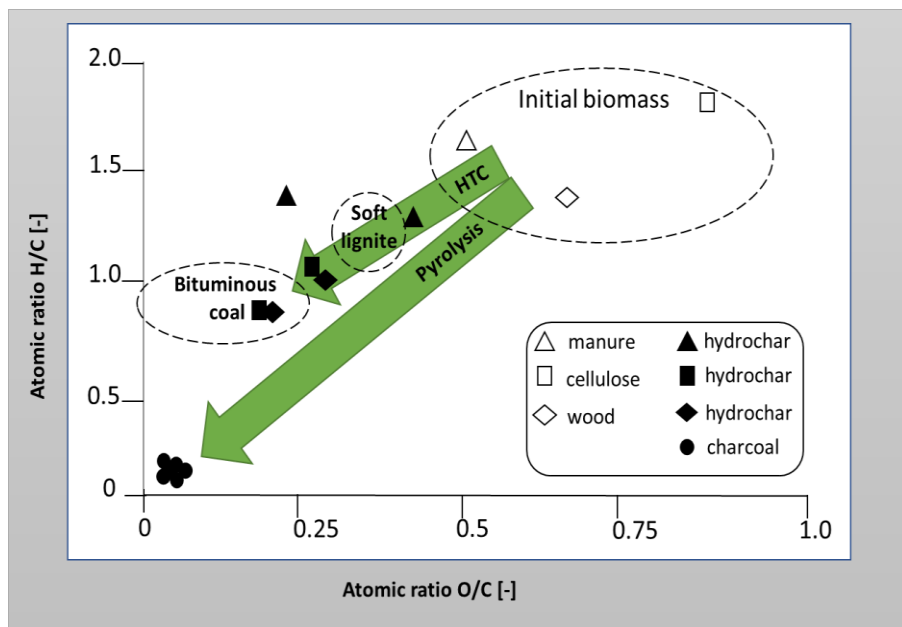
6 h

Fotos: J. Pfingstmann

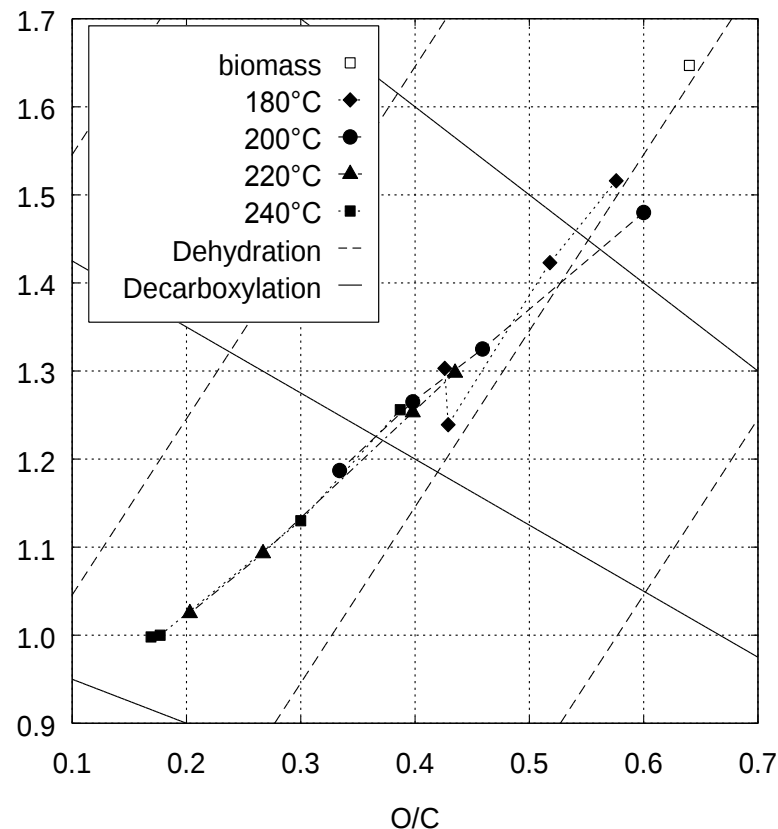
- transformation of biomasses and **wet substrates** into biochar
- heating of biomass in aqueous suspensions to **180 – 250 °C** under elevated pressure
- chars with **high amount of functional groups** and small surface area ($< 10 \text{ m}^2/\text{g}$)



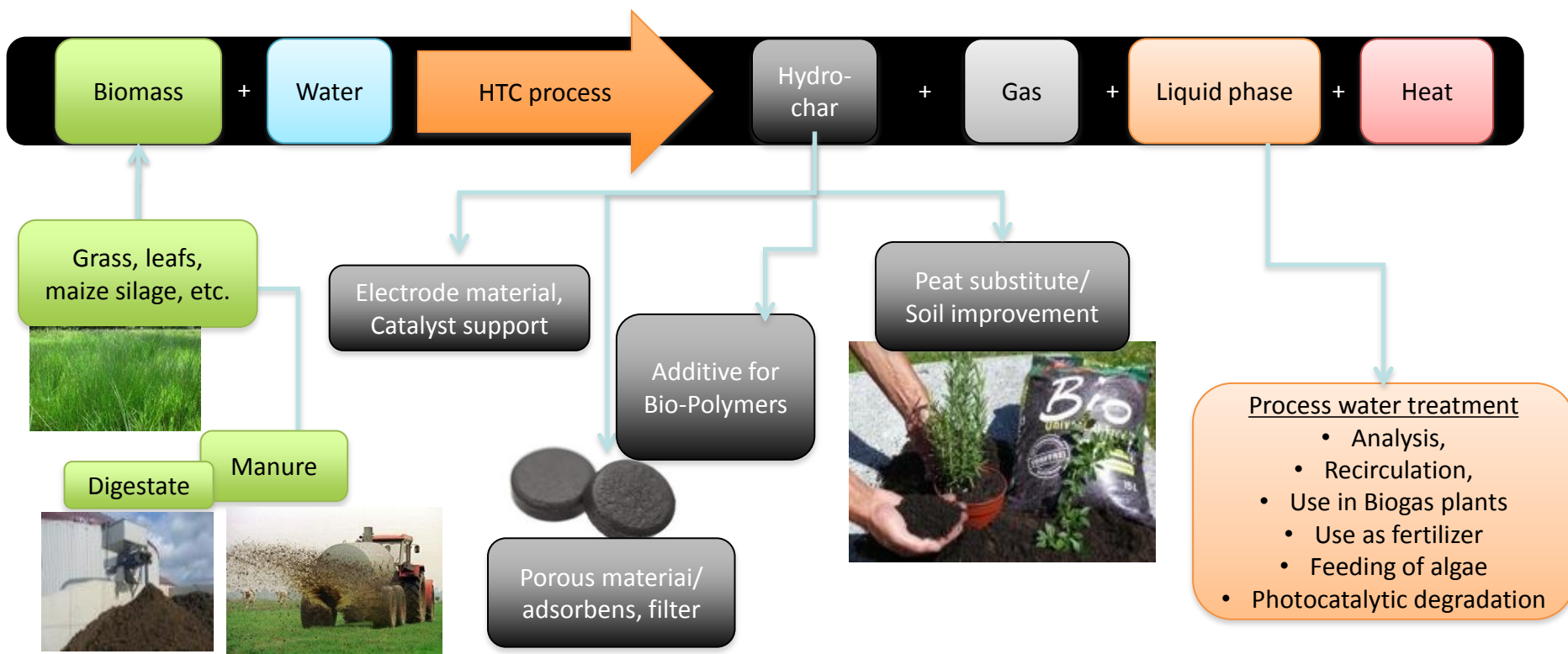
- Pyrolysis takes place at temperatures $> 300\text{ }^{\circ}\text{C}$
 - lower H/C ratio
 - high aromatic content
- HTC takes place under pressure ($\sim 20\text{ bar}$) and moderate temperature ($\sim 200\text{ }^{\circ}\text{C}$)
 - higher O/C ratio
 - high functionality



HTC of landscape management biomass



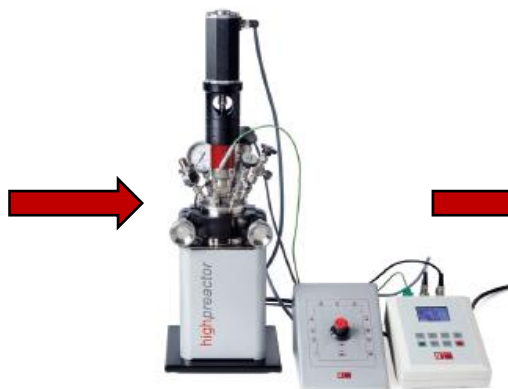
M. Röhrdanz, T. Rebling, J. Ohlert, J. Jasper, T. Greve, R. Buchwald, P. von Frieling, M. Wark, Journal of Environmental Management 173 (2016), 72-78, "Hydrothermal carbonization of biomass from landscape management - Influence of process parameters on soil properties of hydrochars"



Biomass, water,
additive



HTC



Processing



Hydrochars



Glucose, HMF,
Cellulose, Lignin,
Grasses, Coconut shell,
Rice husks, Dandelion
leaves, Algae, Sawdust

180-220 °C
4-36 h
200-2000 mL

Washing, drying, milling

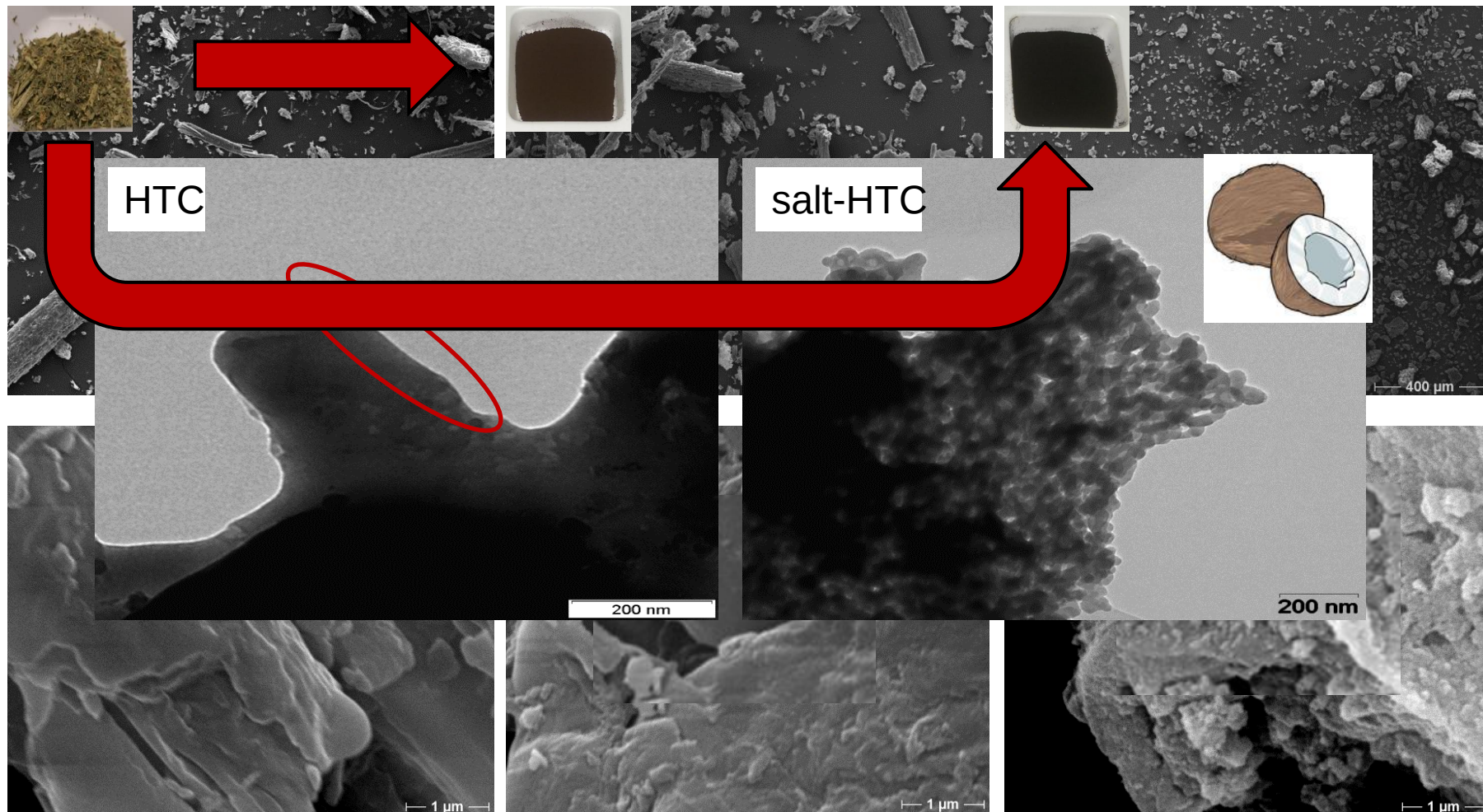
EA, SEM/TEM,
XRD, SS-MAS-
NMR, FTIR, XPS,
gas sorption,
TGA, ...

LiCl, NaCl, KCl, ZnCl₂,
MgCl₂, Na₂SO₄, Na₂CO₃,
...

Biomass (grass mixture)

Hydrochar

Salt-Hydrochar

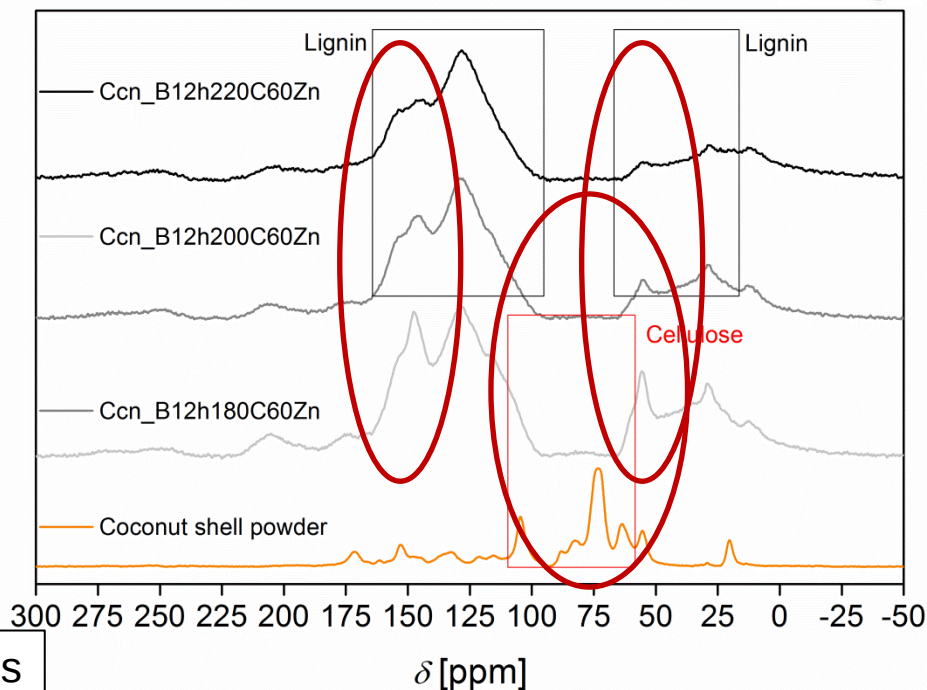
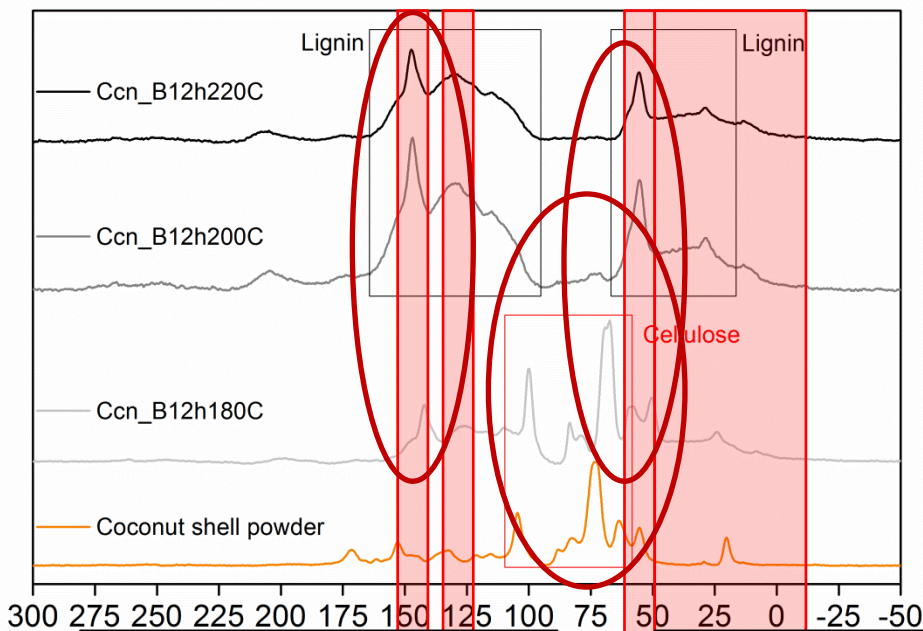




HTC

^{13}C -solid state MAS-NMR spectra

Salt-HTC



aromatic groups,
bound to lignin

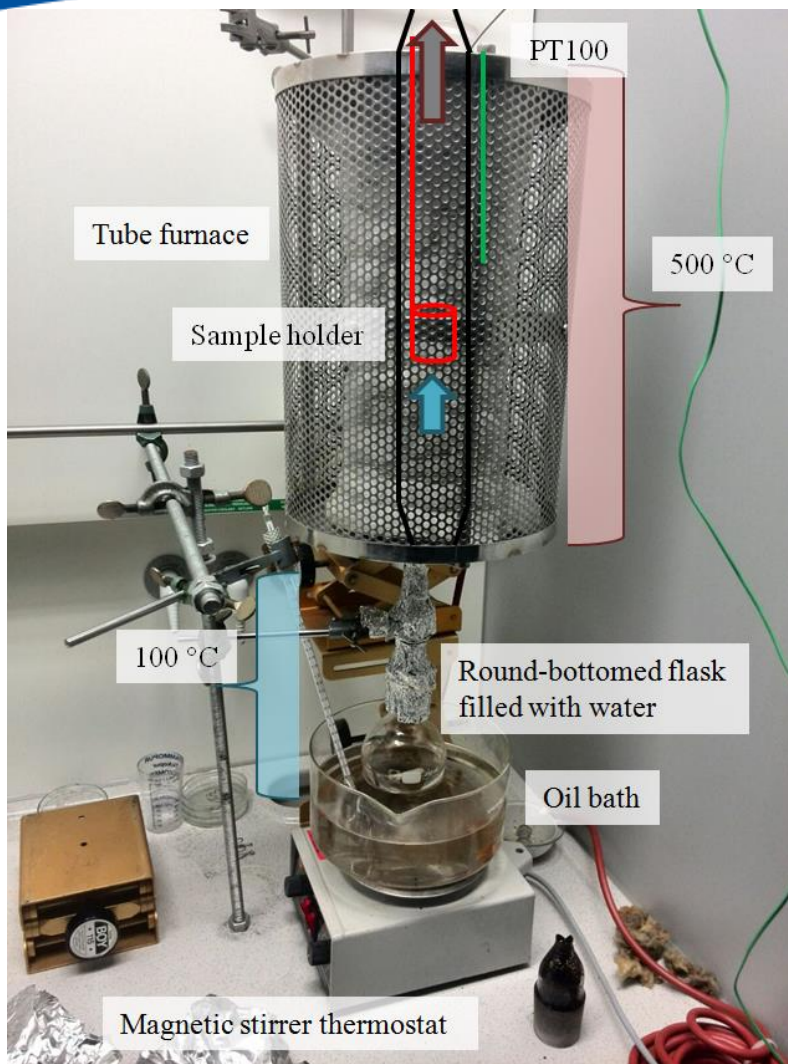
sp^2 -carbon

aliphatic groups

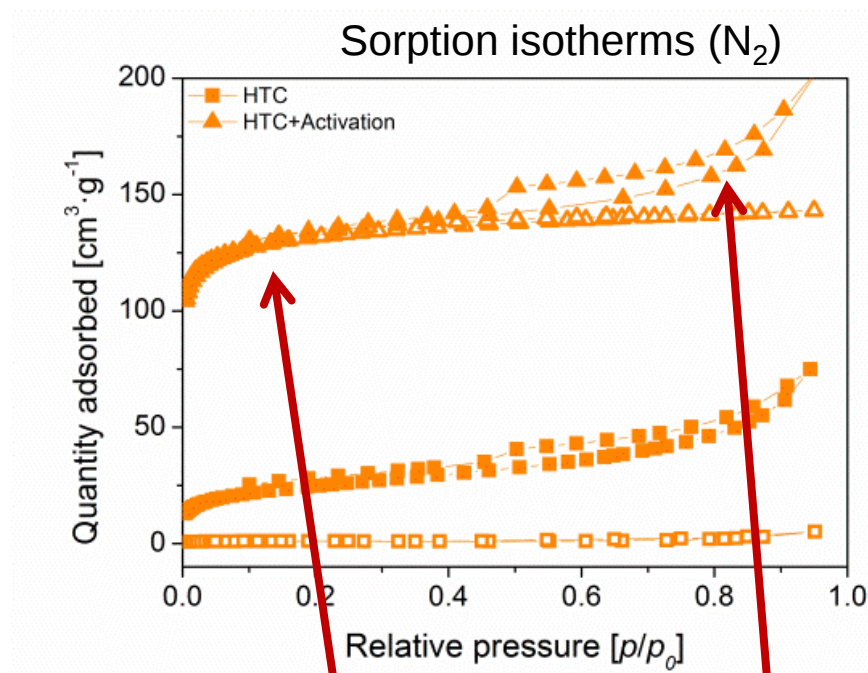
methoxy groups in lignin

Effects of salt addition:

- Cellulose degradation at lower temperatures
- Degradation of lignin: Start at decreased temperature



- Sample in crucible
- Tube furnace (e.g. 500 ° C)
- Water vapor flows through the oven
- Yield: 20-30%



HTC, steam activated: microporous
Salt-HTC, steam act.: micro- and mesoporous

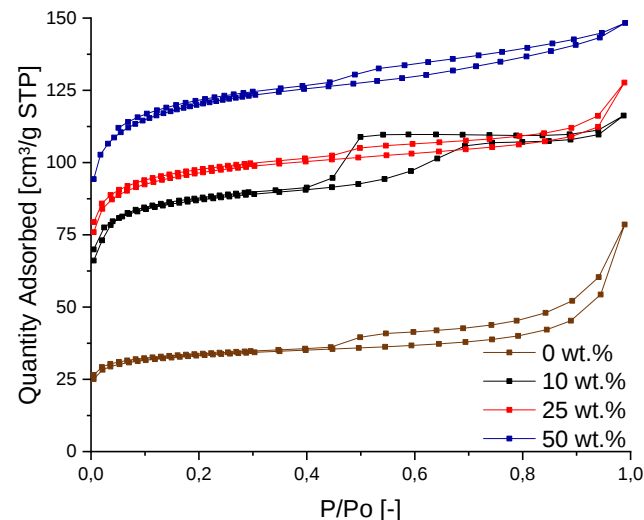
Mixtures of pig manure and coconut shell

Biomass [-]	Reaction time [h]	Amount of Coconut shell [wt.%]	Surface area [m ² /g]
Manure	1	0	143
		10	294
		25	401
		50	364
Manure	4	0	125
		10	335
		25	365
		50	454

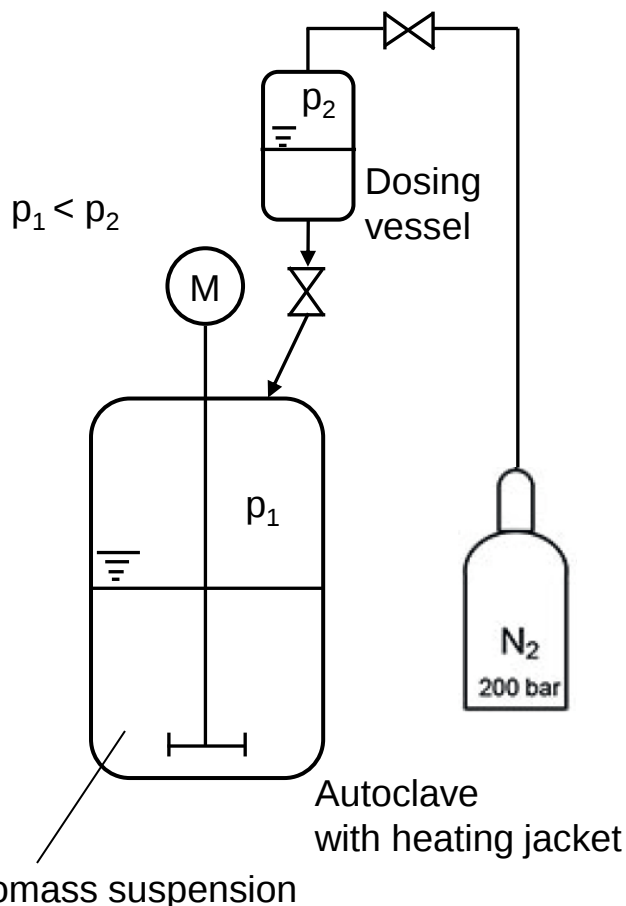
Hydrochars from mixtures of soft and hard biomass:

- Higher surface areas (N₂ sorption) than hydrochars from HTC of pure pig manure
- Comparable properties to hydrochars obtained with salts as additives

4 h ; 220 °C



Addition of solutions during HTC

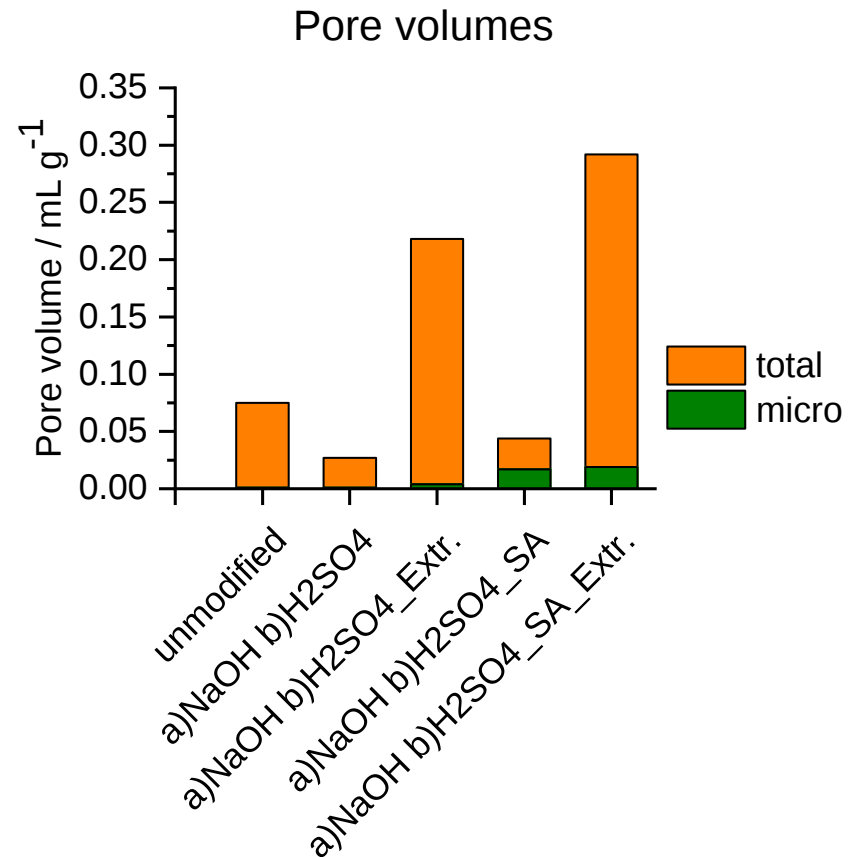
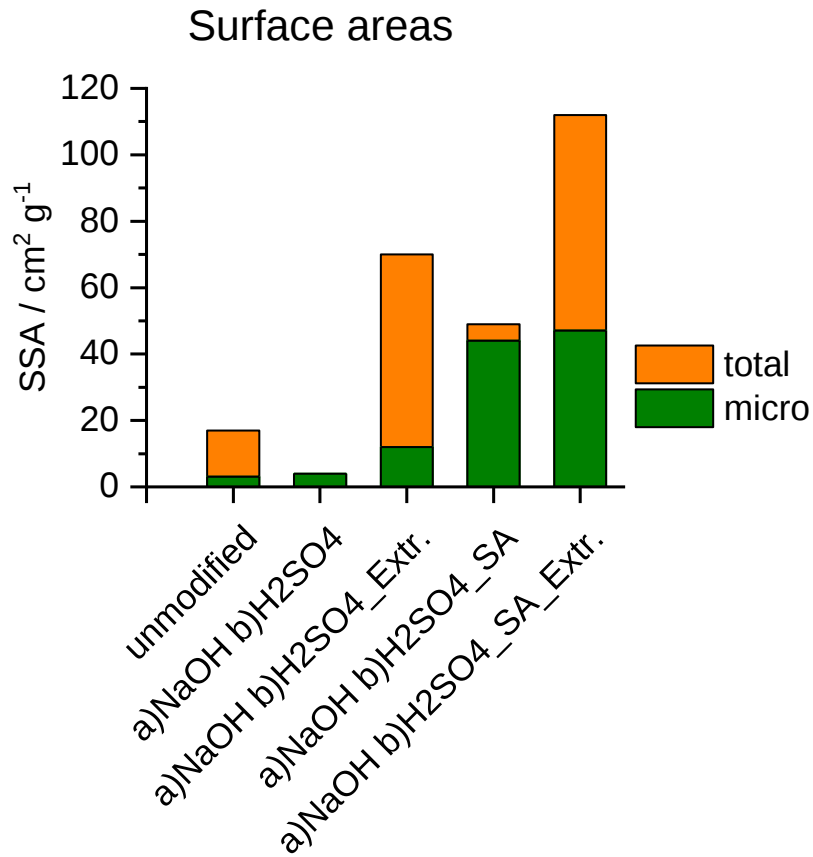


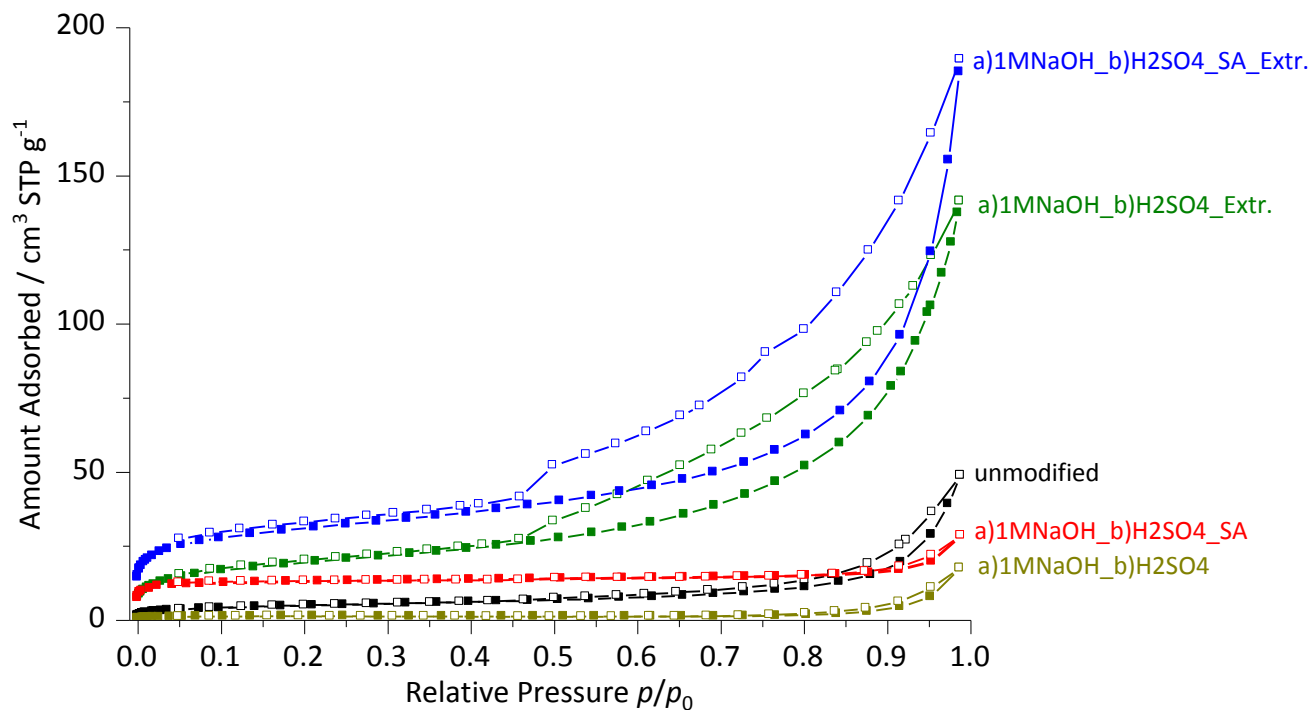
- Making use of sand (SiO₂ source) – being present in the biomass – for creating porosity (as hard template)
 - HTC under alkaline and acidic conditions via addition of NaOH and H₂SO₄
- Alkaline and acidic one-pot HTC

Experiments

Abbreviation	Description
SA	Subsequent Steam activation, 650 °C, 1h
Extr.	Subsequent Suspension in 0.5 M NaOH solution overnight
NaOH	HTC in 1 M NaOH
H ₂ SO ₄	HTC in 0.5 M H ₂ SO ₄
a) NaOH b) H ₂ SO ₄	HTC in 1 M NaOH, addition of 1 equiv. H ₂ SO ₄ after 9.5 h, cooling down after 12 h

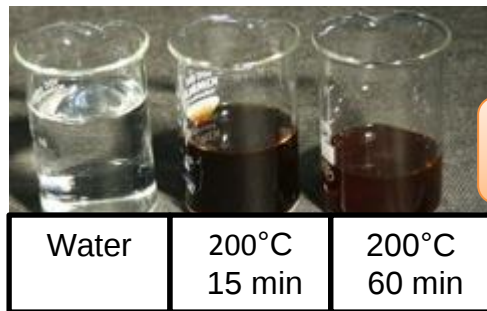
Use of fermentation residues from a biogas plant (provided by Heidekreis)





Sample	SSA _{total} / m ² g ⁻¹	SSA _{micro} / m ² g ⁻¹
Unmodified	17	3
a) NaOH, b) H ₂ SO ₄	4	4
a) NaOH, b) H ₂ SO ₄ _Extr.	70	12
a) NaOH, b) H ₂ SO ₄ _SA	49	44
a) NaOH, b) H ₂ SO ₄ _SA_Extr.	112	47

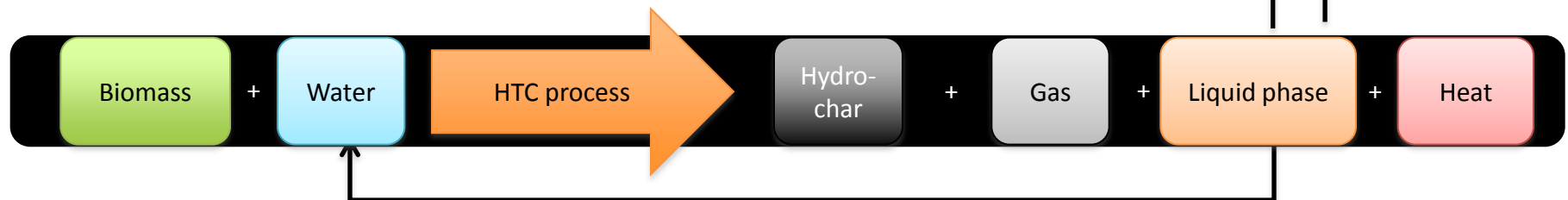
- Direct from HTC process:
Chemical oxygen demand (COD): 10 – 70 g/l
- Required: COD < 1 g/l
- Solution: extraction of valued products,
degradation of organics?

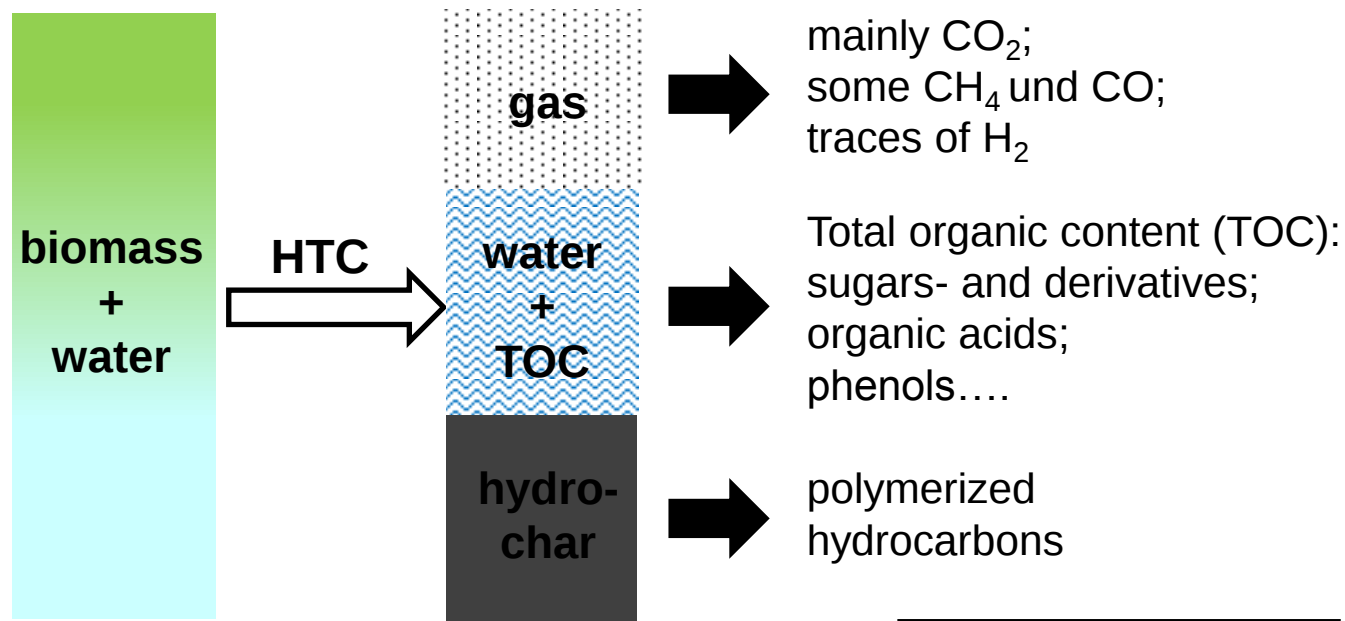


Make-up:
By chemistry? physics? biology?



Extraction of
chemicals, ,
e.g. 5-HMF,
phenols



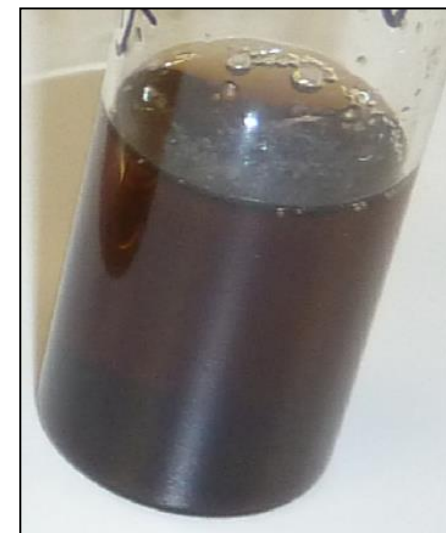


Main research

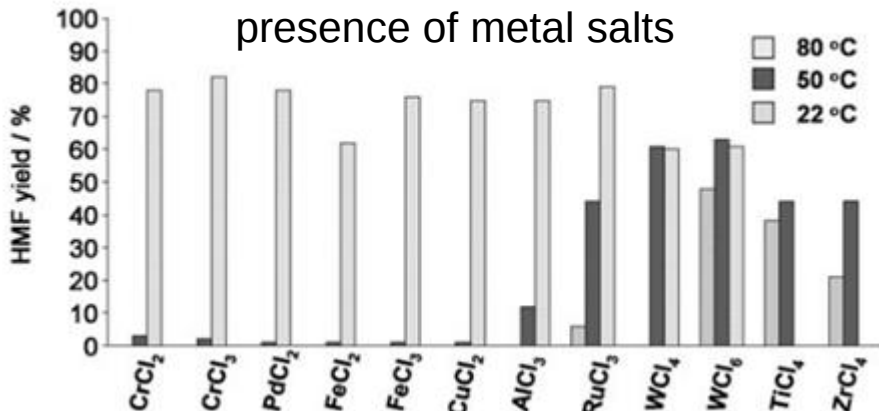
- Qualitative and quantitative analysis of the aqueous HTC process water

Main aims

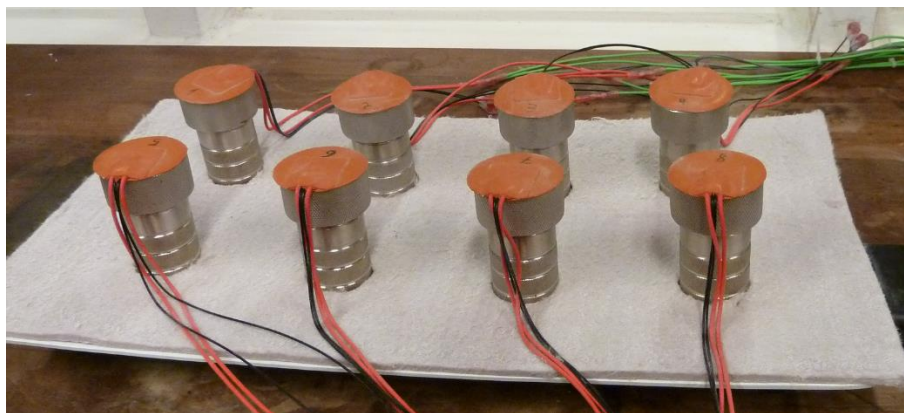
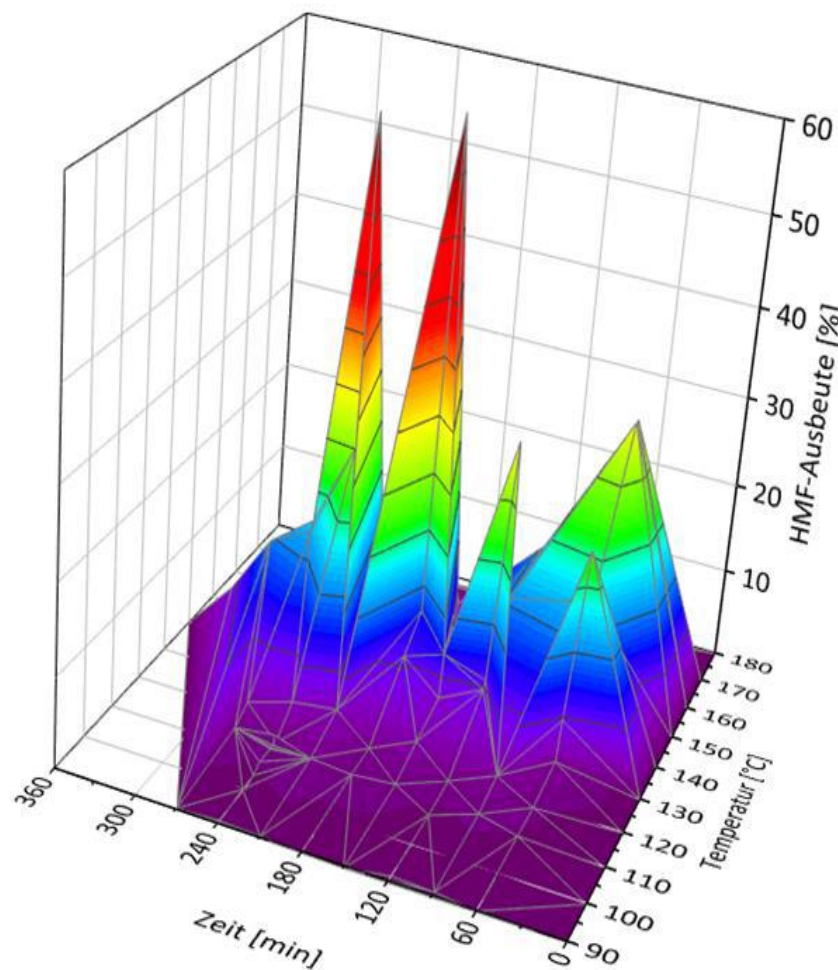
- Reduction of TOC
- Getting information about it's composition



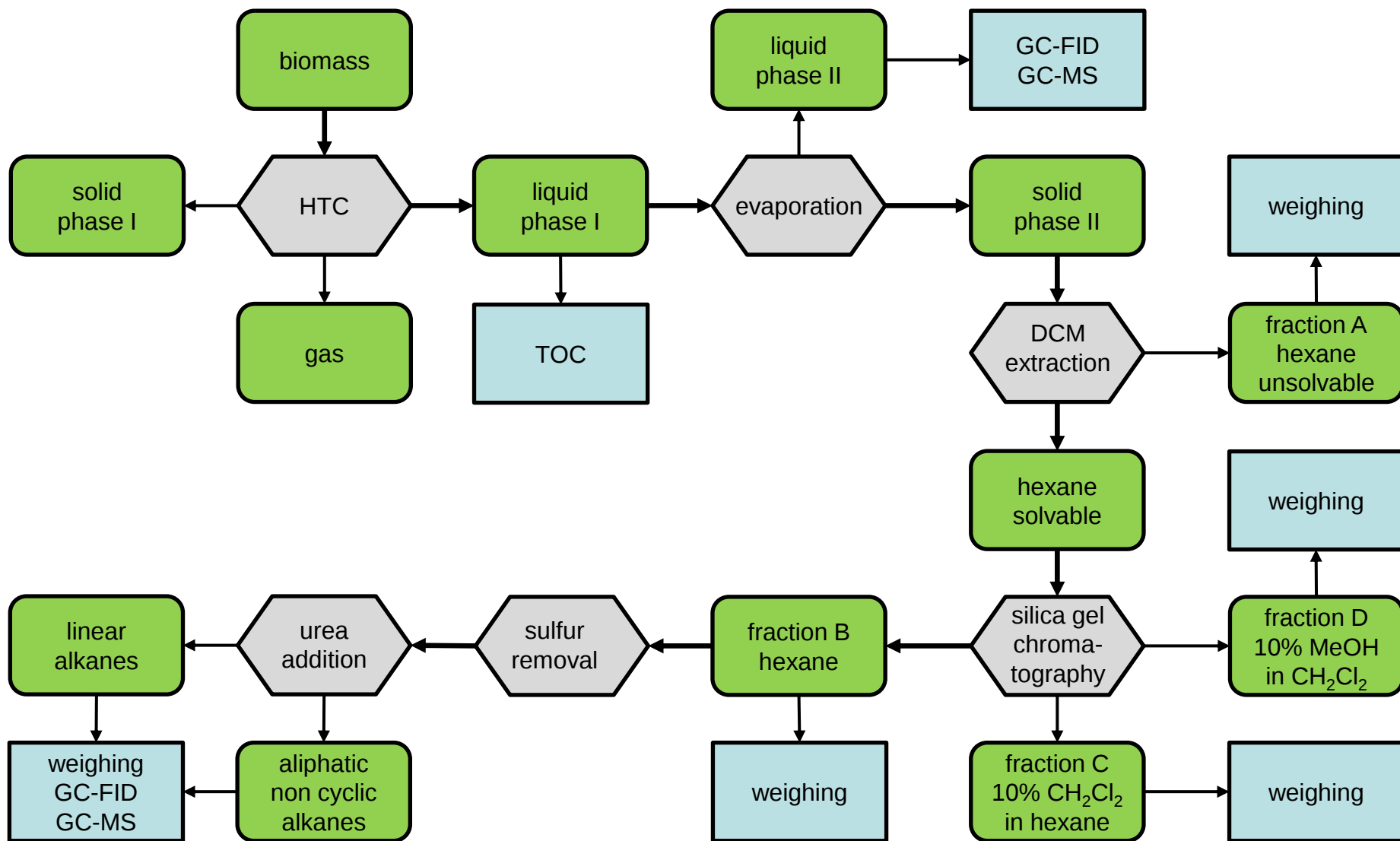
5-HMF extraction in
presence of metal salts

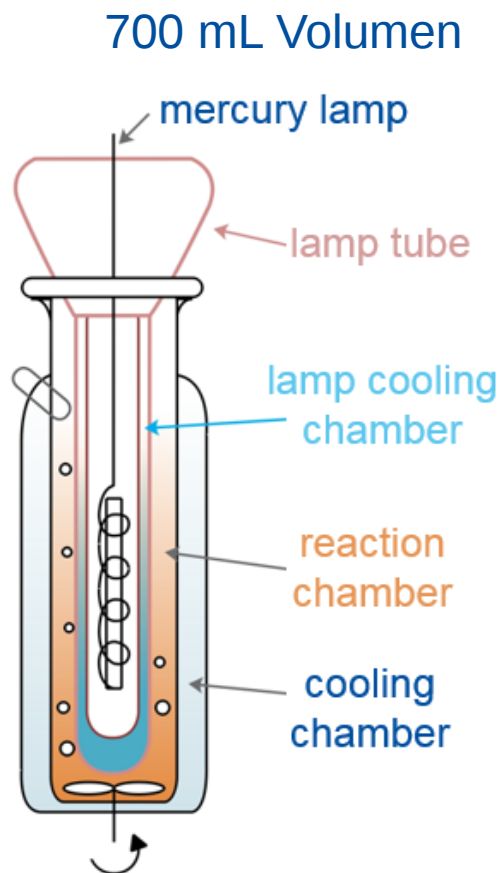


Results of the extraction
of 5-HMF from HTC
waste water



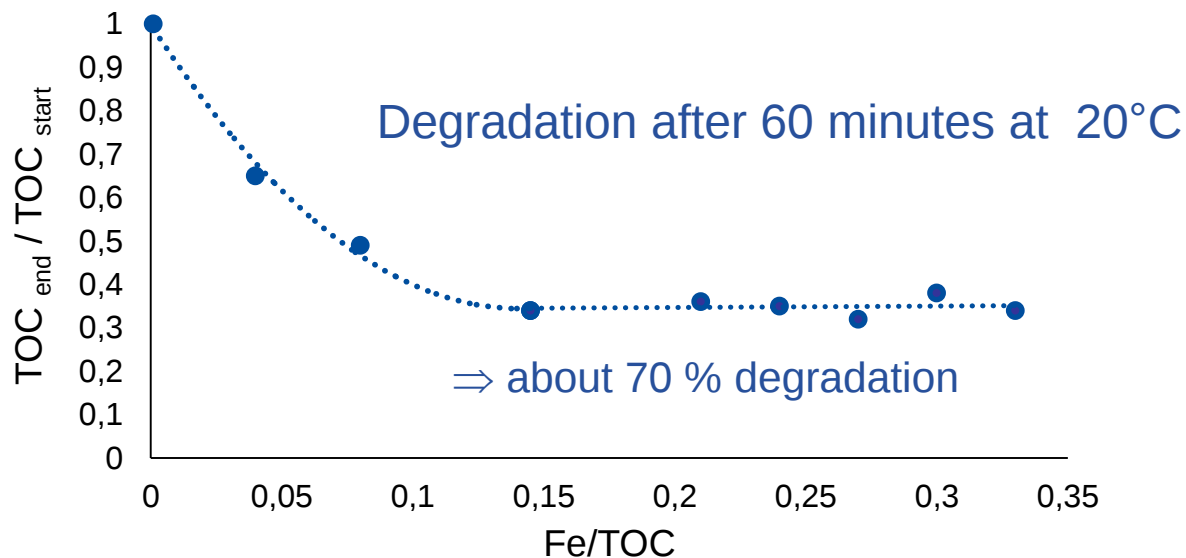
Multi-autoclave setup for Hydrolyzation
10 ml volume Teflon® inlet
Energy comes from a heating plate



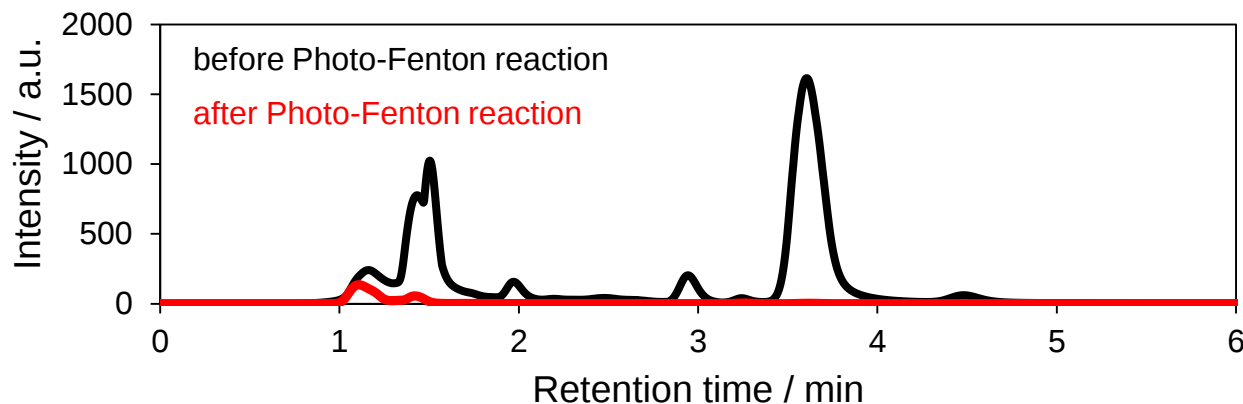


Reagents:

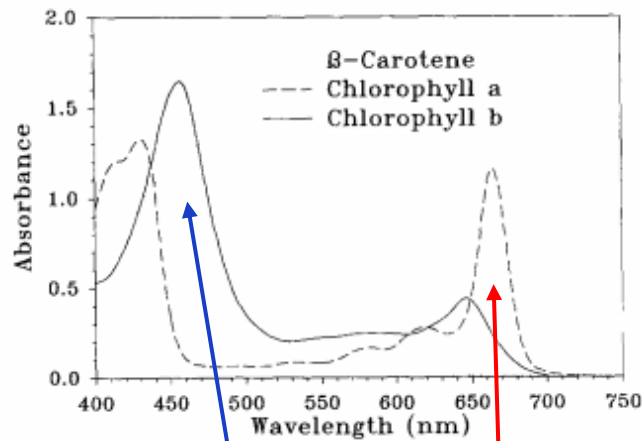
- HTC water phase
- Fe(II) salt
- H_2O_2



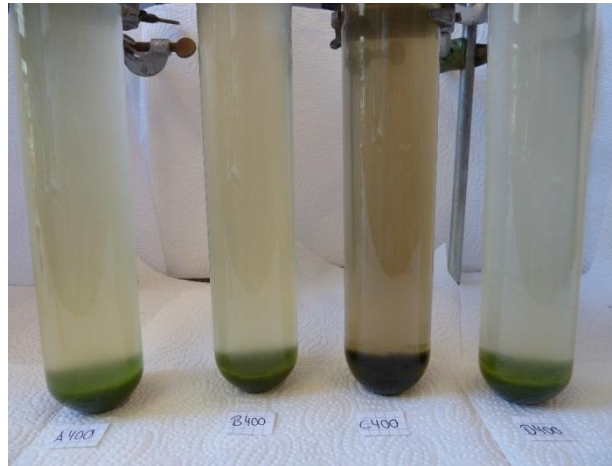
Gas chromatography analysis



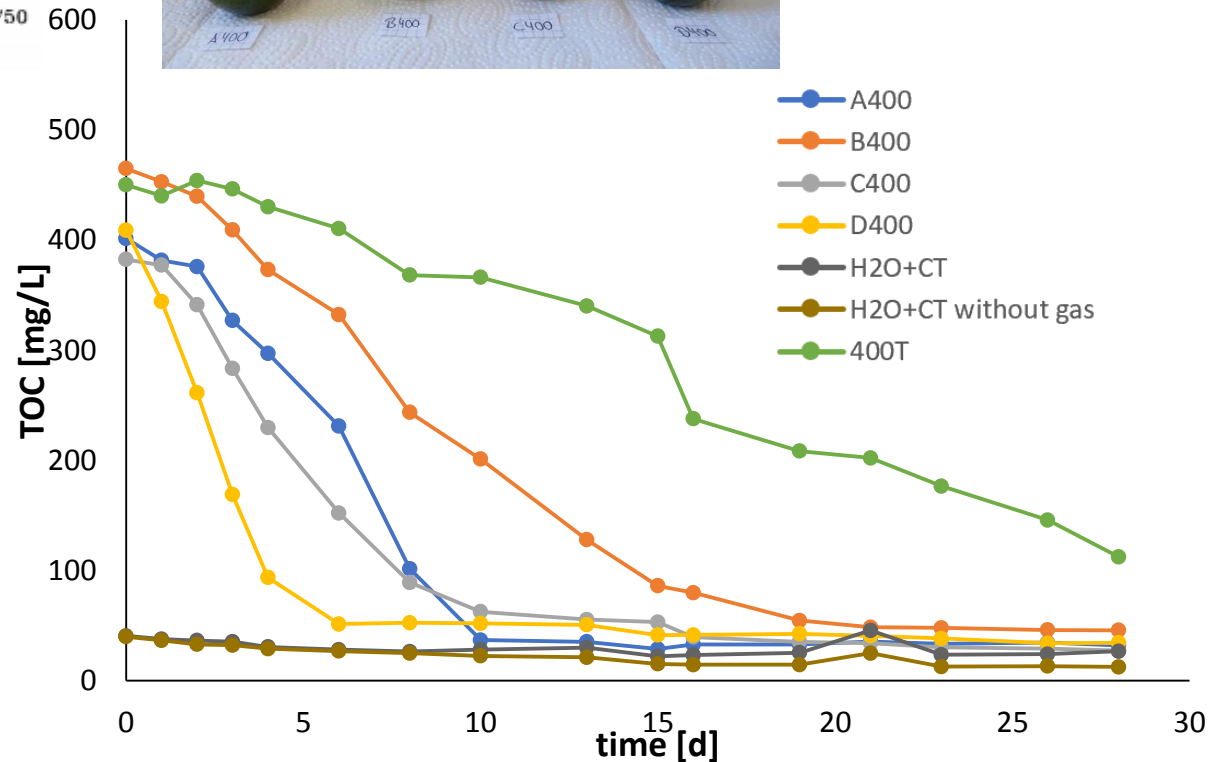
- peak areas decrease by about 95%,
- some compounds are not UV active ⇒ not detectable



LED	blue	green	red
λ [nm]	467-470	570-575	620-625
Status	max	off	max



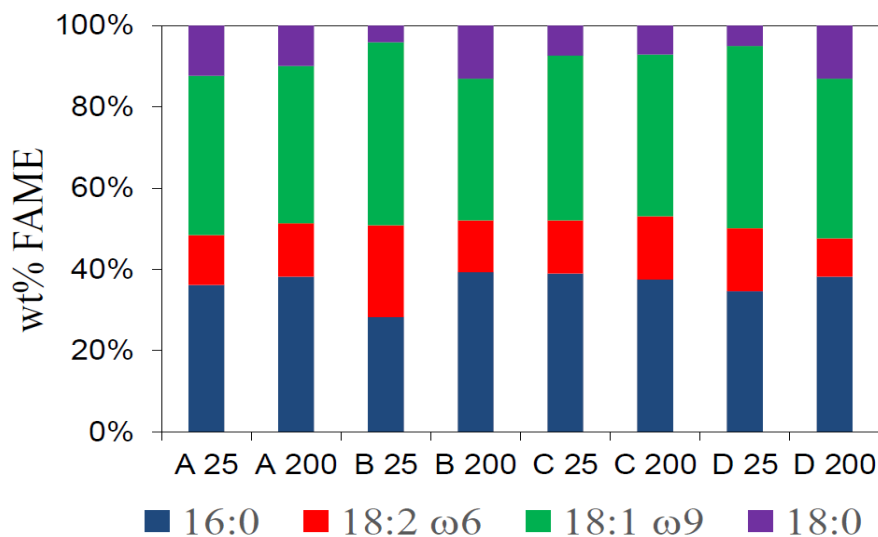
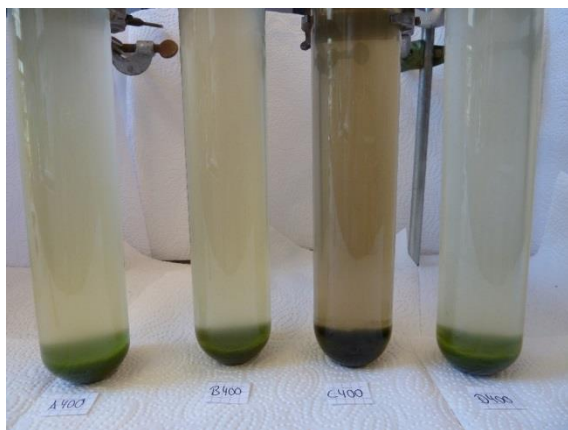
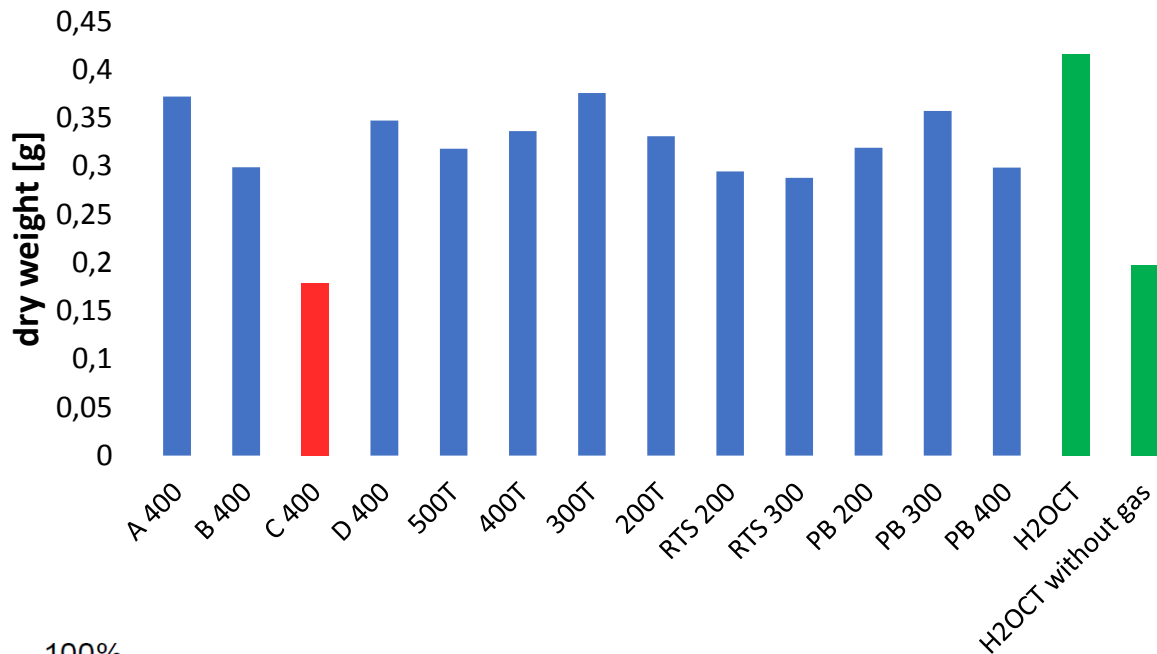
A	insoluble in hexane
B	non-polar
C	aromatics
D	NSO compounds



Dry mass of the grown algae

Organic phases for feeding the algae:

A	insoluble in hexane
B	non-polar
C	aromatics
D	NSO compounds
T	terephthalic acid
RTS	pig manure
PB	cattle manure



Composition of the formed fatty acids

- Hydrothermal Carbonization leads to hydrochars with more chemical functionality (functional groups for interaction with other species) than biochar from pyrolysis.
- HTC hydrochars are peat or brown coal (lignite) like.
- HTC hydrochars show a potential for applications as „green activated carbons“ (adsorption of gases, dyes, herbicides, pesticides, etc.), filter materials, catalyst supports, ...
- Tailored mixtures of soft and hard biomasses can create porosity / inner surface without use of additives after activation with steam. Lignin structures act possibly as internal templating agent. However, the chemical functionality is partly lost due to the treatment with steam.
- The total organic content (TOC) of the HTC water phase can be lowered significantly by Photo-Fenton reaction (requiring hydrogen peroxide) or the use of algae. The (somewhat stressed) algae produce more fatty acids than under ideal growth conditions.
- Extraction of valued products like 5-HMF from the organic contaminants of the HTC waters seems to be possible.

Many thanks for your kind attention

