



Source: <https://www.ghentdredging.be/>

## EVALUATION OF CALCINED DREDGING SEDIMENTS AS SUPPLEMENTARY CEMENTITIOUS MATERIALS

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**Interreg**  
France-Wallonie-Vlaanderen

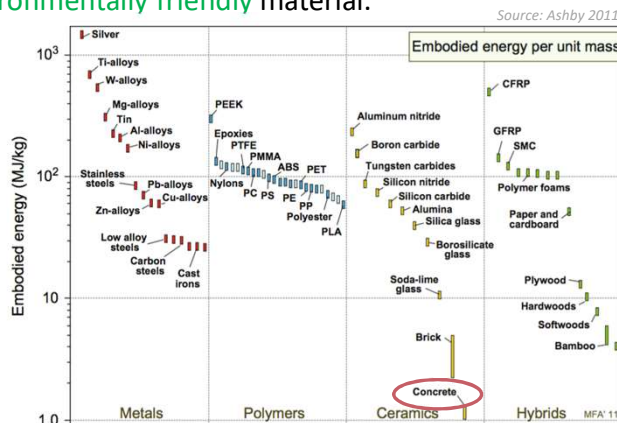


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### INTRODUCTION

- Concrete is an **environmentally friendly** material.



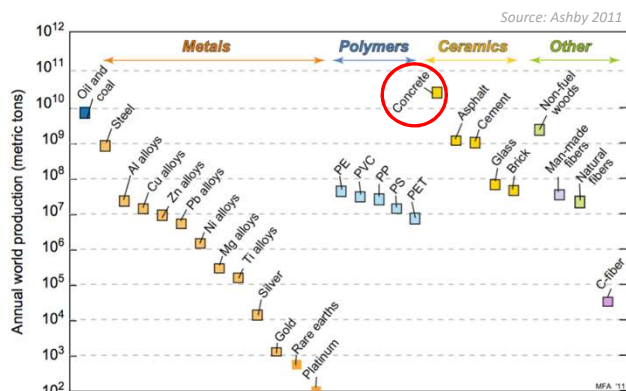
But ...

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## INTRODUCTION

- Concrete is the **most used material** after water.



- That is why is responsible for **8% of the anthropogenic CO<sub>2</sub>** due to the cement production

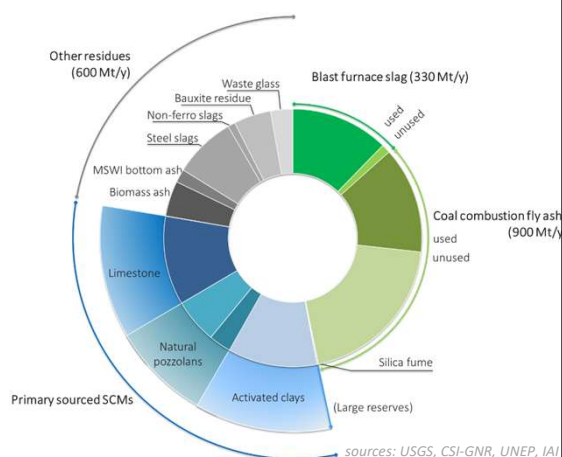
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## MOTIVATION

- The most pragmatic approach to tackle **environmental impact** of concrete is to **reduce** the clinker factor by **using Supplementary cementitious materials (SCM)**.
- Current convectional SCM's** are mostly used in the production of blended cements.
- Potential SCM's locally available in abundance must be explored.**



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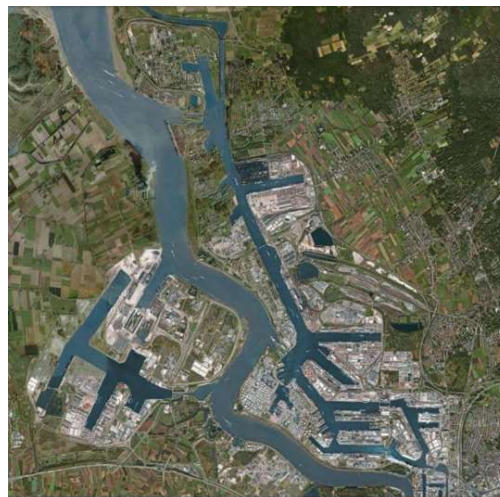
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## RESEARCH SIGNIFICANCE

- Previous research on valorisation of dredging sediments from Port of Antwerp:

- Sandy fraction ( $>63 \mu\text{m}$ ) used in civil works
- Fine fraction dewatered by membrane filter press
- Fine fraction low in organics and heavy metals
- Flash calcination ( $750-900^\circ\text{C}$ ) of fine fraction produced reactive SCM
- First pilot testing successfully completed, further upscaling ongoing



Port of Antwerp

[Snellings et al. 2016, 2017; Van Bunderen et al. 2017, 2019]

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## RESEARCH SIGNIFICANCE

- Extension of research to canal/waterway sediments (Interreg FWVL V VALSE project):
- Cross-border Scheldt basin – supporting integration of dredging sediment management and valorisation
- 3 canal locations:
  - Lens (F)
  - Brussels – Charleroi (B-Wallonia)
  - Ghent-Terneuzen (B-Flanders)



Source: Simina Margareta Stanc, et al. 2009

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## RESEARCH CHALLENGES AND OBJECTIVES

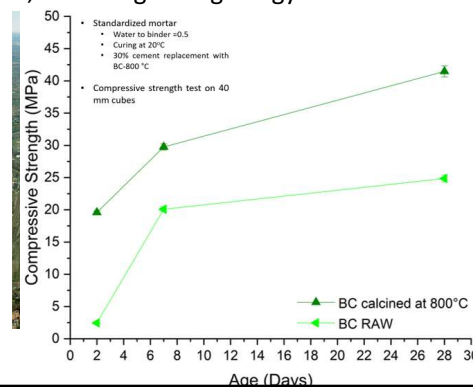
- Region of intense (historic) industrial activity (coal mining, steel production, chemistry)

### Challenges:

- Higher levels of **organics** and **heavy metals**, concentrated in fine fraction
- Canal sediments are more **variable in composition**, reflecting local geology and activities

### Objectives:

- Evaluate impact on calcined **SCM reactivity** and **PC hydration** of:
  - Organics and heavy metals
  - Variations in sediment composition
- Establish **optimum calcination temperatures** for each sediment



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## Results

# Calcined dredging sediment characterisation

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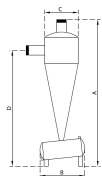
## MATERIALS: PRE-TREATMENTS BEFORE CALCINATION

### Sediment's

- Lens Channel
- Brussel – Charleroi Channel
- Gent-Terneuzen Channel



Removal of sand  
fraction ( $>63 \mu\text{m}$ )



dewatering using a  
pilot filter press



filter cake



Crush and oven  
dry in lab at  
40°C for 24 h



Ball mill



Calcination in a box furnace



Static conditions, 1 h at  
500 °C, 600 °C, 700 °C  
and 800 °C, in air.

2 min @ 400 rpm  
90 g sample with 6 balls  
of 30 mm diameter.

Ready for chemical and  
physical analysis

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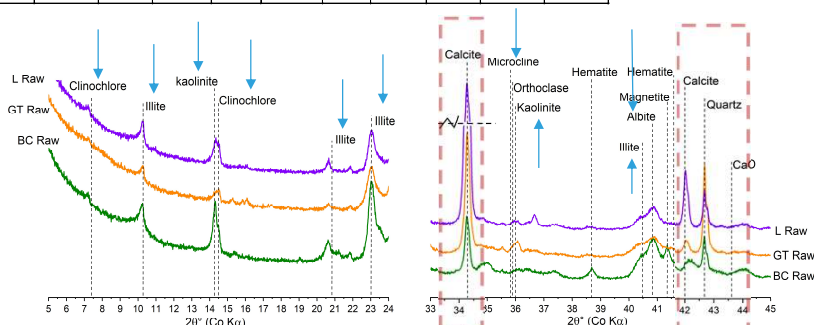
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## MATERIALS: CHEMICAL COMPOSITION AND MINERALOGY

	Chemical composition (wt.%)											
Raw Sediment	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	CaO	MgO	K <sub>2</sub> O	Na <sub>2</sub> O	SO <sub>3</sub>	P <sub>2</sub> O <sub>5</sub>	LOI	TC	TOC
BC	40.8	13.9	10.0	2.6	1.1	2.1	0.5	1.0	1.1		13.18	11.97
GT	61.8	9.9	4.1	5.5	1.3	1.4	<0.7			13.3	4.50	3.62
L	32.2	8.2	4.3	10.8	0.9	1.6	1.9	4.0	1.4		13.79	10.35

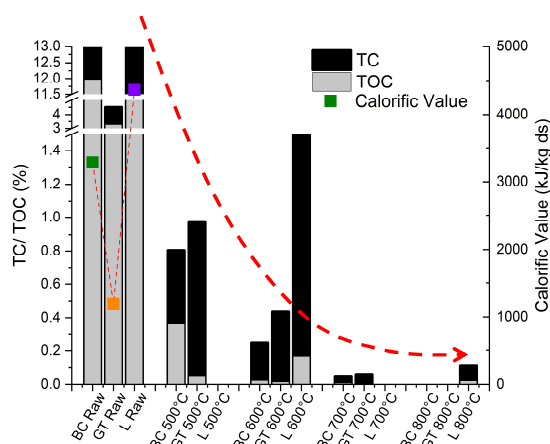
- Presence of quartz GT>L>BC
- Presence of calcite L>GT>BC
- Clay minerals (Clinocllore, kaolinite, and illite)
- Feldspar minerals (albite and microcline).



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## RESULTS: TOTAL CARBON (TC), TOTAL ORGANIC CARBON (TOC), AND CALORIFIC VALUE FROM BOMB CALORIMETRY



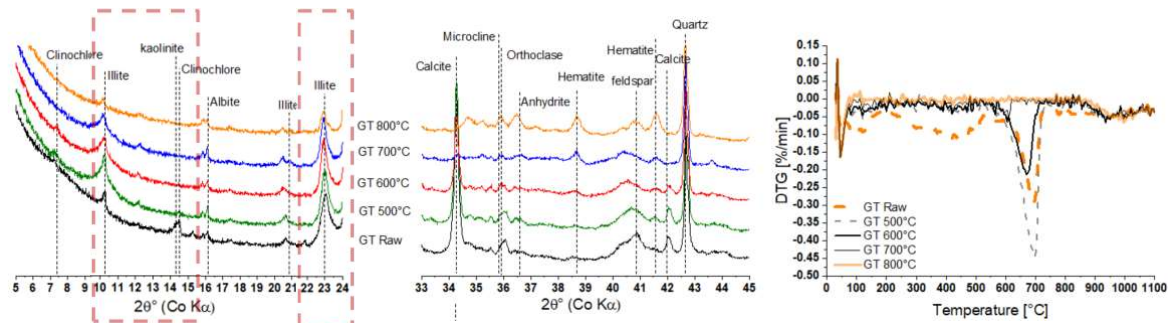
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- Already at 500 °C, most organic matters have been decomposed.
- By 800 °C the TC and TOC values in case of BC and GT are negligible.
- L, organic matter persisted up to 600 °C, and even at 800 °C.
  - Rock-Eval pyrolysis can be carried out to better understand the source of its organic carbon

## CHEMICAL CHARACTERIZATION OF SEDIMENTS AS FUNCTION OF CALCINATION TEMPERATURE

- Mineralogical modification of clay fractions:
  - Reflection of some clays such as kaolinite and Illite to disappeared or weaken making the sediment more amorphous. For example in case of GT:



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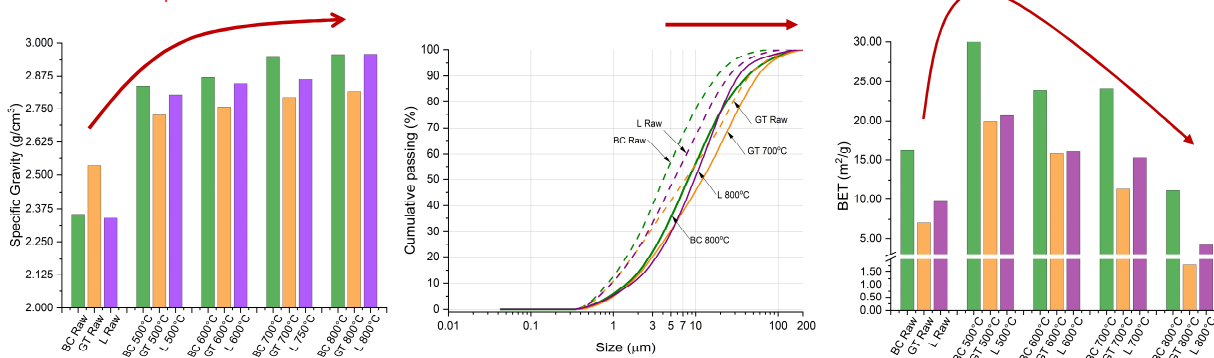
## RESULTS: SPECIFIC GRAVITY, PARTICLE SIZE DISTRIBUTION (PSD), BET SPECIFIC SURFACE AREA

- Mineralogical modification & changes in chemical composition of the dredging sediments is reflected on their physical characteristics

Specific gravity of all sediments increases with increasing calcination temperature up to 700 to 800 °C

PSD shifts to coarser size with increasing calcination temperature



Increase of the specific surface at 500 °C, and thereafter they continuously decrease



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# Results

## R<sup>3</sup> test

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### R<sup>3</sup> TEST - INTRODUCTION

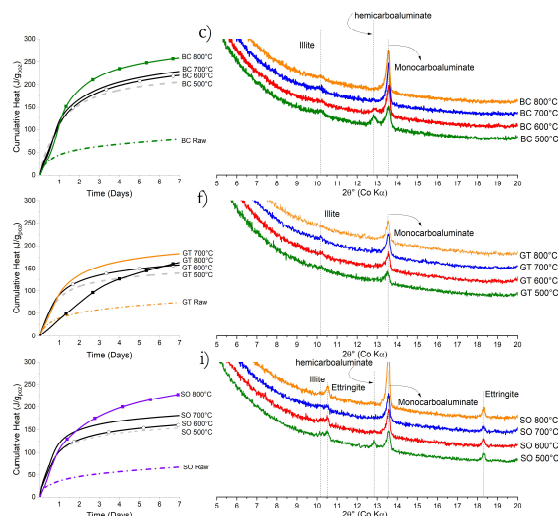
- Incorporating materials with different origins in Portland cement is a **subjective method** in assessing their **pozzolanic reactivity**, due to the **reaction of the OPC phases**.
- In **RILEM TC 267-TRM**, a **model system** has been established where the material is mixed with **Calcium hydroxide** and **alkali sulfates, limestone powder and water**.
- The **rate and magnitude** of the **chemical reaction** of the material within the mix, represent the **reactivity potential of the material**.
- For **quantification of this reactivity**, **Isothermal calorimetry** cumulative heat is used and supplemented with **XRD and TGA**.

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## RESULTS: R<sup>3</sup> TEST

- R<sup>3</sup> tests suggest that the optimum calcination temperature for BC and L is around 800 °C while in case of GT is around 700 °C.
- GT calcined at 800 °C decreases the reactivity
  - Recrystallization and formation of high temperature phases.
  - Decrease in the amorphous content coming from the clays mineral content.



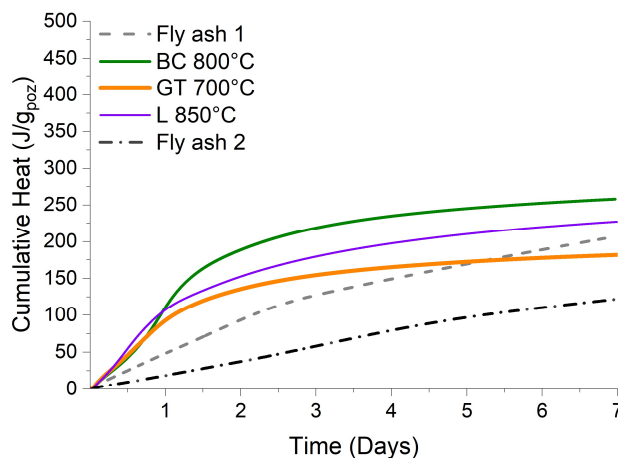
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## RESULTS: R<sup>3</sup> TEST: COMPARISON WITH CONVECTIONAL SCMS

- The calcined dredging sediments, show similar or higher reactivity than of silica rich fly ashes.



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## CONCLUSIONS

- Calcination of the 3 dredging sediments led to
  - Breakdown of the organic matters.
  - Activation of clay minerals (Kaolinite and chlorite) and to some extent Illite.
  - Mineralogical and chemical changes due to calcination, resulted in changes in physical characteristics of the dredging sediments:
    - Higher specific gravity
    - Coarser particle size distribution
    - Lower specific surface area.
- R<sup>3</sup> test proposes the optimum calcination temperatures for use of dredging sediments as SCMs:
  - L and BC at 800 °C
  - GT at 700 °C

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## WHAT'S NEXT



- The optimum calcination temperature obtained for each dredging sediment will be used to carry out:
  1. Hydration study, at the paste scale at 20°C isothermal conditions
    - Reaction kinetics and mechanism
  2. Mechanical properties development, at the mortar scale at 20°C isothermal conditions.
  3. Environmental quality by carrying out leaching tests


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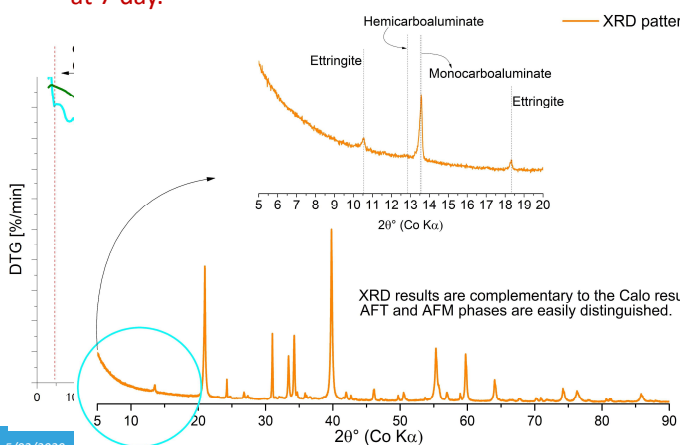
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### R<sup>3</sup> TEST - INTRODUCTION

- For objectivity, and quick assessment, the tests are carried out at **40°C** and the results are reported **at 7 day**.



XRD results are complementary to the Calo result. AFT and AFM phases are easily distinguished.

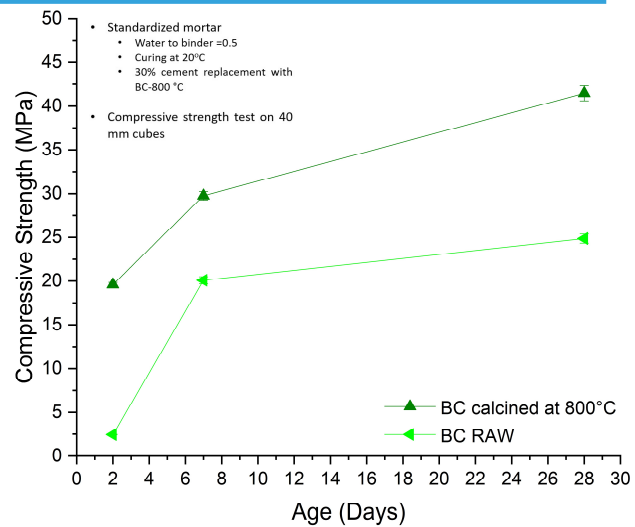


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## PRELIMINARY RESULTS

- Standardized mortar
  - Water to binder =0.5
  - Curing at 20°C
  - 30% cement replacement with BC-800 °C
- Compressive strength test on 40 mm cubes



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