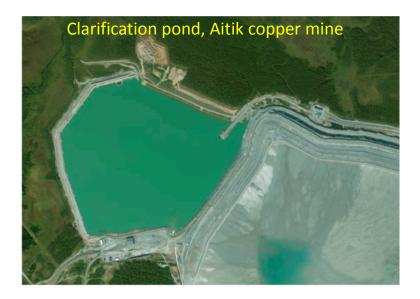
### Nitrogen removal at cold-climate mine sites in northern Sweden



Landbúnaðarháskól

Islands

Agricultural University of Iceland



Lino Nilsson, Anders Widerlund Division of Geosciences and Environmental Engineering Luleå University of Technology, Sweden







UNIVERSIT

OF TECHNOLOG







# **Background – the problem**

Ammonium nitrate based explosives are a major nitrogen source in the mining industry





- Mining operations are characterized by large water volumes with moderately high N concentrations (<30 mg N/L)</li>
- Active treatment for N removal is expensive and usually not feasible
- Passive or semi-passive removal methods should therefore be tested

# **Background – project WaterPro**

In Sweden, WaterPro was coordinated with project *miNing*<sup>1</sup> at the Kiruna and Aitik mines in northern Sweden

Project *miNing* consisted of three sub-projects, each aiming to develop low-cost, passive or semi-passive treatment methods that require only a minimal amount of energy to maintain in operation:

- Microbial denitrification in mine ponds
- A denitrifying bioreactor system for nitrate removal
- Wetland systems for nitrogen removal by macrophytes, algae, and denitrification

# The Swedish part of WaterPro investigated denitrification in mine ponds

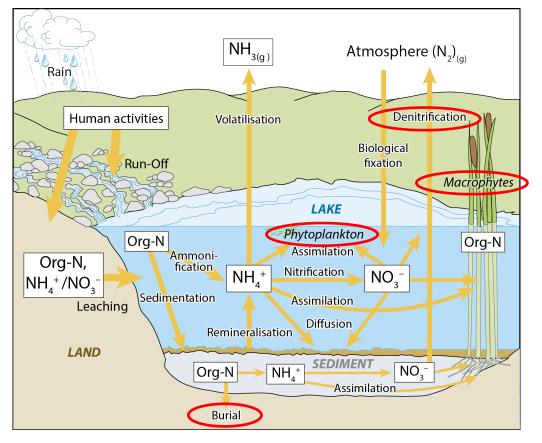
<sup>1</sup> **Project** *miNing*: *Reduction of nitrogen discharges in mining processes and mitigating its environmental impact* Project period 2013–2018. Coordinated by Swedish University of Agricultural Sciences





### **Background – nitrogen cycling in mine ponds**

Nitrogen cycling in mine ponds is similar to that in lakes where permanent N removal occurs



Permanent nitrogen removal is possible trough:

- Denitrification in pond sediments
- Nitrogen uptake in phytoplankton (algae) and macrophytes followed by partial burial of nitrogen in sediments

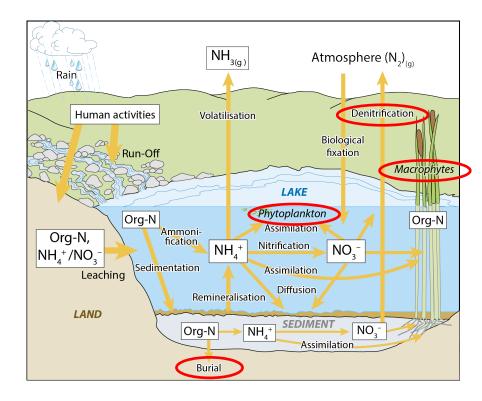
#### **Research question**

Can N removal in mine ponds be increased so that ponds can be used as passive N removal systems? Laboratory studies showed that:

- Denitrification in the Kiruna pond sediment was carbon-limited
- Phytoplankton was an efficient carbon source for driving denitrification

### Hypothesis for the pond study

- Phytoplankton can be used as a natural carbon source to increase denitrification in ponds
- Phytoplankton growth and denitrification can be increased by phosphorus fertilization of mine ponds



# Work done – methods

### Field work in the mine ponds at Kiruna

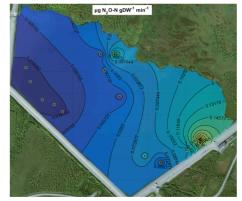
- Water and sediment chemistry
- Potential denitrification activity (PDA)

### Field work in mesocosms at Kiruna

- Nitrogen removal from pond water
- Potential denitrification activity

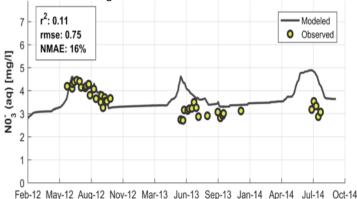
### **Computer modelling of nitrogen cycling**

 Modelling of N transformations in mesocosms and ponds PDA in the Kiruna clarification pond sediment

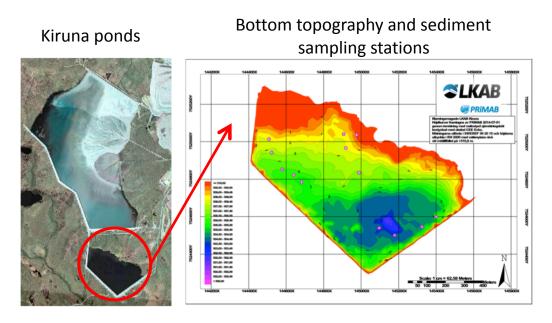


Mesocosms at the Kiruna mine

#### Modelled NO<sub>3</sub> concentration at the Aitik mine

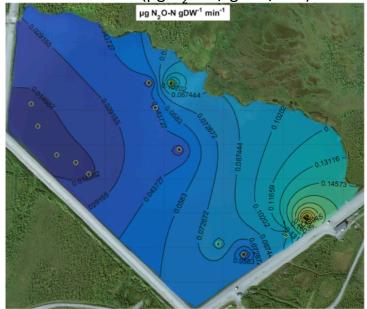


### **Results – field work in the clarification pond, Kiruna**



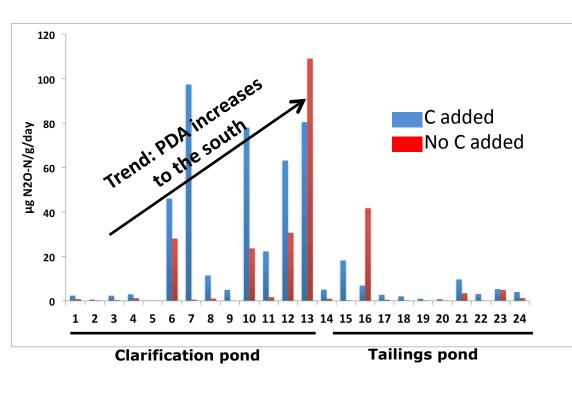
- The clarification pond collects tailings that have not settled in the tailings pond before the mine water is released into the receiving waters
- Water and sediment were sampled in the clarification pond

Potential denitrification activity in pond sediment (µg N<sub>2</sub>O-N/g sed/min)



- Potential denitrification activity (PDA) was measured at a number of stations in the pond
- PDA is higher in the southern, deeper part of the pond

## Potential denitrification activity (PDA) in pond sediments



#### **Clarification pond**

PDA in clarification pond sediments is higher than in the tailings pond sediments

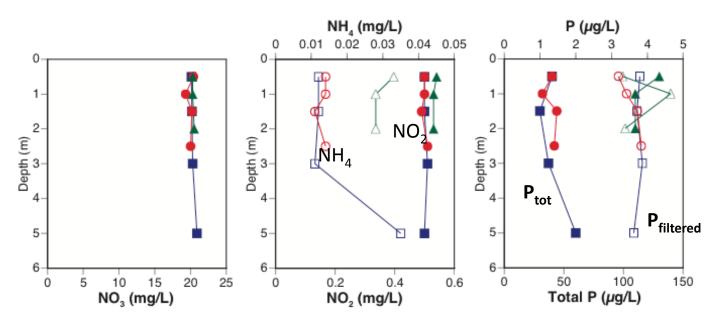
PDA in pond sediment (0–2 cm): **0.094 g N/m<sup>2</sup> d** (no C added) **0.20 g N/m<sup>2</sup> d** (with C added)

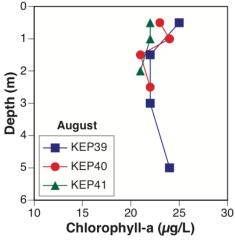
0.094 g N/m<sup>2</sup> d corresponds to ca 5 tonnes of N/90 days (three month summer period) or a N removal rate of 0.1 %/day

Adding carbon to pond sediment may approximately double denitrification in the pond from 5 to 10 tonnes of N per 90 days, or from 0.1 to 0.2 %/day

Hypothesis: Phytoplankton carbon can be increased through phosphorus fertilization

### **Results – field work in the Kiruna pond** Vertical profiles of N, P and Chl-a concentrations

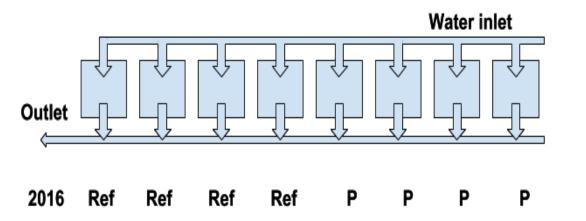




- Nitrate is the dominating N species (~20 mg/L)
- Filtered P (<0.2  $\mu$ m) is 3–4  $\mu$ g/L (bioavailable?)
- Total P is ~50 µg/L
  - Chl-a is 20–25 µg/L

### **Mesocosm study – experimental design**

- Mesocosms fed with water from the clarification pond were used to simulate conditions in the pond if phosphorus fertilization is used to stimulate phytoplankton growth
- Mesocosms: 1000-L containers filled with 0.3 m of tailings and circulating pond water
- Four mesocosms fertilized with P / four reference mesocosms with no P added





# Potential denitrification activity in mesocosm sediments

- Measured on surface sediment (0–2 cm) from four reference and four phosphorus fertilized mesocosms
- Three replicates from each mesocosm
- Potential denitrification activity (PDA) measured with / without additional C at 15 °C



Sediment in

reference mesocosm

Algae growing in fertilized mesocosm



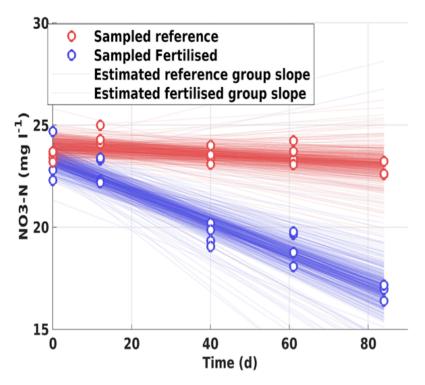
Sediment in fertilized mesocosm with layer of algae



### **Results – mesocosm study**

### Nitrate removal in mesocosms

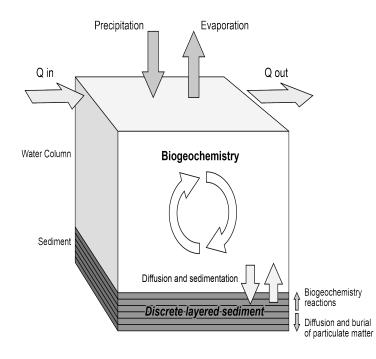
- Nitrate concentrations in fertilized mesocosms decreased from 23 to 17 mg NO3-N/L after 86 days (June to September)
- Nitrate decrease is caused by anaerobic denitrification in tailings, and possibly also aerobic denitrification in the water column



Red: Reference mesocosms with no P added (lines are statistically generated)

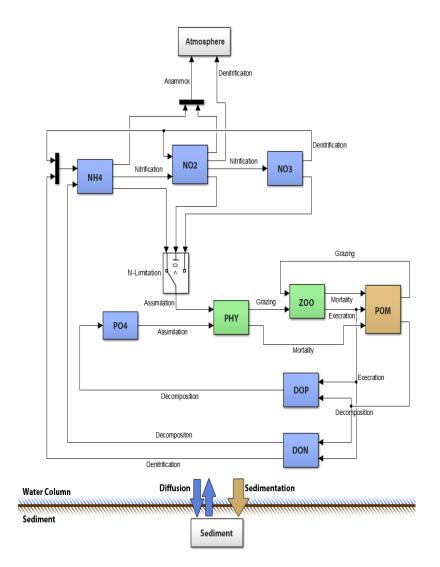
Blue: Mesocosms with P added (lines are statistically generated)

# **Biogeochemical model – conceptual model**



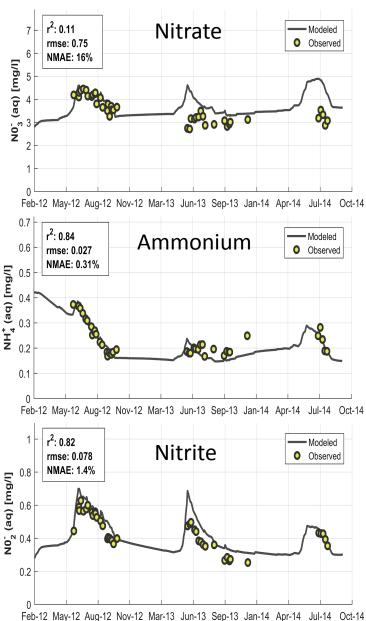
- Biogeochemical reactions calculated in water column and sediment
- Precipitation, evaporation, ice formation, and water flow are included

#### Model with 10 state variables



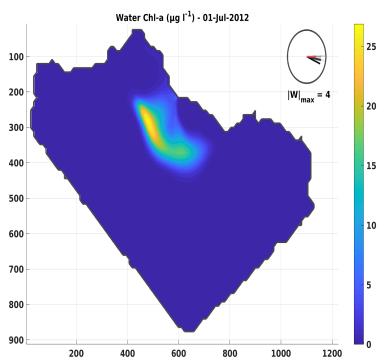
### **Results – biogeochemical modelling at Aitik** Pond concentrations 2012–2014

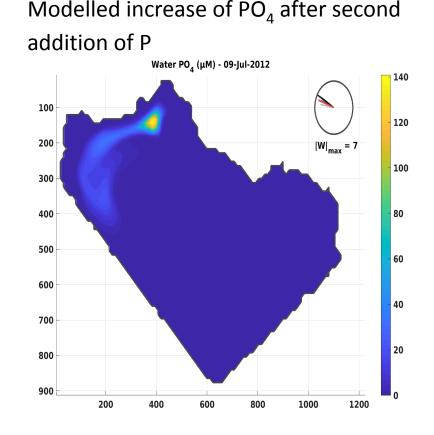
- Based on monitoring data from the mining company Boliden Mineral AB
- Ammonium and nitrite accurately modelled with r<sup>2</sup> > 0.8
- Nitrate shows less satisfactory r<sup>2</sup> (0.11), and modelled NO<sub>3</sub> increases with time
- Allowing for aerobic denitrification in the water column improves model accuracy for nitrate (r<sup>2</sup> = 0.54)



## **Results – biogeochemical modelling at Kiruna**

- Testing P fertilization at full scale in the pond would be controversial. Biogeochemical modelling was therefore used to test the effect of P fertilization on nitrate removal
- 4.5 kg of P was added *in the model* on six occasions during summer in the northern part of the pond

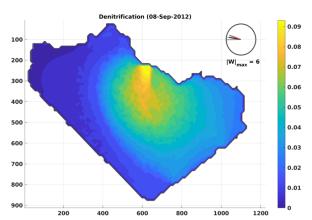




Modelled increase of Chl after first addition of P

# **Results – biogeochemical modelling at Kiruna**

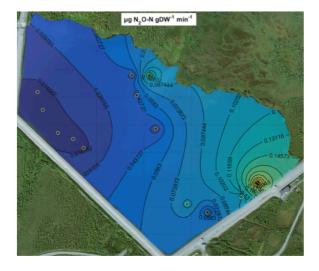
Modelled potential denitrification activity (PDA) is approximately similar to measured PDA in pond sediment



#### NO, Fertilised:Reference 100 99 Fertilised/reference (%) 98 97 96 95 94 93 0 50 100 150 200 250 300 350

time (d)

#### Measured PDA



After P fertilization, modelled  $NO_3$ decreased by about 5% (100 to 95 conc units)

Considering the recirculation of pond water back into the process,  $NO_3$  decrease over time can be expected to be higher than 5%

#### Modelled PDA

## Algae as a carbon source in mine ponds Risks and disadvantages

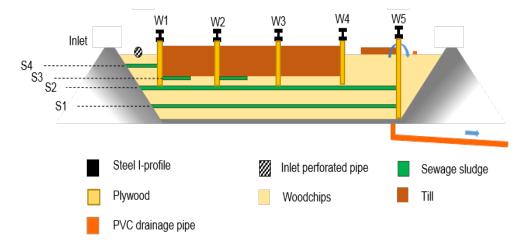
• Fertilization with phosphorus

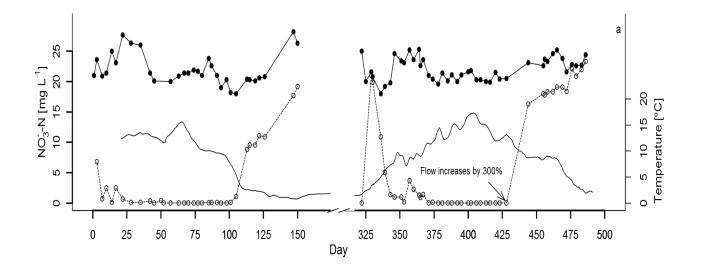
 To some extent added P will be retained in mine ponds, but receiving waters may still be eutrophicated

- Production of ammonium (NH<sub>4</sub>), dinitrogen oxide (N<sub>2</sub>O) and methane(CH<sub>4</sub>)
  - $-NH_4$  a nutrient and toxic to biota
  - $-N_2O$  and  $CH_4$  are greenhouse gases
- Quality of process water recycled from ponds
  - Higher concentrations of NH<sub>4</sub>, Chlorophyll, and dissolved organic carbon

### Denitrifying woodchip bioreactor at the Kiruna mine A subproject in project *miNing*

- Pilot-scale denitrifying bioreactor constructed in fall 2014 at LKAB's Kiruna mine
- Operations started June 2015
- Hydraulic residence time of ca 2 days until day 429
- Generally complete removal of nitrate unless < 5°C and/or HRT < 1 day





### EIT Raw Materials project NITREM A continuation of the bioreactor project

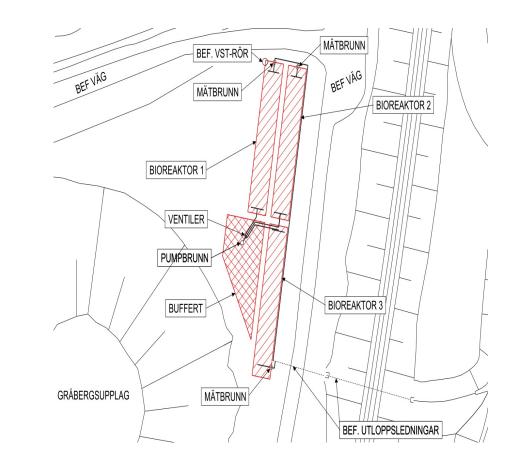
- An upscaling project financed by European Institute for Innovation and Technology (EIT) Raw Materials initiative for period 2018 – 2020.
- "NITREM establishes a service for the passive removal of nitrogen from waste rock leachate with a bioreactor technology, in conjunction with long-term waste rock management. Technology development has been driven by EUs Water Framework Directive and will enable the industry to meet current and future discharge requirements. The outcome is a low cost and low maintenance technology that is ready for market introduction and customer testing."





### Biroeactors at Kiruna site Installation completed fall 2018

- Water reservoir (*buffert*) collects groundwater seepage, which consists almost exclusively of leachate from waste rock dump (*gråbergsupplag*)
- Water pumped from a pumping chamber (*pumpbrunn*) through gate valves (*ventiler*) to the bioreactors.
- Water discharges from each bioreactor through a 1600 mm diameter) monitoring chamber (*mätbrunn*) and further to a drainage ditch (*Bef. Utloppsledningar*)
- Performance will be evaluated through all of 2019 and 2020





# **Conclusions**

- Denitrification in pond sediments is carbon limited, and increases when algae production increases in the system
- The denitrification activity increased by a factor of two after carbon addition to pond sediments, suggesting a possible additional removal of ca 5 tonnes of N/year in the Kiruna pond if carbon is not limiting denitrification
- Nitrate concentrations in P fertilized mesocosms decreased from 23 to 17 mg NO3-N/ L after 86 days (June to September)
- Modelling of the Kiruna pond indicated an increase in potential denitrification activity (PDA) that was similar to PDA measured in the pond sediment
- A question that remains to be answered is how much of the added PO<sub>4</sub>-P (6 x 4.5 kg in model) that actually will be discharged from the pond, potentially resulting in eutrophication of the receiving waters



Northern Periphery and Arctic Programme



#### EUROPEAN UNION

Investing in your future European Regional Development Fund