

# Thermal Spray Coatings For Industrial Applications



**SAPIENZA**  
UNIVERSITÀ DI ROMA



INSTM - Italian Consortium  
on Materials Science  
and Technology

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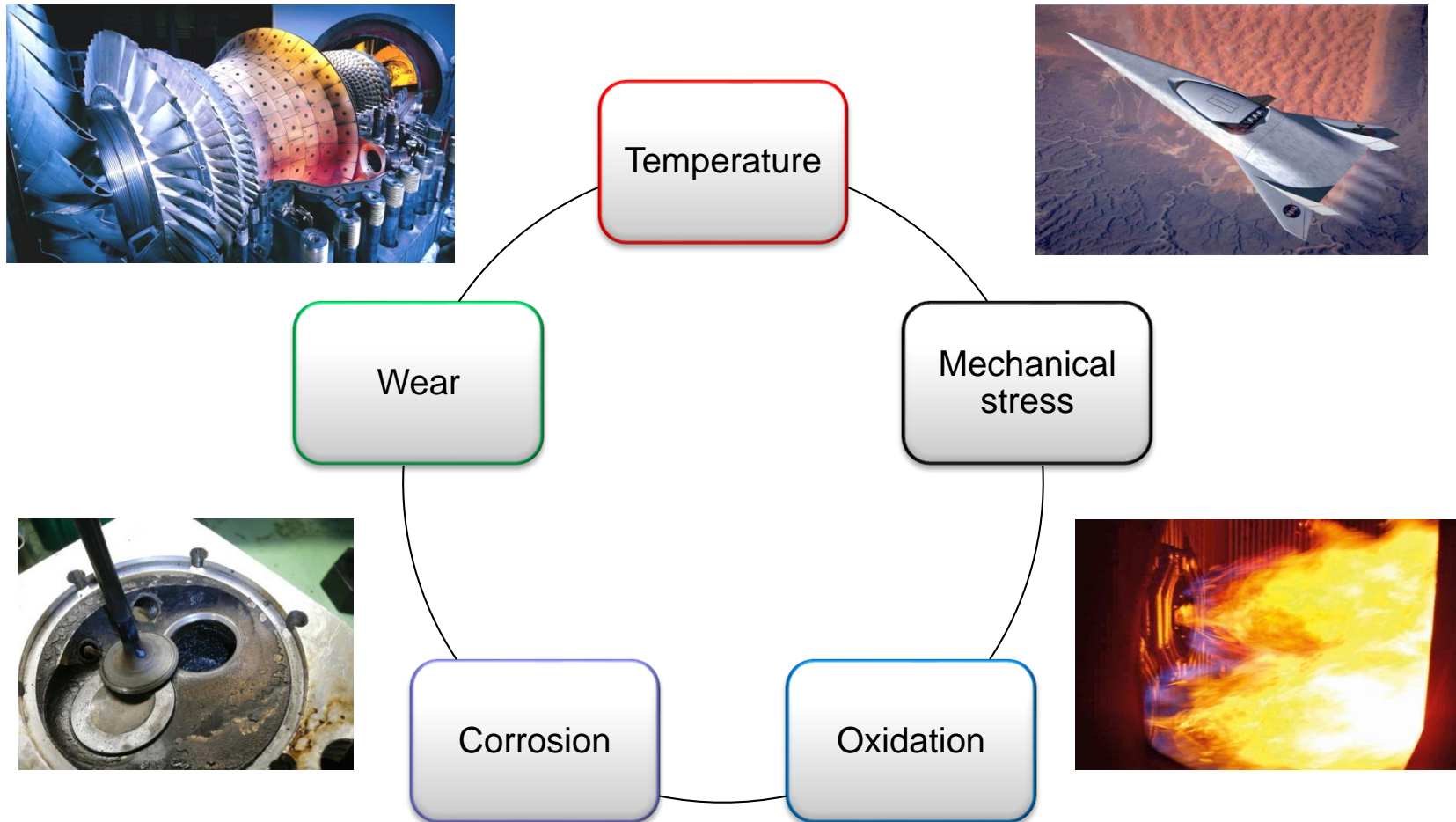
INSTM Research Unit "Roma La Sapienza"

Dept. of Chemical and Materials Engineering (ICMA)

Sapienza University of Rome

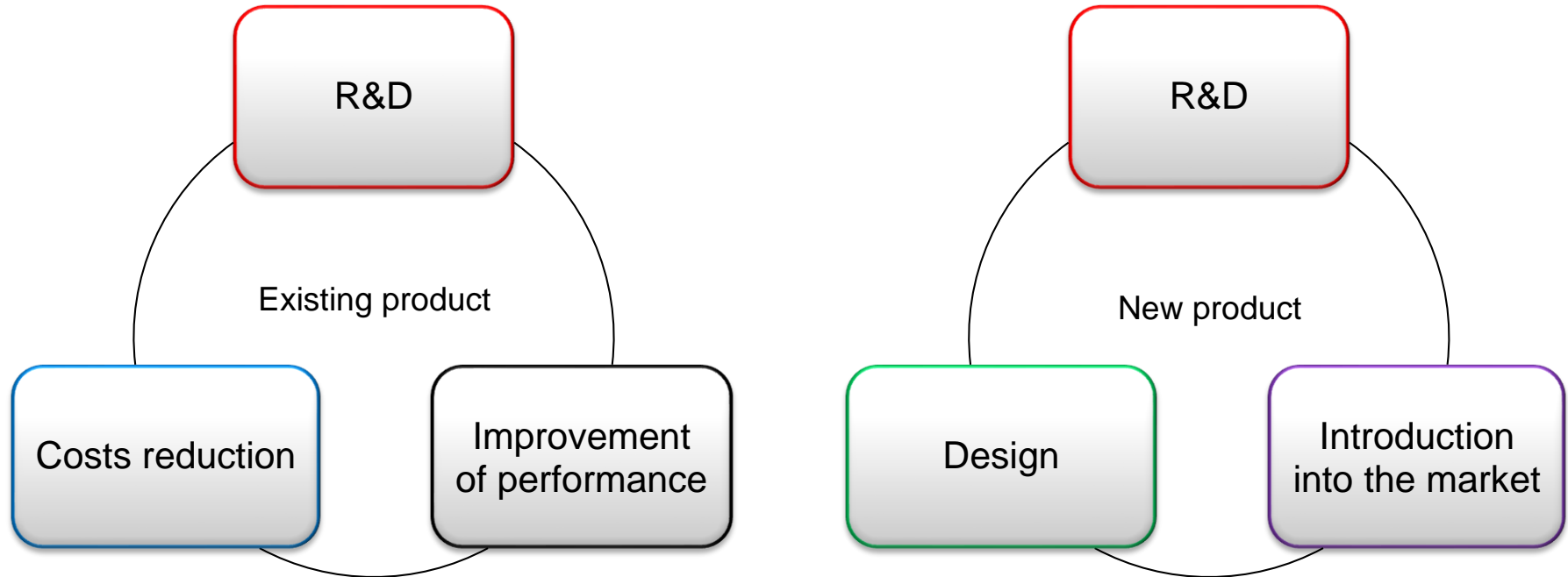
## Coatings for severe operating environment

## Heavy combination of mechanical stresses and aggressive environment



# Coatings for severe operating environment

R&D plays a key role for component working in critical operating conditions



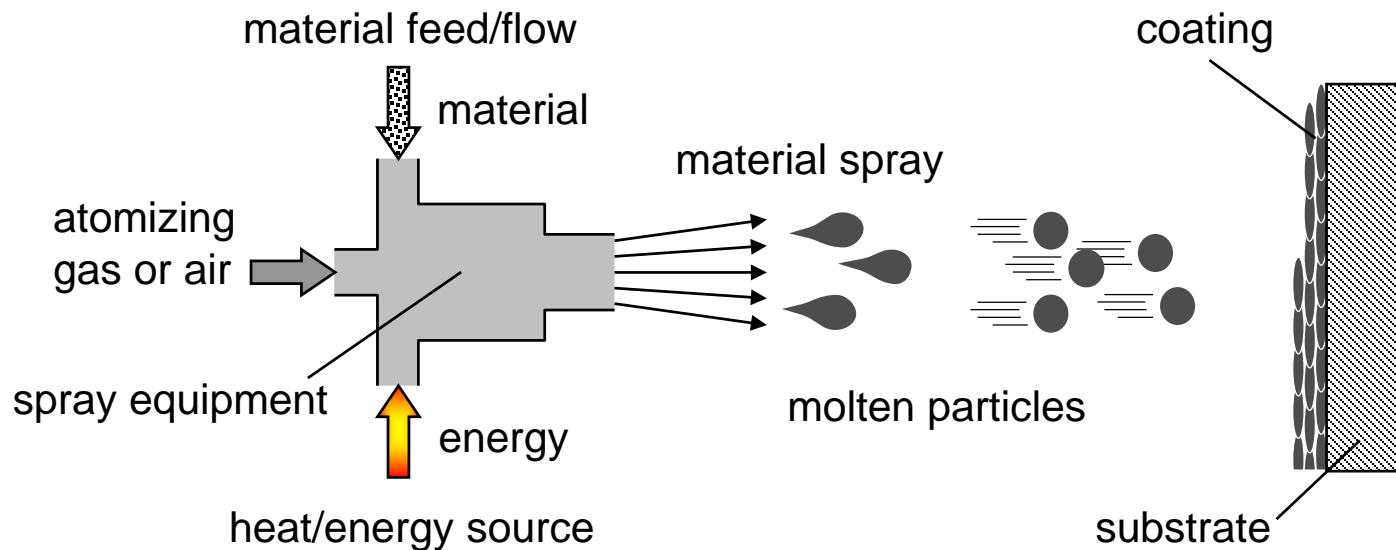
**Aim of presentation:**

**Description of case studies related to innovations and R&D in thermal spray field**

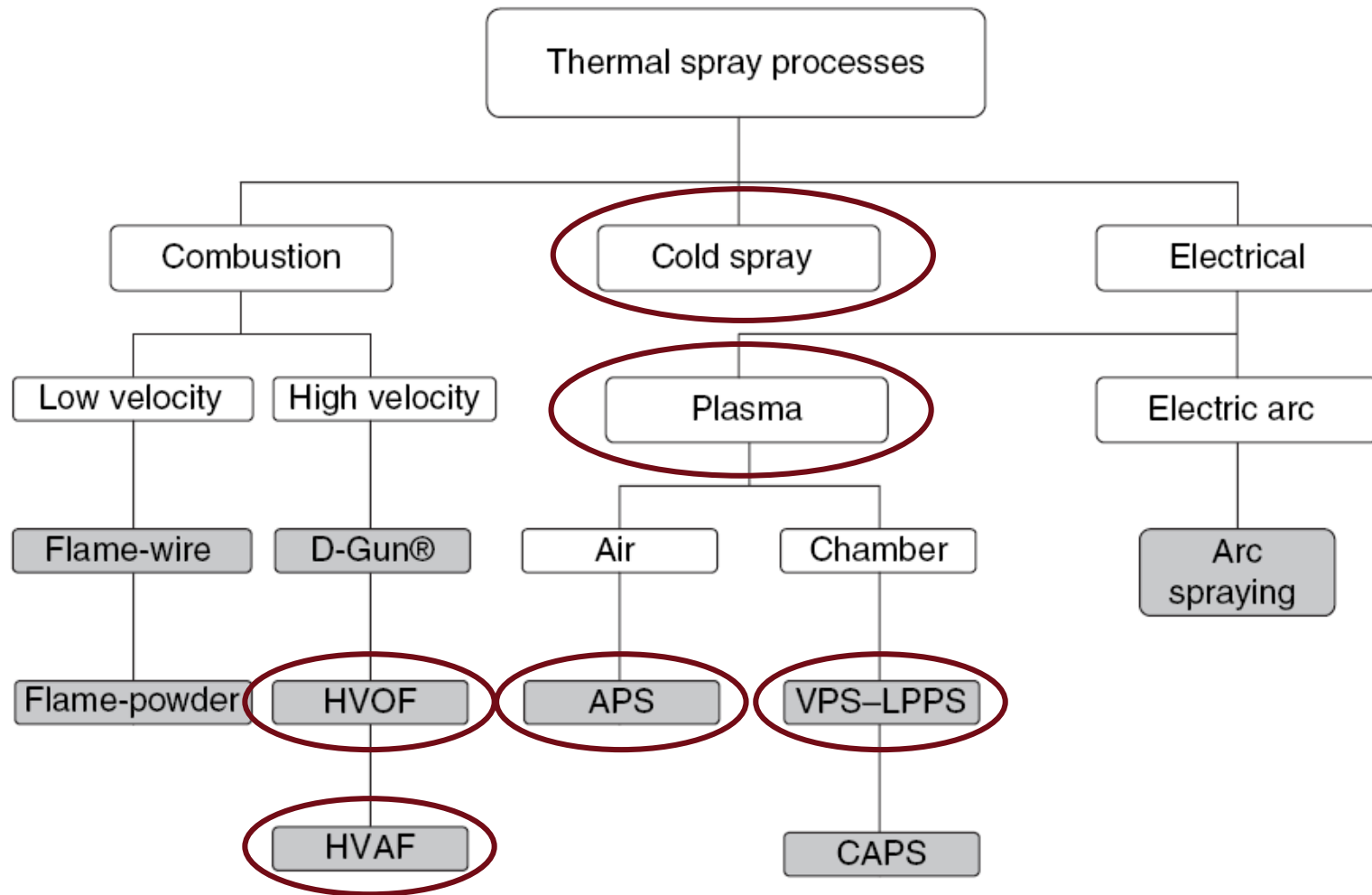
# Surface engineering by thermal spray process

**A thermal spray process can be divided into four sections:**

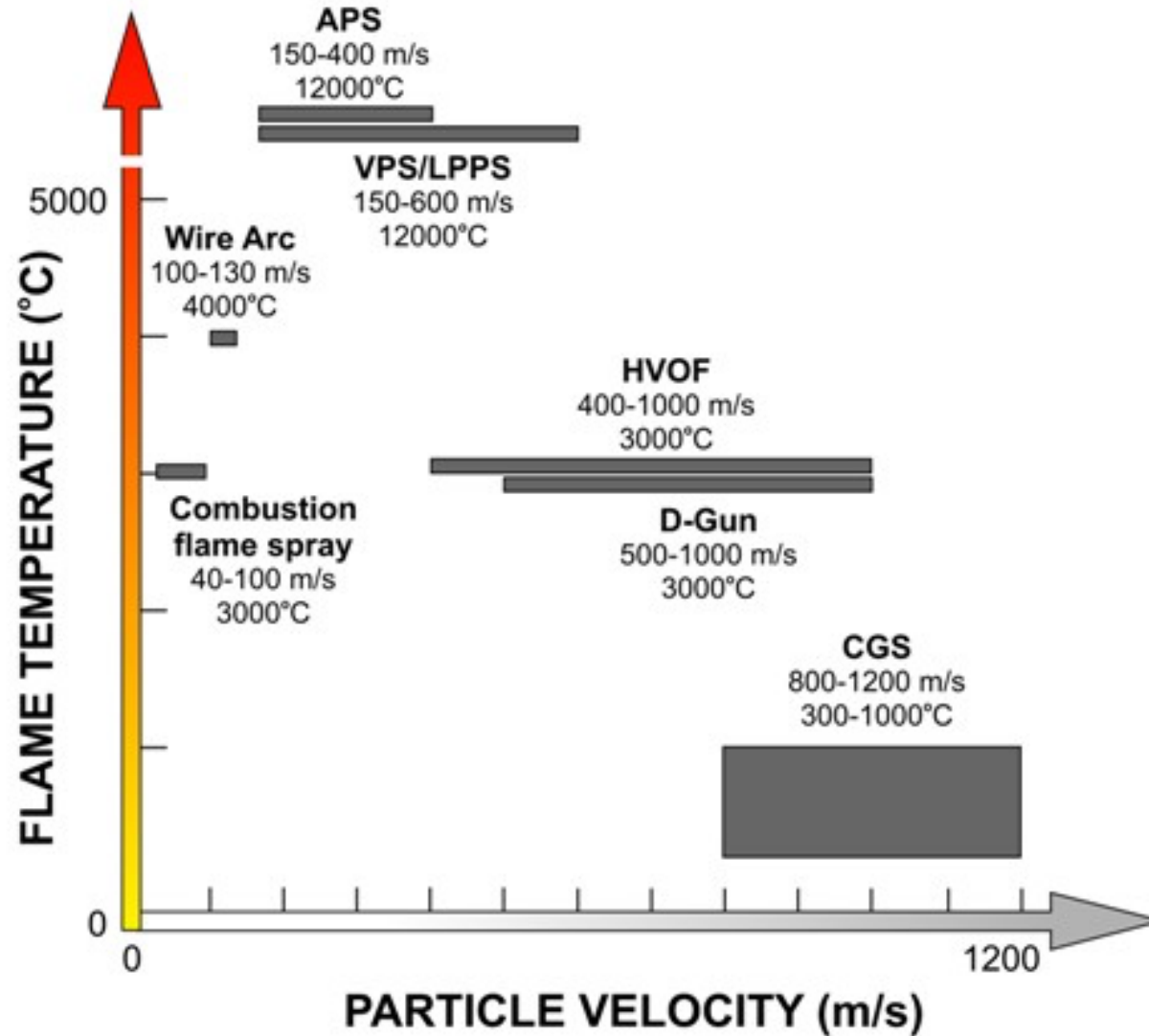
- heat/energy source (thermal energy for heating and melting)
- material feed/flow
- material spray (kinetic energy for propelling dispersion)
- material deposition



# Surface engineering by thermal spray processes

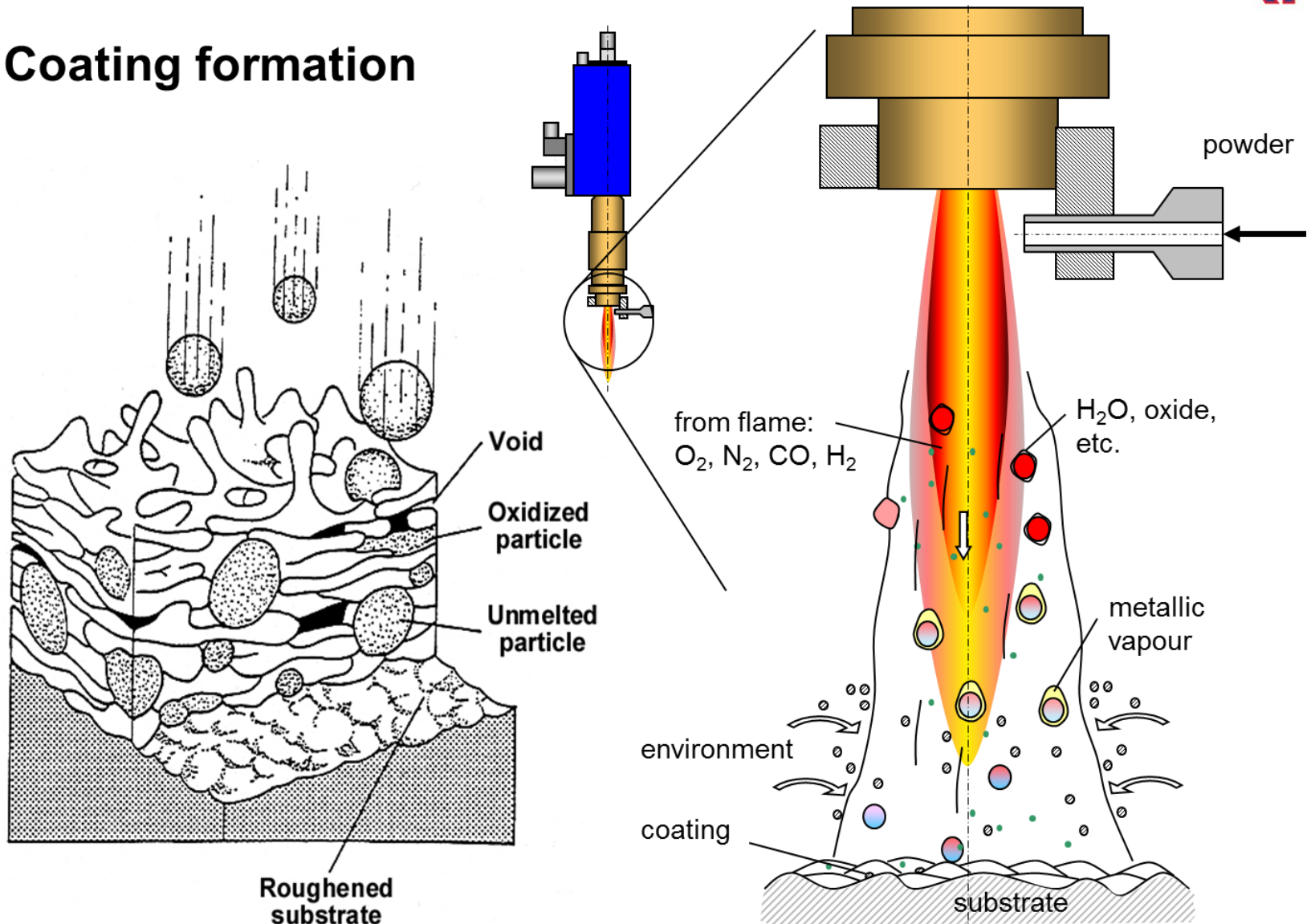


# Surface engineering by thermal spray processes

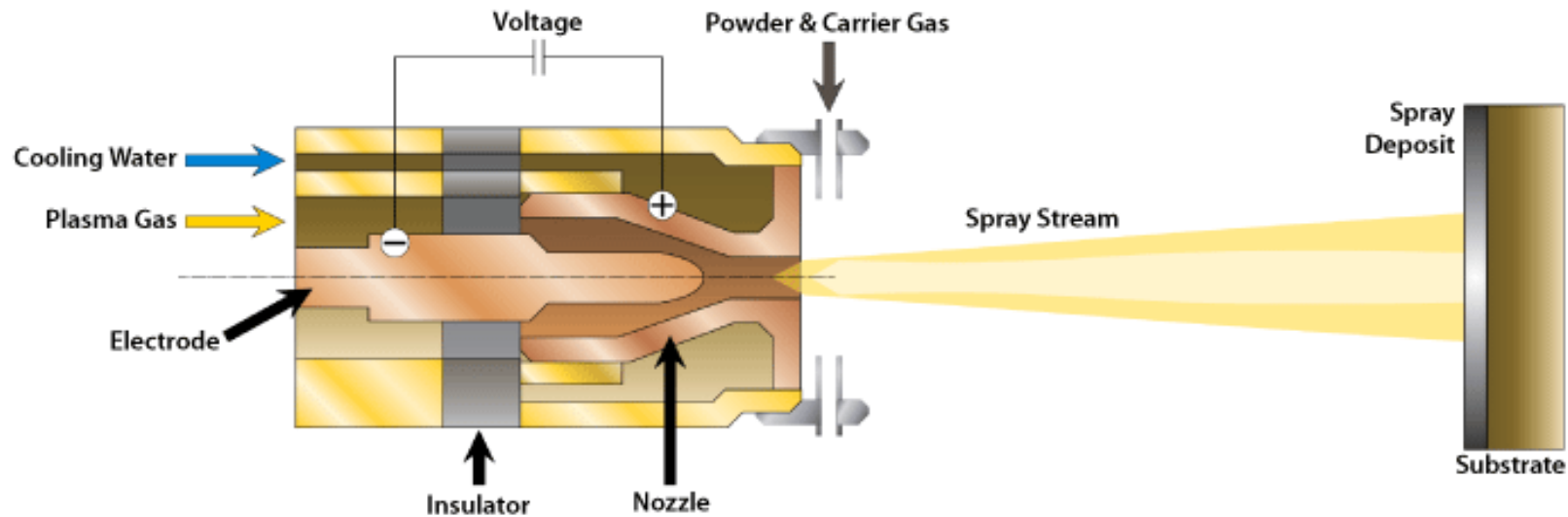




## Coating formation



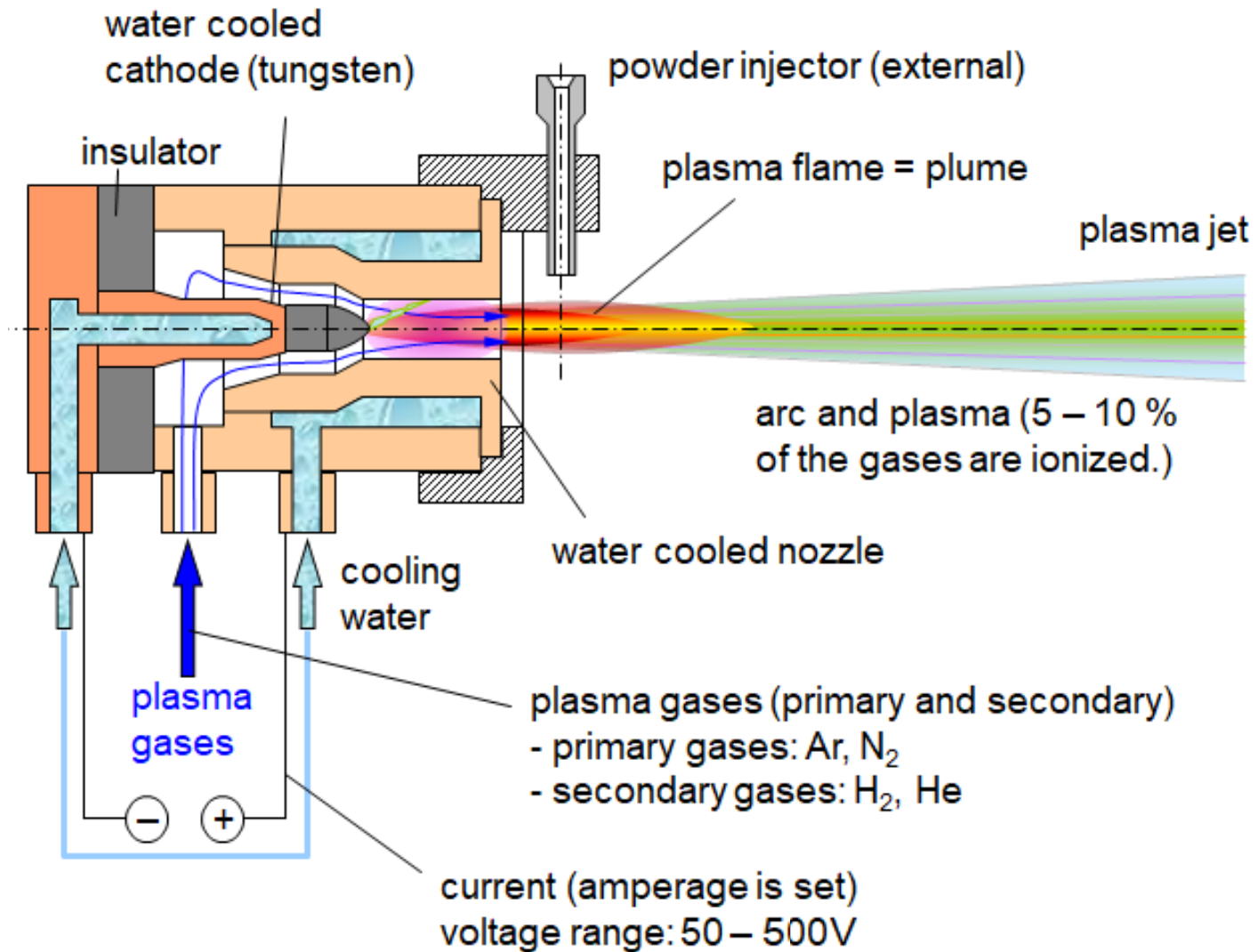
# Plasma spraying



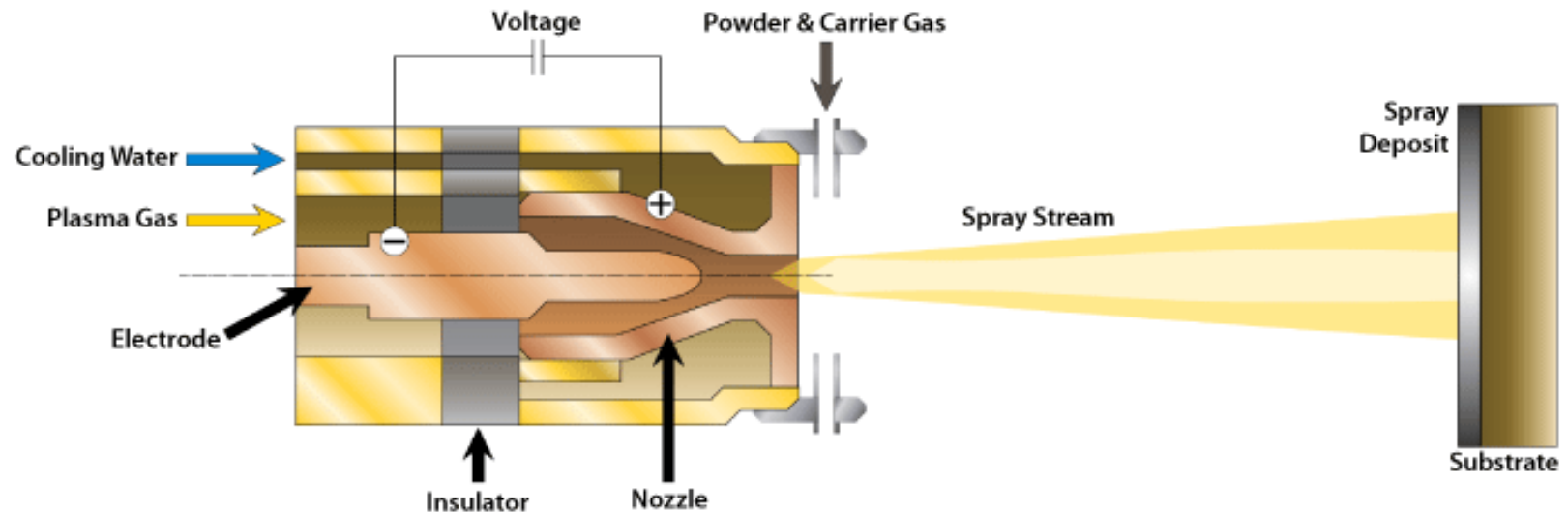
- An inert gas is passed through a potential differential.
- An arc is formed between the two dipoles.
- The gas is ionized and recombines after going through a free expansion.
- The recombination effect releases heat and creates a plasma flame (“plume”).
- Powder is inserted into the flame and is propelled towards the substrate.



# Plasma spraying



# Plasma spraying

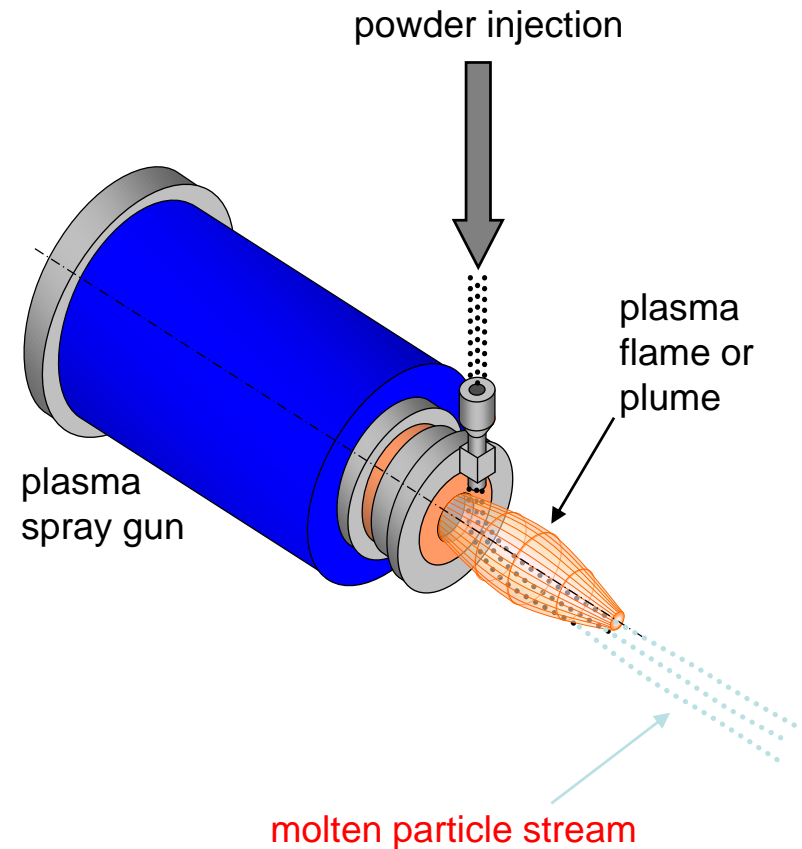


- Max. plume temperature  $\approx 12000^{\circ} \text{C}$
- Impact speed of particles  $200\text{-}400 \text{ m/s}$
- In-flight time of particles  $\approx 10^{-3} \text{ s}$
- Heating/cooling rate  $\approx 10^7 \text{ K/s}$

# Plasma spraying

## Process parameters

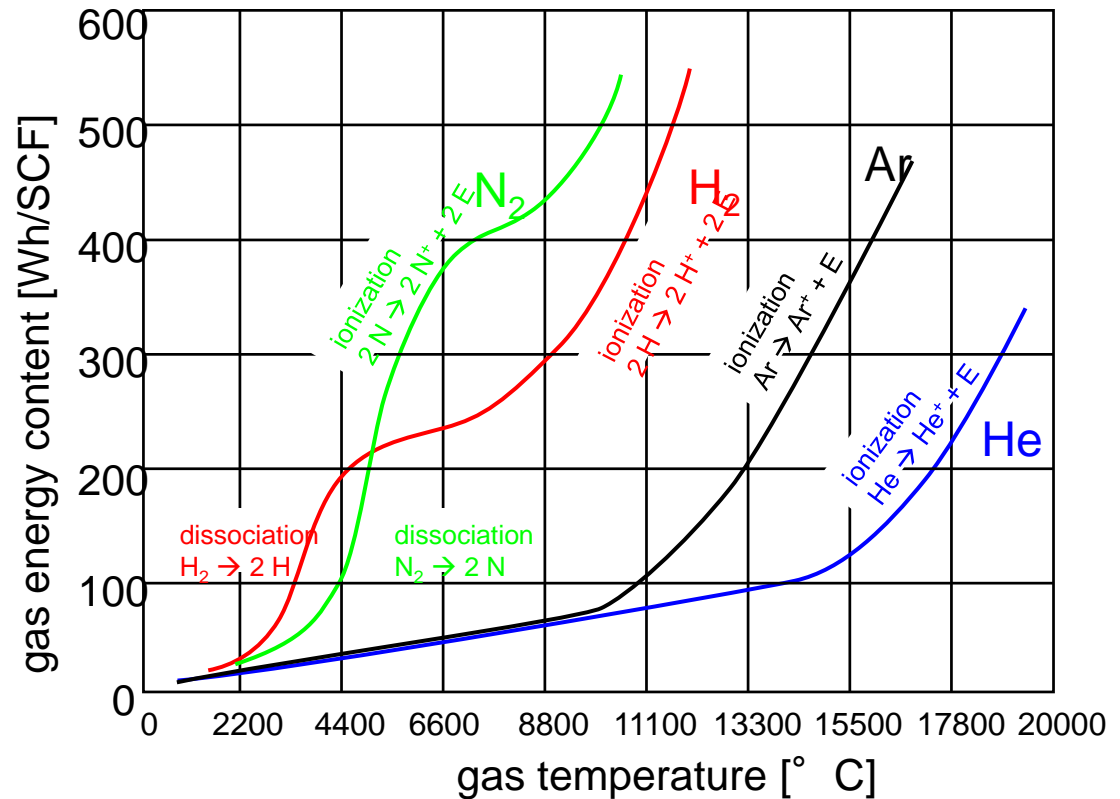
- Plasma (Primary) gas flow
- Secondary gas flow
- Plasma power
- Spray distance
- Injector angle, length and size
- Carrier gas flow
- Gun spray angle



# Plasma spraying

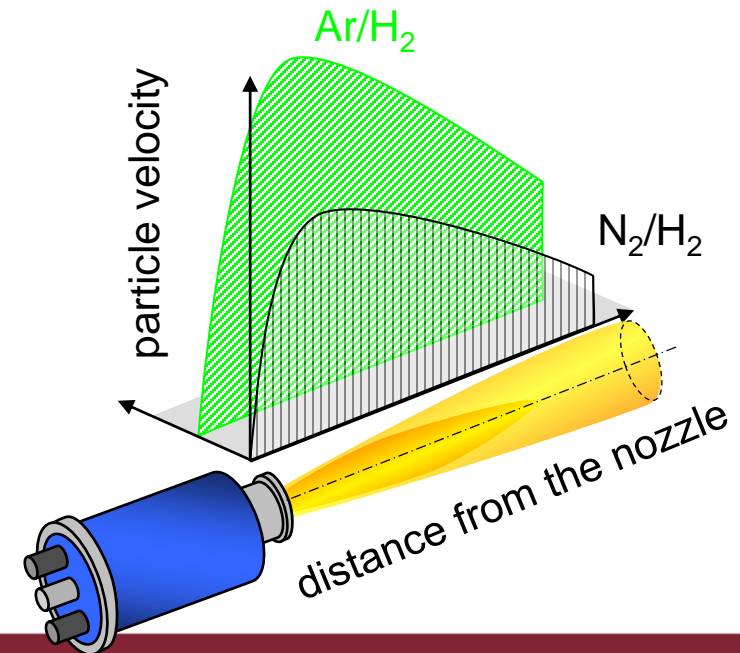
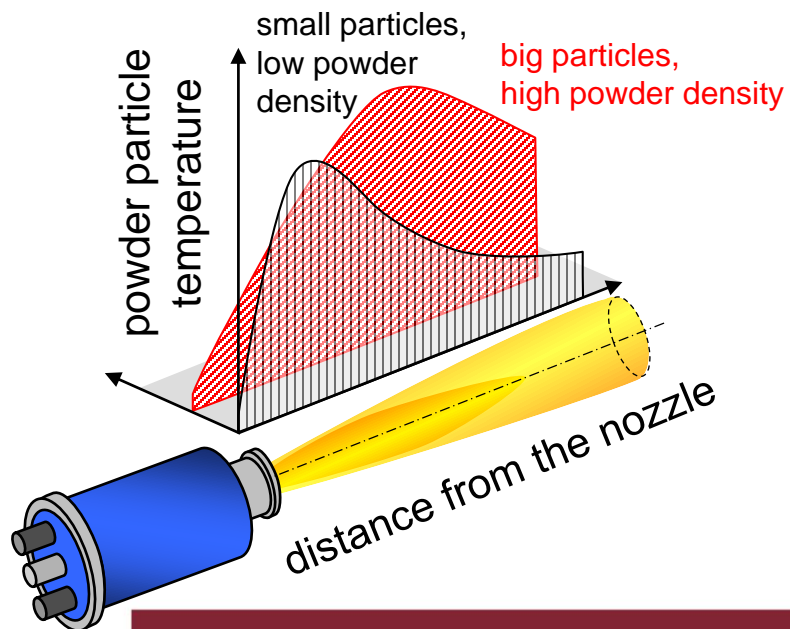
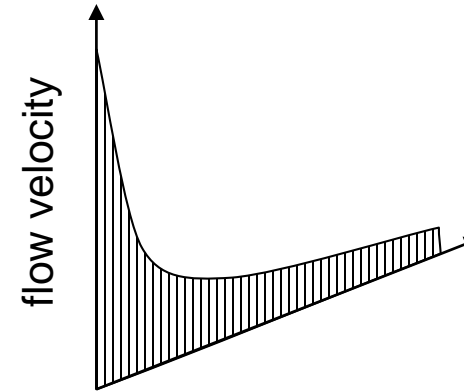
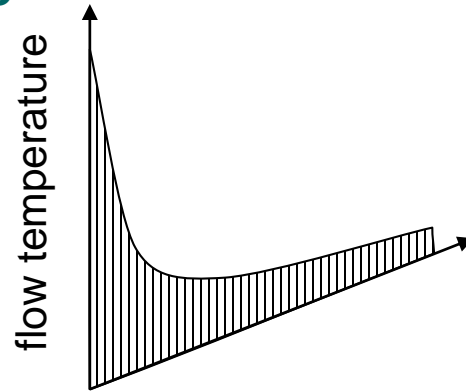
## Plasma gases

- Primary gases (Ar, N<sub>2</sub>): plasma stability and transport properties
- Secondary lighter gases (He, H<sub>2</sub>), enthalpy adjustment.
- N<sub>2</sub> and H<sub>2</sub> are diatomic gases. These plasmas have higher energy contents for a given temperature than argon and helium because of the energy associated with dissociation of molecules.



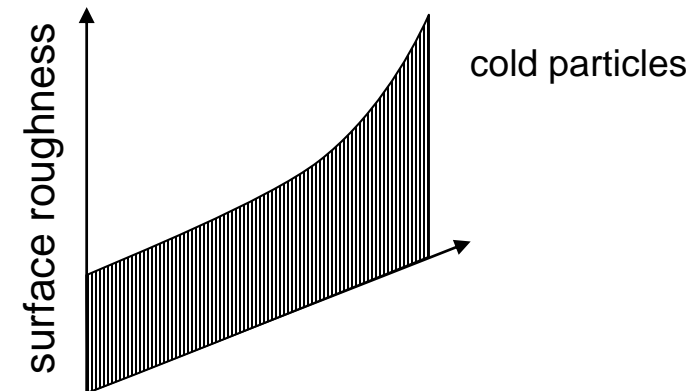
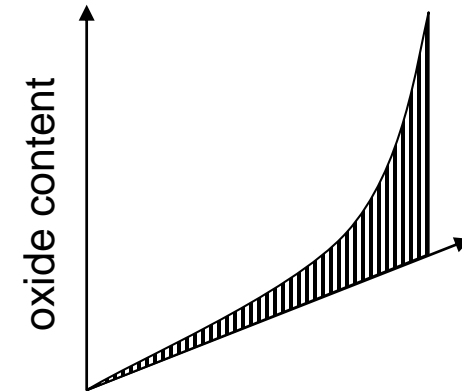
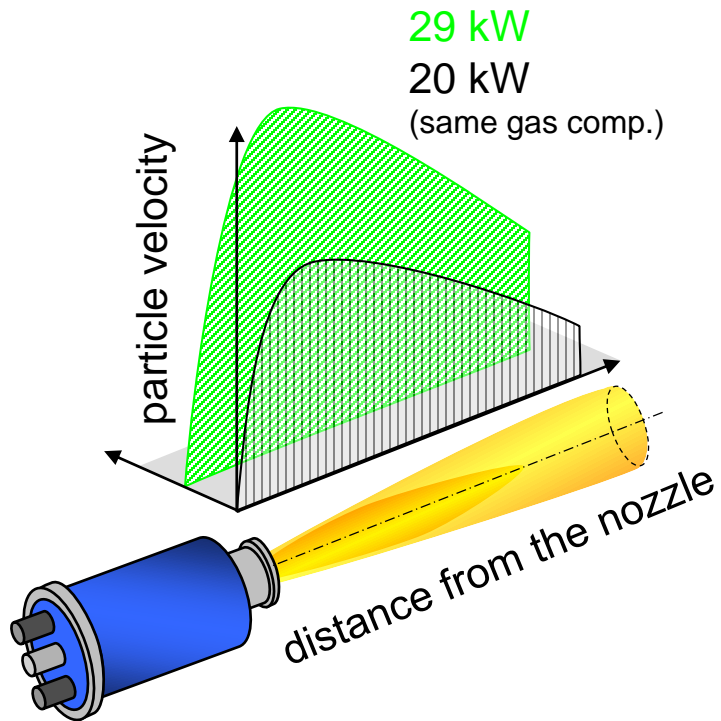
# Plasma spraying

## Spray distance



# Plasma spraying

## Spray distance





# Plasma spraying

## Controlled Atmosphere Plasma Spray

chamber is backfilled after evacuation, either at

- near vacuum (1 - 10 mbar) **VPS**
- reduced pressure ( $> 50$  mbar) **LPPS**
- standard pressure **APS**
- elevated pressure (up to 4 bar) **HPPS**
- with a substitute atmosphere (reactive ( $C_xH_y$ ), inert (Ar)) **RPS / IPS**



## VPS / LPPS / LVPS

- spraying in near vacuum or under reduced pressure conditions
- spray particles are unimpeded by frictional forces of the atmosphere

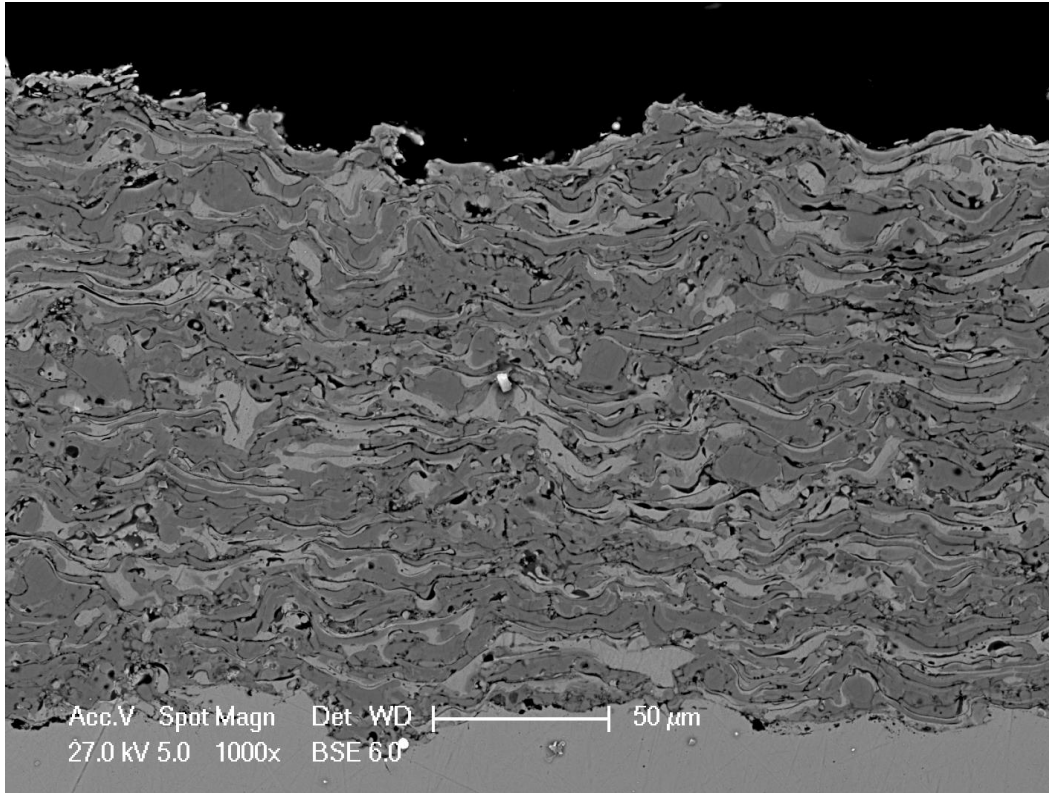
# Plasma spraying

## Materials

- Ceramics (high melting point)
  - $\text{Al}_2\text{O}_3$ ,  $\text{Al}_2\text{O}_3/\text{TiO}_2$ ,  $\text{Cr}_2\text{O}_3$ ,  $\text{ZrO}_2 - \text{Y}_2\text{O}_3$ ,  $\text{La}_2\text{Zr}_2\text{O}_7$ , ecc.
  - $\text{ZrB}_2$ ,  $\text{TiB}_2$ , SiC (non standard)
- Metals (refractory, oxidation control is mandatory, controlled atmosphere)
  - MCrAlY, Ni-and Mo-based alloys
- Cermets
  - $\text{Cr}_3\text{C}_2$  - based (high T, generally NiCr matrix)
  - WC - based (better mechanical properties,  $T_{\text{max}} 500^\circ \text{C}$  (decarburation), Co - CoNiCr - based matrix)

# Plasma spraying

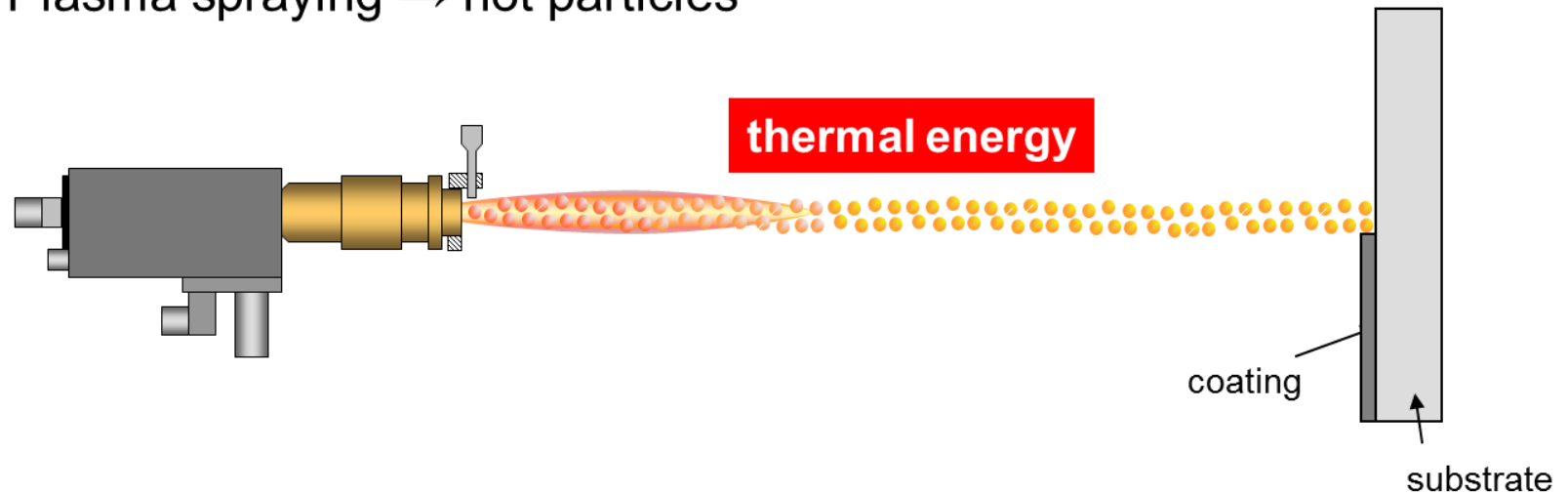
## Cermet $\text{Cr}_3\text{C}_2$ – Ni-Cr



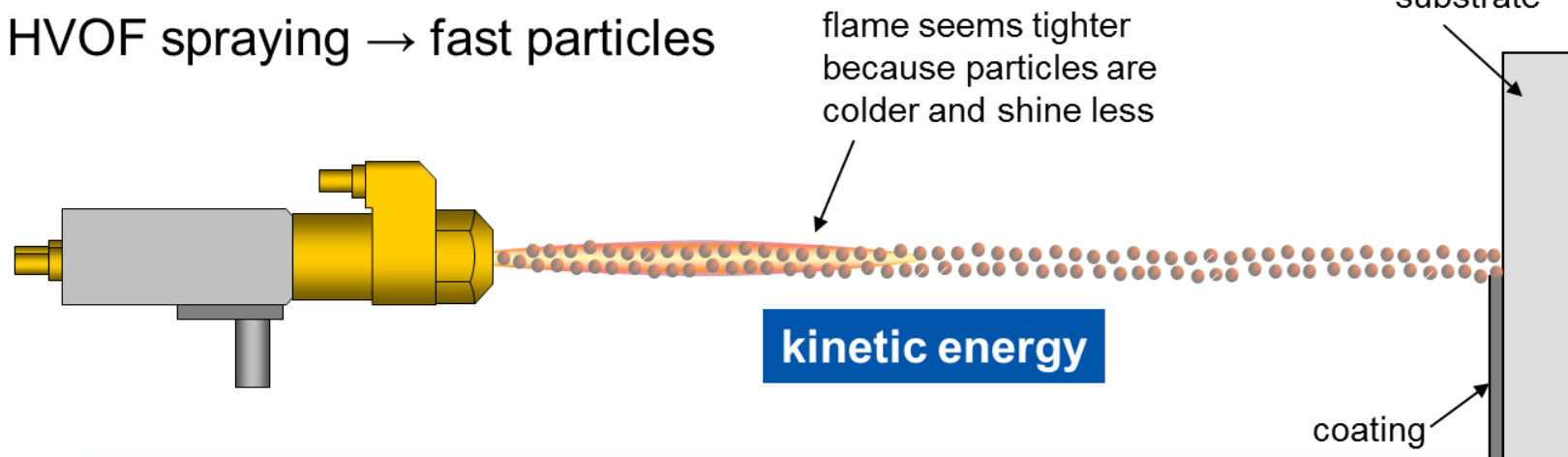
- Low porosity
- Lamellar microstructure, partially molten ceramic phase

# Coating formation – plasma vs. HVOF

Plasma spraying → hot particles



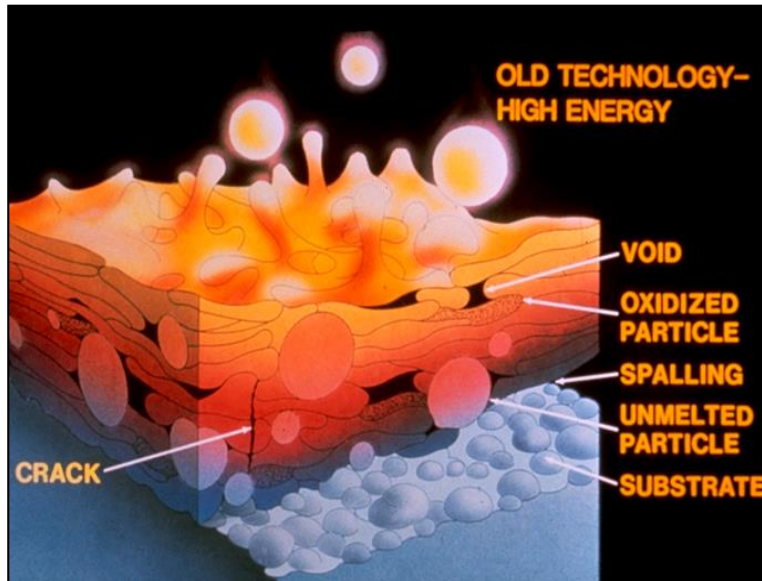
HVOF spraying → fast particles



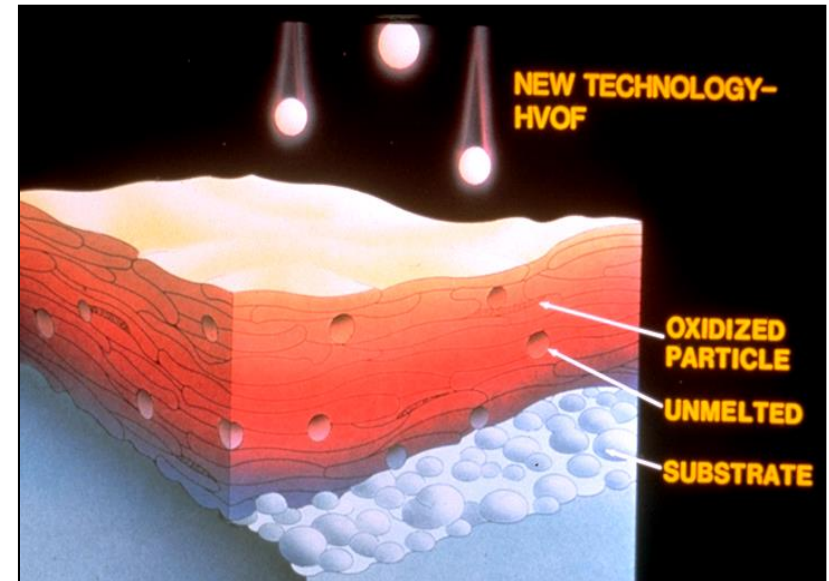
# HVOF deposition

- Max. jet temperature  $\approx 2800^{\circ} \text{C}$
- Particles impact speed  $400\text{-}700 \text{ m/s}$
- In-flight time  $\approx 10^{-3} \text{ s}$
- Heating/cooling rate  $\approx 10^7 \text{ K/s}$

## Plasma Spray



## HVOF





# Applications

## Wear and corrosion resistant coatings

(Mechanical, chemical, aerospace, textile industries)

- Cermet (WC,  $\text{Cr}_3\text{C}_2$  or TiC in Co, Co-Cr or Ni-Cr metal matrix);
- Ceramics ( $\text{Cr}_2\text{O}_3$ ,  $\text{Al}_2\text{O}_3$ -TiO<sub>2</sub>);
- Metallic (Ni or Co base, e.g. NiCrBSiC).



## Thermal barrier coatings (TBC)

(Aerospace, energy industries)

- $\text{ZrO}_2$  -  $\text{Y}_2\text{O}_3$ .



## High temperature oxidation resistant coatings

(Aerospace, energy, chemical industries)

- Ni - Cr - Al base (e.g. NiCoCrAlY).



## Biocompatible coatings

Porous Ti , hydroxy apatite ( $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$ )



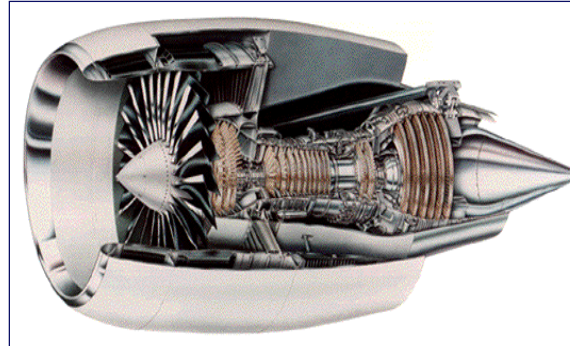


# Materials and Surface Engineering Lab

## Research partners and clients



# Aircraft engines and power gas turbines

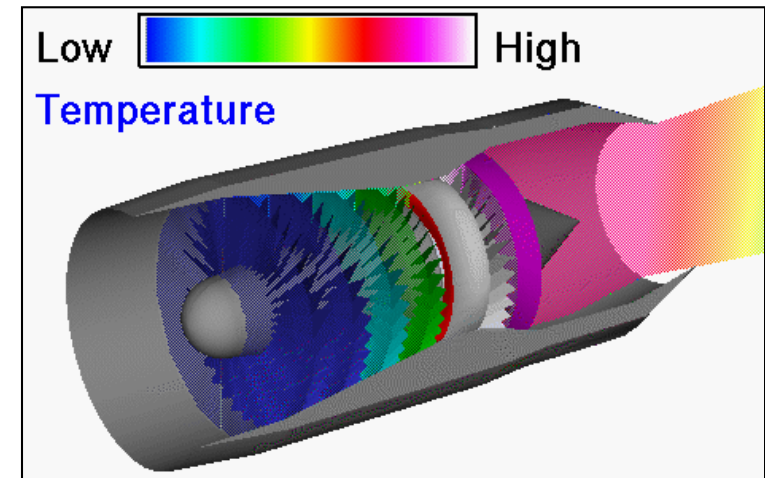


Max gas temperature in turbine:  
900 - 1400 ° C

Rise in gas temperature in turbine



Improvement of thermodynamic  
efficiency



# Turbine blades protection

High temperature and combustion products (e.g.  $\text{SO}_2$ ,  $\text{SO}_3$ ,  $\text{V}_2\text{O}_5$ ) cause surface damages on the Ni-base superalloy

**Hot corrosion**  
(700 – 925 ° C)

( $\text{SO}_3$  attack, sulphides formation)

**High temperature oxidation**  
( $T > 1000^\circ \text{ C}$ )

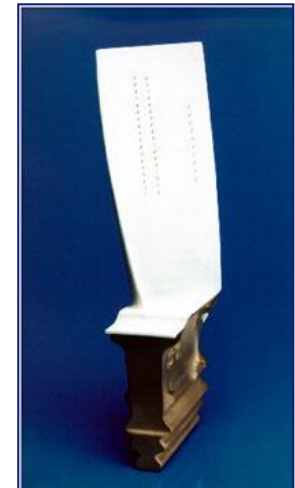
**Erosion**

(Solid particles in gas stream)

**Max service temperature of superalloy!**



**Protective ceramic overlay**  
**Thermal Barrier Coating (TBC)**



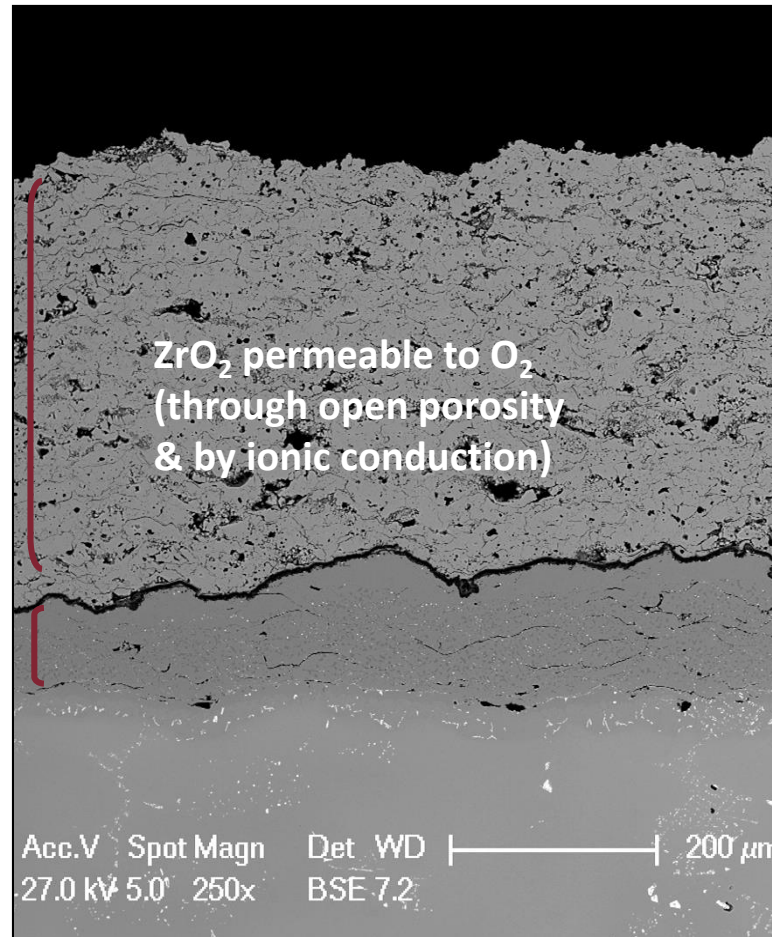
# Thermal Barrier Coatings (TBC)

**Top Coat ( $\text{ZrO}_2\text{--}6\text{--}8\%\text{Y}_2\text{O}_3$ )**

**TGO ( $\alpha\text{-Al}_2\text{O}_3$ )**

**Bond Coat (Ni/Co-CrAlY)**

**Substrate (Ni-base superalloy)**



- Thermal insulation
- Erosion protection

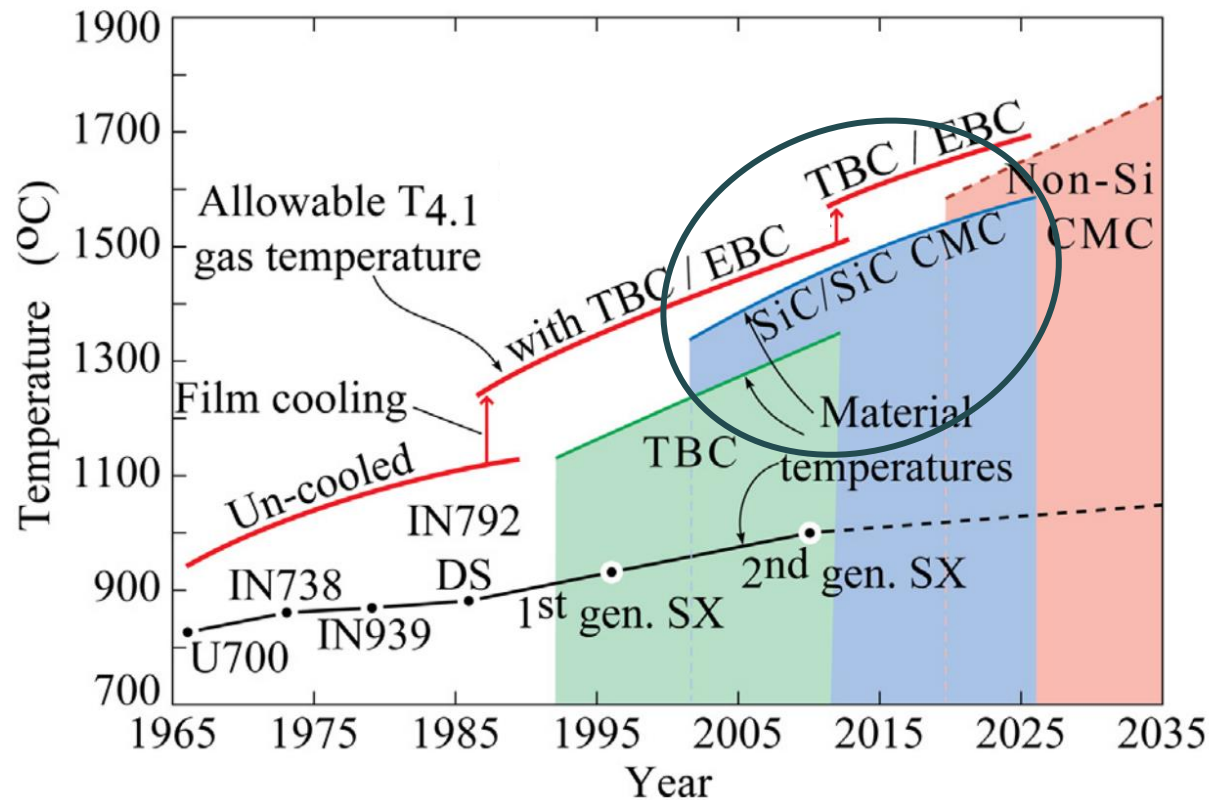
- Oxidation barrier
- Enhanced adhesion
- Hot corrosion resistance

- Thermo-mechanical loading



# Innovative materials for gas turbines

## Strategies to improve turbines performance: use of CMC materials



# Innovative materials for gas turbines

## Strategies to improve turbines performance: use of CMC materials

### Modification of substrate-coating system

(12) **United States Patent**  
**Subramanian et al.**

(10) **Patent No.:** **US 8,114,799 B2**  
(45) **Date of Patent:** **Feb. 14, 2012**

(54) **FUNCTIONALLY GRADIENT SIC/SIC  
CERAMIC MATRIX COMPOSITES WITH  
TAILORED PROPERTIES FOR TURBINE  
ENGINE APPLICATIONS**

(75) Inventors: **Suresh Subramanian**, Mason, OH (US);  
**James Steibel**, Mason, OH (US);  
**Douglas Carper**, Trenton, OH (US);  
**Toby Darkins, Jr.**, Loveland, OH (US)

(73) Assignee: **General Electric Company**,  
Schenectady, NY (US)

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# Innovative materials for gas turbines

**Strategies to improve turbines performance:  
use of CMC materials**

## **Modification of substrate-coating system**

GE Rolls-Royce – development of F136 engine (second choice for F-35 aircraft)

3rd low pressure stage – stator in SiC/SiC



# Innovative materials for gas turbines

**Strategies to improve turbines performance:  
use of CMC materials**

**Modification of substrate-coating system**



GE Aviation

## GE Successfully Tests World's First Rotating Ceramic Matrix Composite Material for Next-Gen Combat Engine

F414 low-pressure turbine blades prove silicon carbide CMC material for unprecedented deployment in GE's adaptive cycle combat engine

February 10, 2015

CINCINNATI, OH – February 10, 2015 – GE Aviation successfully tested the world's first non-static set of light-weight, ceramic matrix composite (CMC) parts by running rotating low-pressure turbine blades in a F414 turbofan demonstrator engine designed to further validate the heat-resistant material for high-stress operation in GE's next-generation Adaptive Engine Technology Demonstrator (AETD) program currently in development with the United States Air Force Research Lab (AFRL).

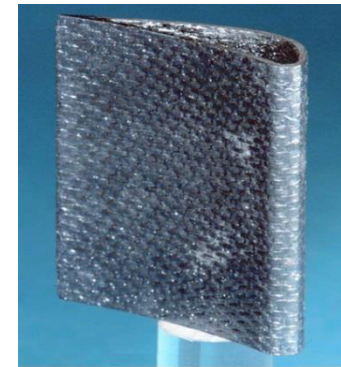
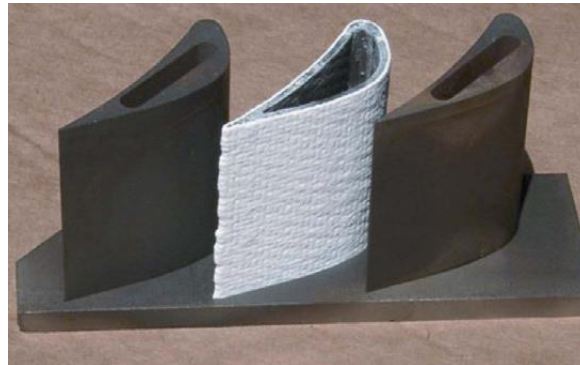
# Innovative materials for gas turbines

Strategies to improve turbines performance:  
use of CMC materials

## Modification of substrate-coating system

Problem:

- Active oxidation of SiC/SiC, especially in presence of H<sub>2</sub>O steam: Thermal & Environmental Barrier Coatings

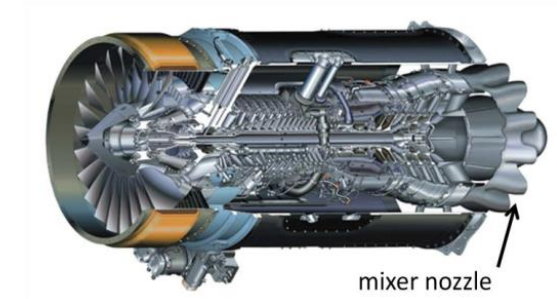
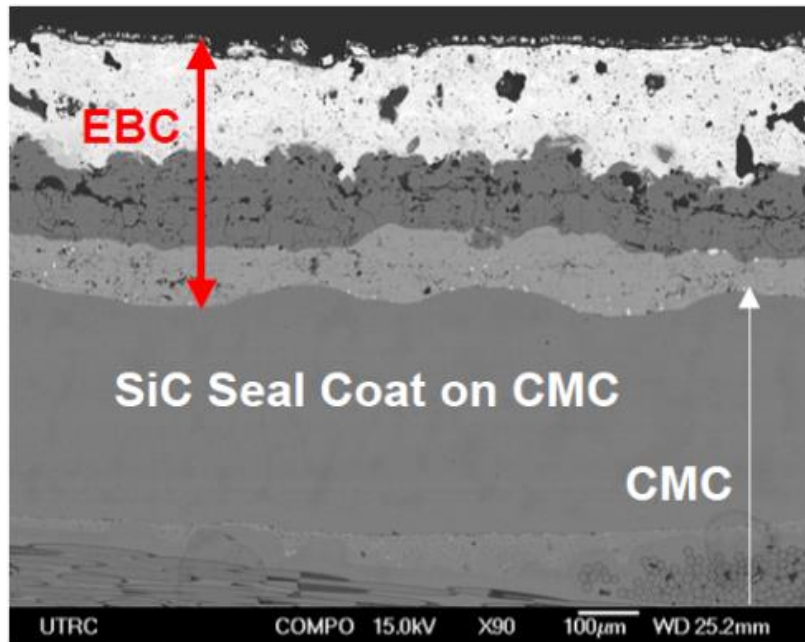


# Innovative materials for gas turbines

## Environmental Barrier Coatings

### *EBC deposited by APS (examples)*

- $\text{HfO}_2\text{-Y}_2\text{O}_3\text{-Gd}_2\text{O}_3\text{-Yb}_2\text{O}_3$
- Barium-Strontium-Alumino-Silicate (BSAS)



CMC shroud for 7F fleet helps increase GT output, reduce heat

# Need for innovation in surface engineering

## Case study: Hard chrome replacement

Hard Chrome plating is an electrolytic method of depositing chrome for engineering applications, from a chromic acid solution.

### Cr <sup>6+</sup> compounds

#### EU classification





COMMISSION REGULATION (EU) No 348/2013

of 17 April 2013

amending Annex XIV to Regulation (EC) No 1907/2006 of the European Parliament and of the Council on the Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH)

Substance	Intrinsic property(ies) referred to in Article 57	Transitional arrangements
		Sunset date <sup>(2)</sup>
Chromium trioxide EC No: 215-607-8 CAS No: 1333-82-0	Carcinogenic (category 1A)  Mutagenic (category 1B)	21 September 2017
Acids generated from chromium trioxide and their oligomers	Carcinogenic (category 1B)	21 September 2017

This Regulation shall be binding in its entirety and directly applicable in all Member States.

Done at Brussels, 17 April 2013.

*For the Commission*

*The President*

José Manuel BARROSO



# Need for innovation in surface engineering

## Hard chrome replacement

### Properties of hard chrome:

- **LOW COEFFICIENT OF FRICTION** –  
coefficient against steel of 0.16 (0.21 dry),
- **HIGH HARDNESS**  
Typical values of 850 - 1050 HV
- **WEAR RESISTANCE**  
extremely good resistance to abrasive  
and erosive wear
- **CORROSION RESISTANCE**  
extremely high resistance to atmospheric  
oxidation, and a good resistance to most  
oxidising and reducing agents



- **MACHINING**  
Can be successfully finished

## Case study

### Hard chrome replacement in marine engine components



# WÄRTSILÄ

**Wärtsilä aims to be the leader in power solutions for the global marine markets**



# Case study

## Hard chrome replacement in marine diesel engine components





# Case study

## Hard chrome replacement in marine diesel engine components



# Case study

## Hard chrome replacement in marine diesel engine components

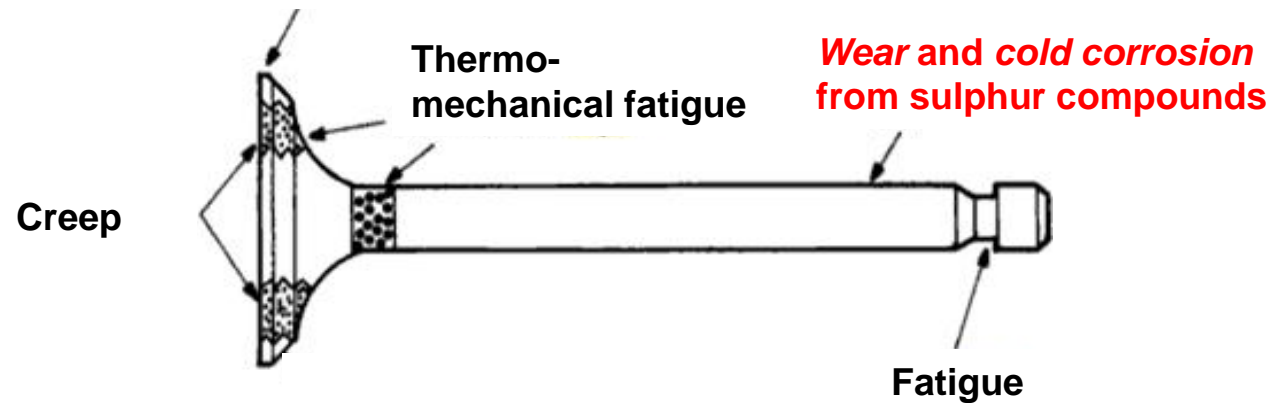


### Valves problems:

- High temperature
- Wear
- Cold and hot corrosion

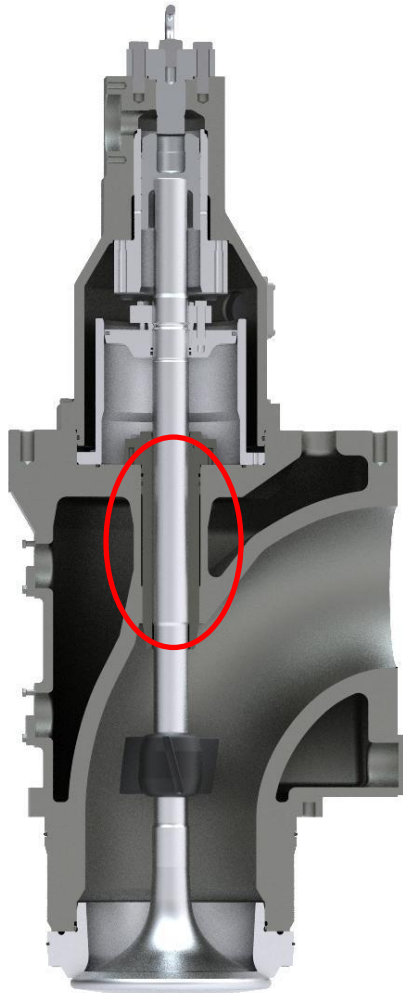


### Hot-corrosion and oxidation



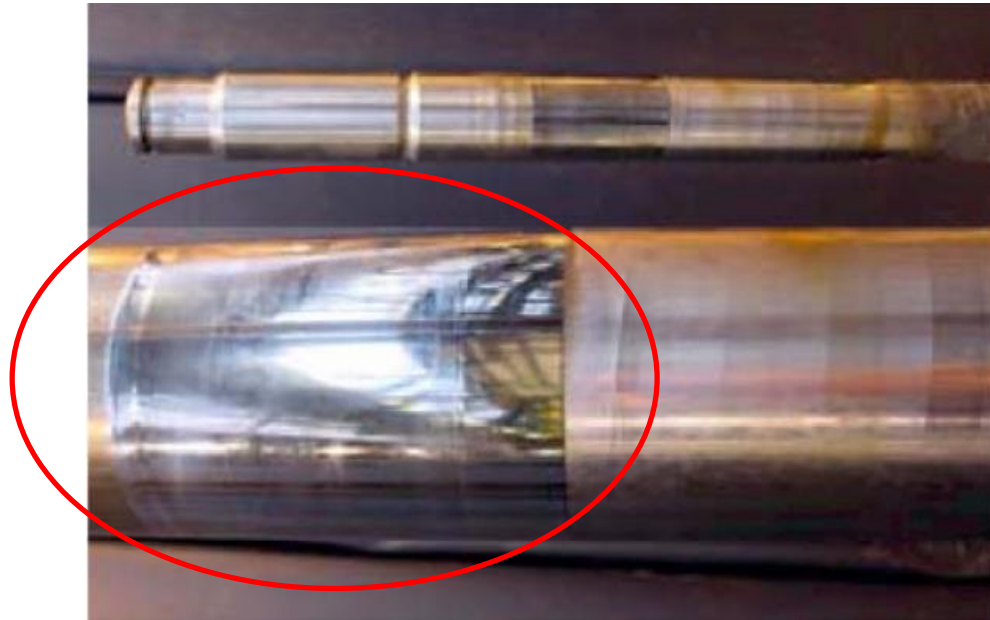
## Case study

### Hard chrome replacement in marine diesel engine VALVES



#### **Problem:**

*Wear and cold corrosion in exhaust and intake valves stem*



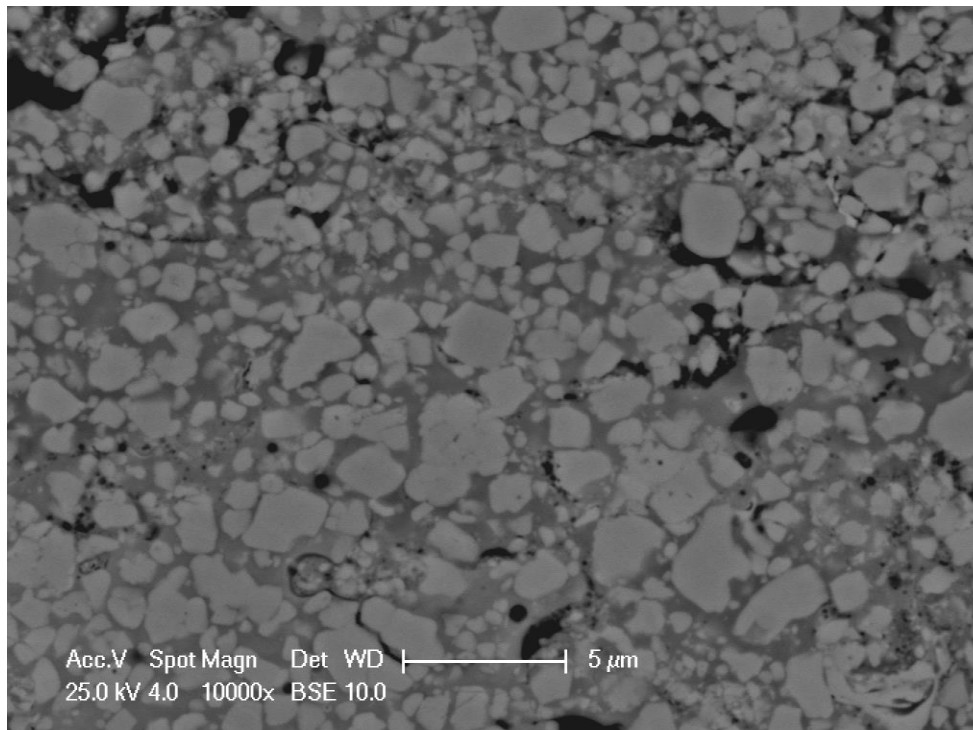
**Hard chrome on stem has a short service-life**



## Case study

### Hard chrome replacement in marine diesel engine valves

#### Ceramic-metal (cermet) HVOF coatings



**2 phases - microstructure**

**Hard ceramic phase:**  
**wear resistance**

**Metal matrix:**  
**toughness and corrosion resistance**

## Case study

Hard chrome replacement in marine diesel engine valves

Materials (Sulzer powders):

✓ WC-CoCrNi **WCN**

✓ (WC-Co)-NiCrSiFeBC **WSF**

✓  $\text{Cr}_3\text{C}_2 - 25$  (Ni-Cr) **CRC**

Facility:

HVOF – JP5000 gun



# Wärtsilä requirements and constrain

## Cermet HVOF coatings

- Thickness higher than 150  $\mu\text{m}$
- Porosity as low as possible
- Hardness higher than 900 HV
- High deposition efficiency



Optimization of deposition  
parameters by  
*Design of Experiment (DoE)*

### 2 factors

1) Kerosene -  $\text{O}_2$  flow

2) Spray distance

### 3 levels

1) Low

2) Medium

3) High

### 3<sup>2</sup> parameter sets

		Keros. (gph) – $\text{O}_2$ (scfh)		
		5,5-1700	6-1850	6,5-2000
Distance (mm)	320			
	355			
	380			

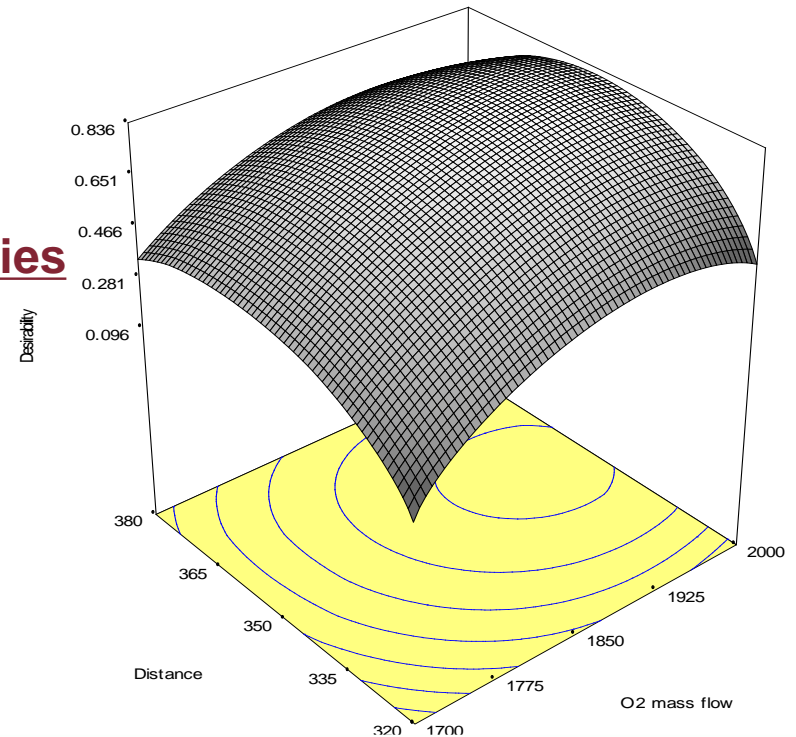
# Phase 1- HVOF coatings optimization (DoE)

## Investigated properties

- Deposition efficiency
- Porosity
- Hardness

as a function of total flow and spray distance

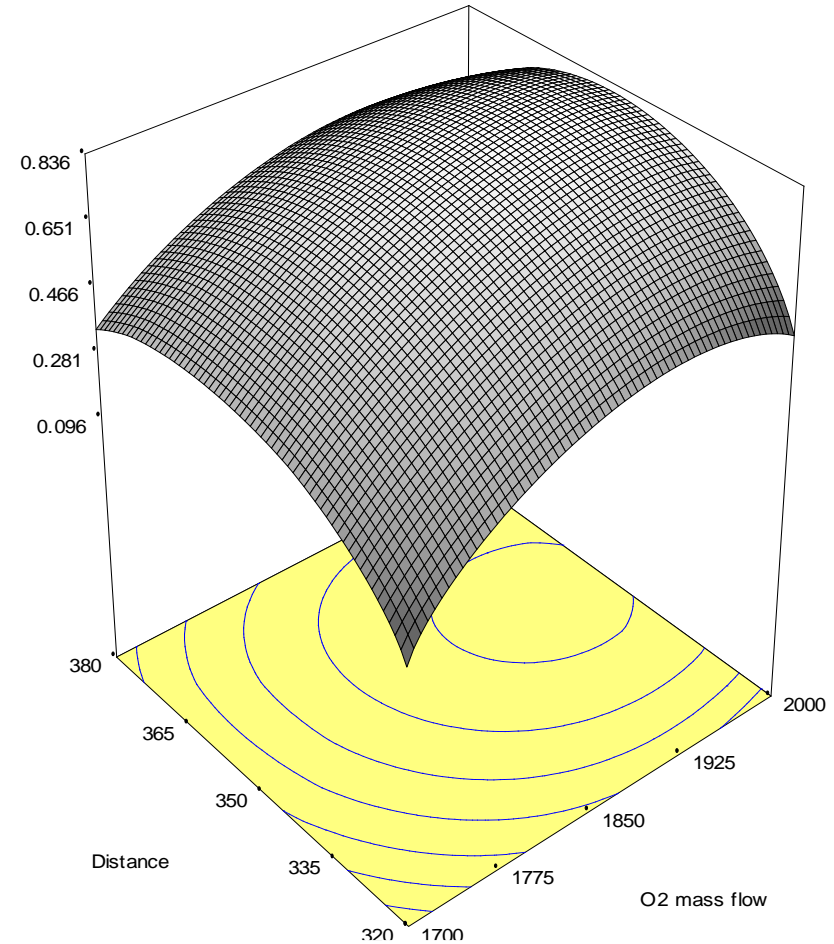
Weighted combination of these properties  
provide the “desirability function”



# WC-CoCrNi

Properties	Optimized value
Deposition efficiency (um/pass)	24,3
Hardness (HV <sub>100</sub> )	1610
Porosity (%)	3,04
<b>Desirability</b>	<b>0,84</b>

Desirability



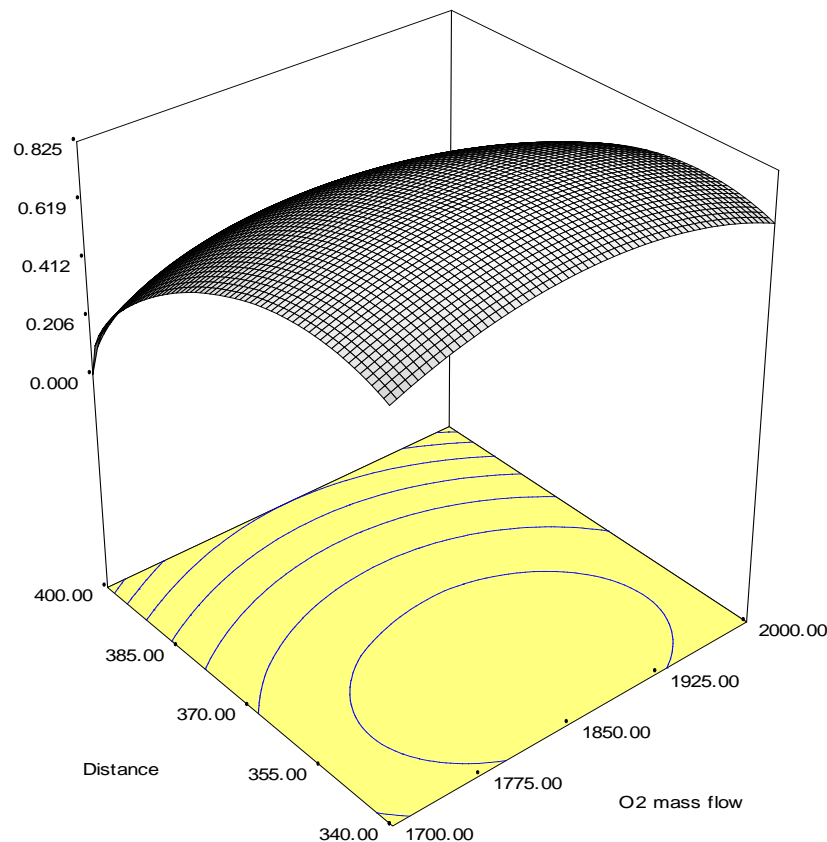
		Keros. (gph) – O <sub>2</sub> (scfh)		
		5,5-1700	6-1850	6,5-2000
Distance (mm)	320			
	355			
	380			



Optimized parameters

# (WC-Co) - NiCrSiFeBC

Properties	Optimized value
Deposition efficiency (um/pass)	10,8
Hardness (HV <sub>50</sub> )	1053,5
Porosity (%)	1,79
<b>Desirability</b>	<b>0,82</b>



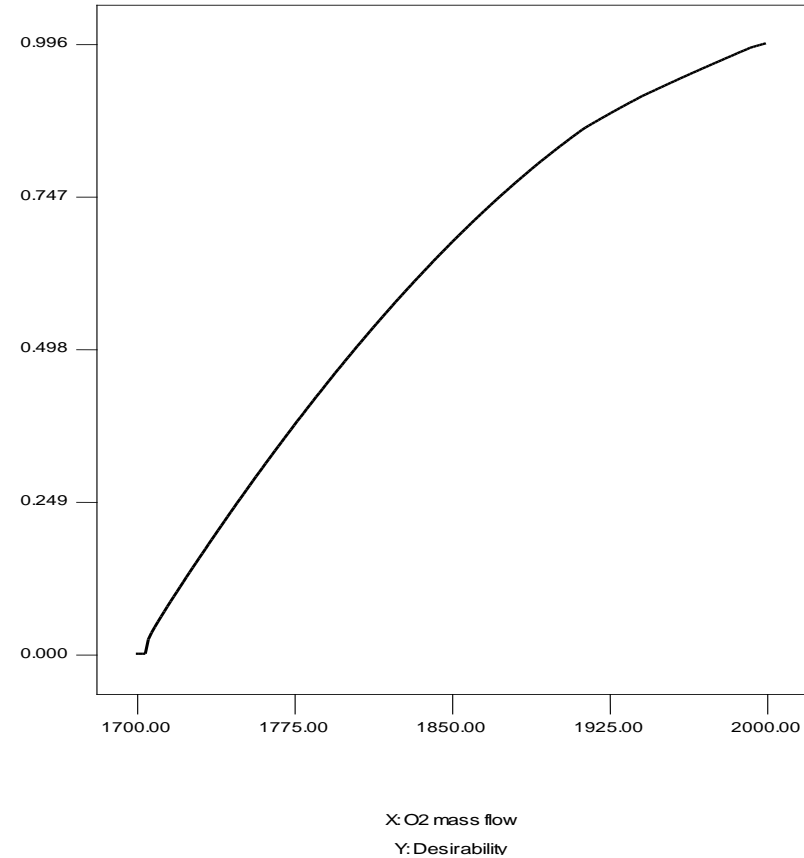
		Keros. (gph) – O <sub>2</sub> (scfh)		
		5,5-1700	6-1850	6,5-2000
Distance (mm)	400			
	370			
	340			

Optimized parameters



# Cr<sub>3</sub>C<sub>2</sub> – 25 (Ni-Cr)

Properties	Optimized value
Deposition efficiency (um/pass)	34,3
Hardness (HV <sub>100</sub> )	1369
Porosity (%)	4,06
<b>Desirability</b>	<b>0,996</b>



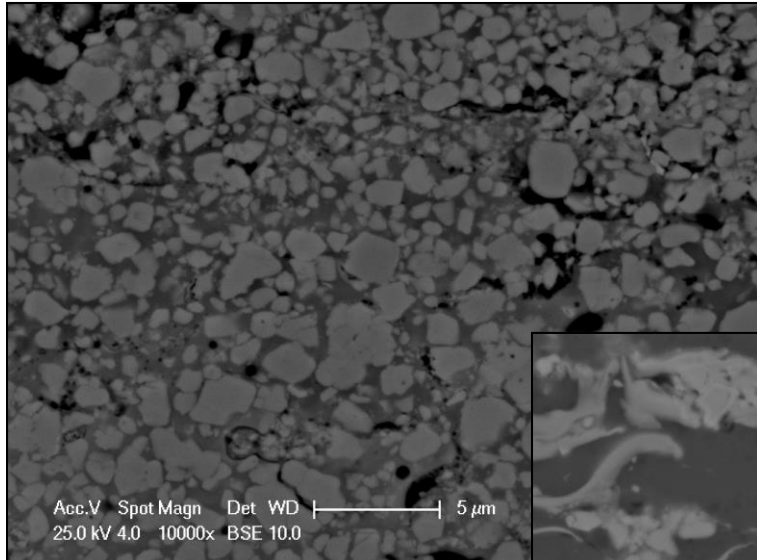
		Keros. (gph) – O <sub>2</sub> (scfh)		
		5,5-1700	6-1850	6,5-2000
Distance (mm)	320			
	355			
	380			



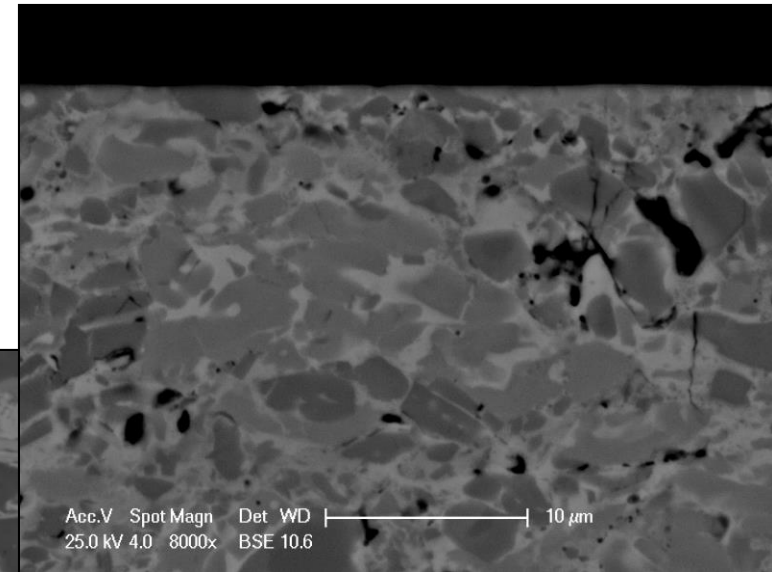
Optimized parameters

# Phase 1- HVOF coatings optimization (DoE)

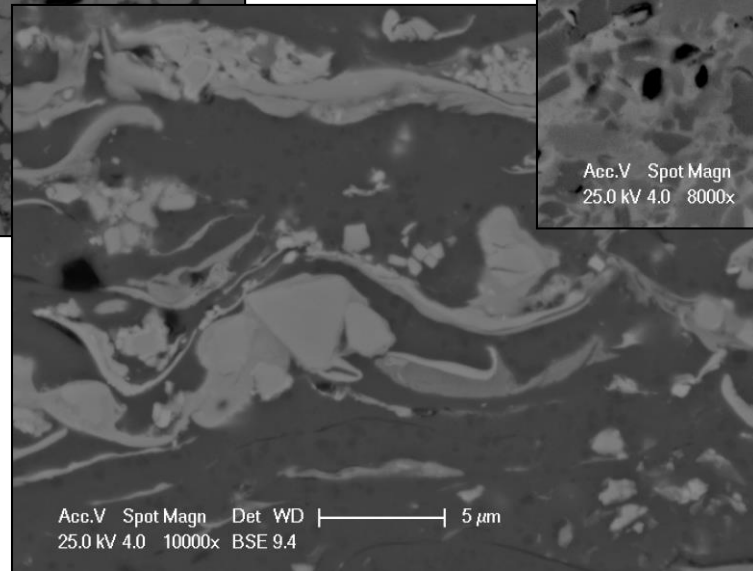
## Optimized coatings



**WCN**



**CRC**



**WSF**

## Phase 2- optimized coatings characterization

### Corrosion tests



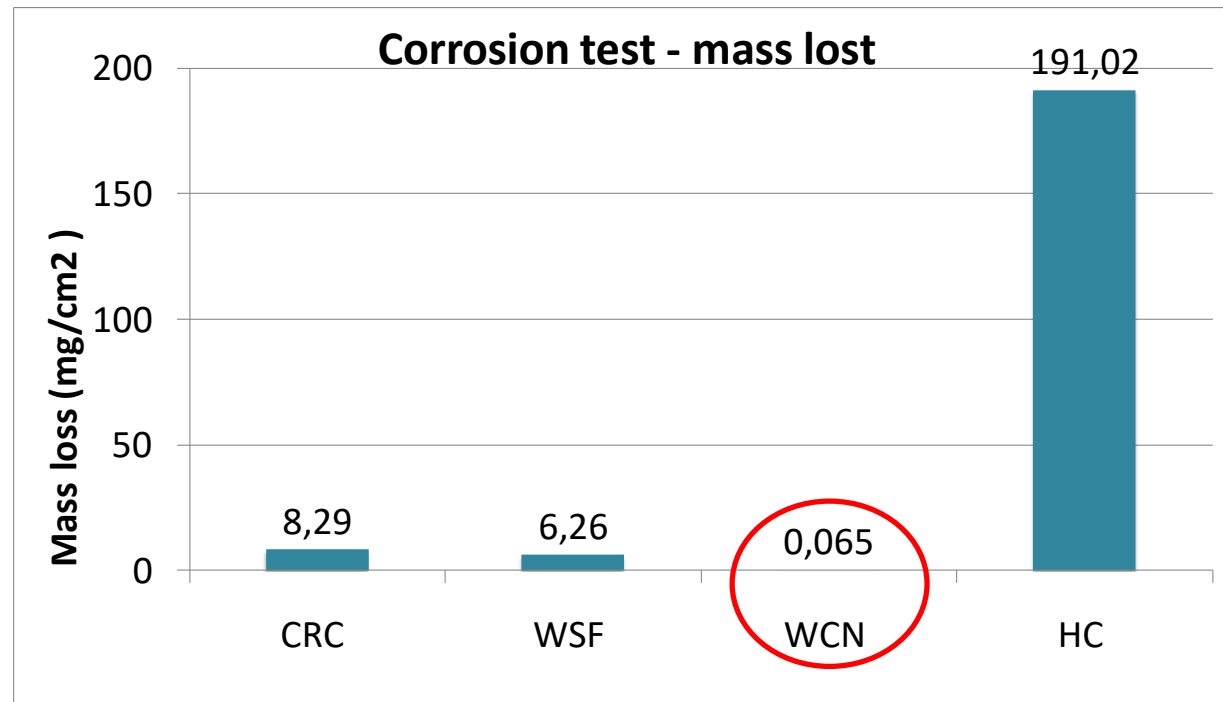
### Wear tests



## Phase 2- optimized coatings characterization

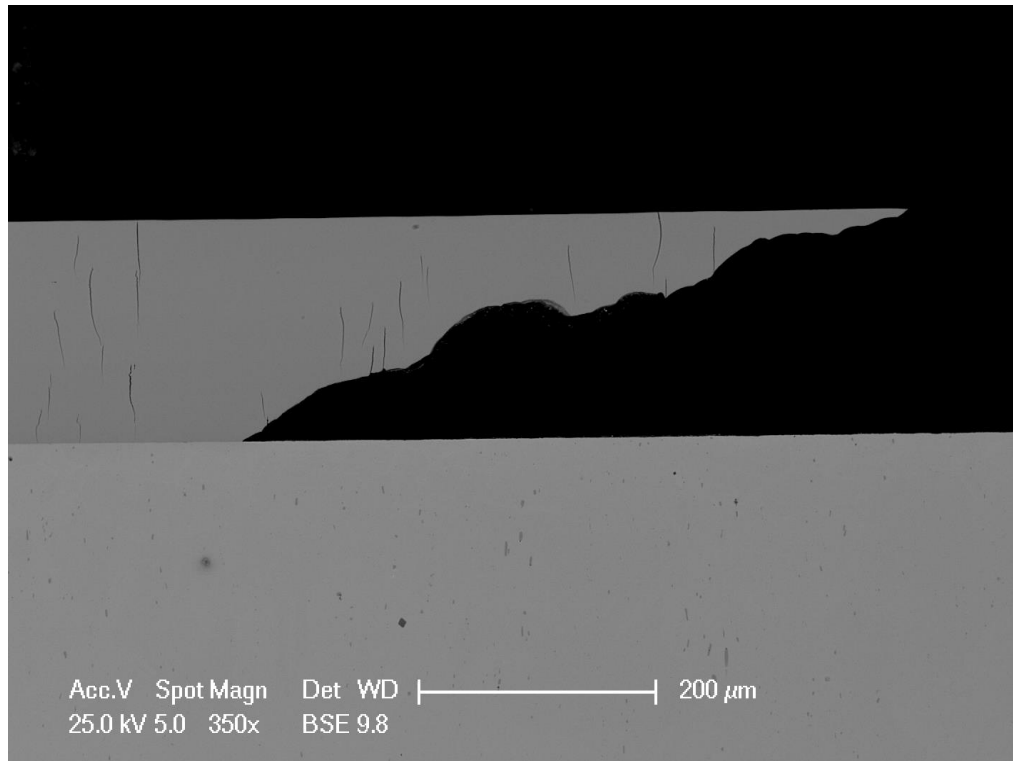
### Corrosion tests

1 hour immersion test – boiling 5%  $\text{H}_2\text{SO}_4$  solution



## Phase 2- corrosion tests

### Hard chrome

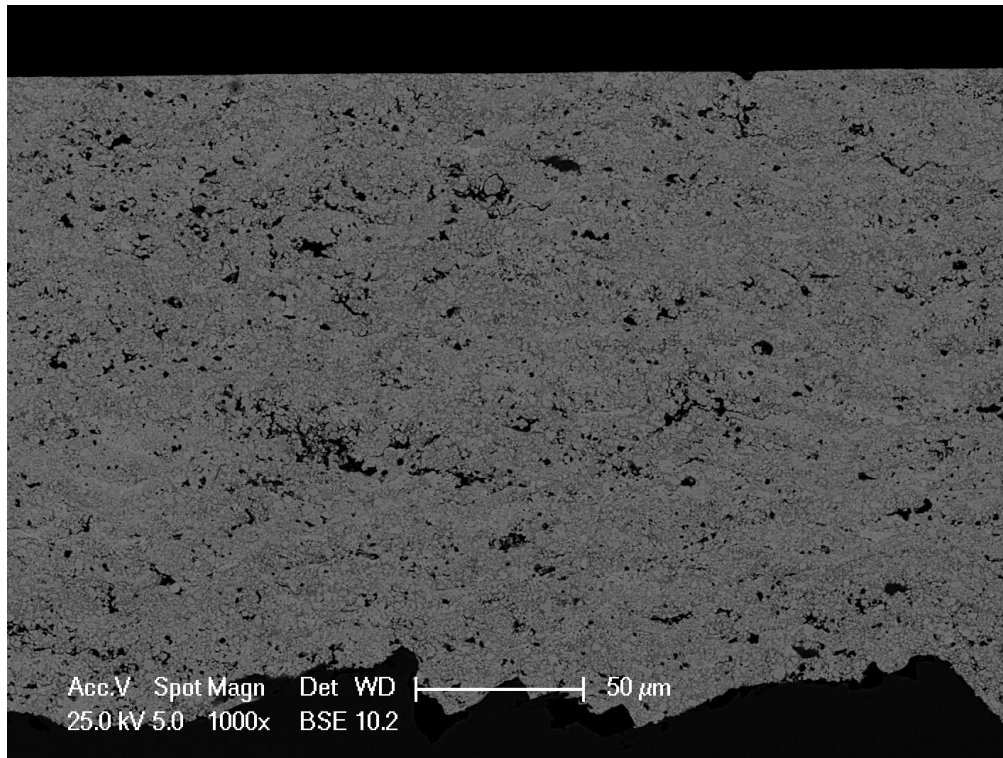


**Not protective**



## Phase 2- corrosion tests

### WCN coatings



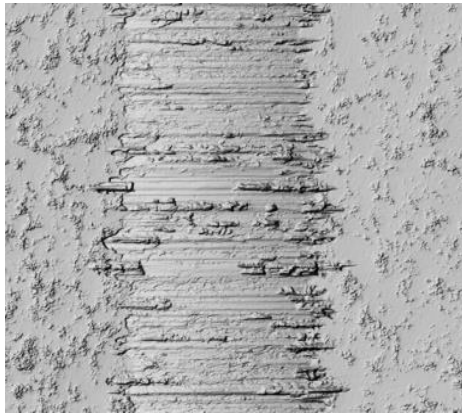
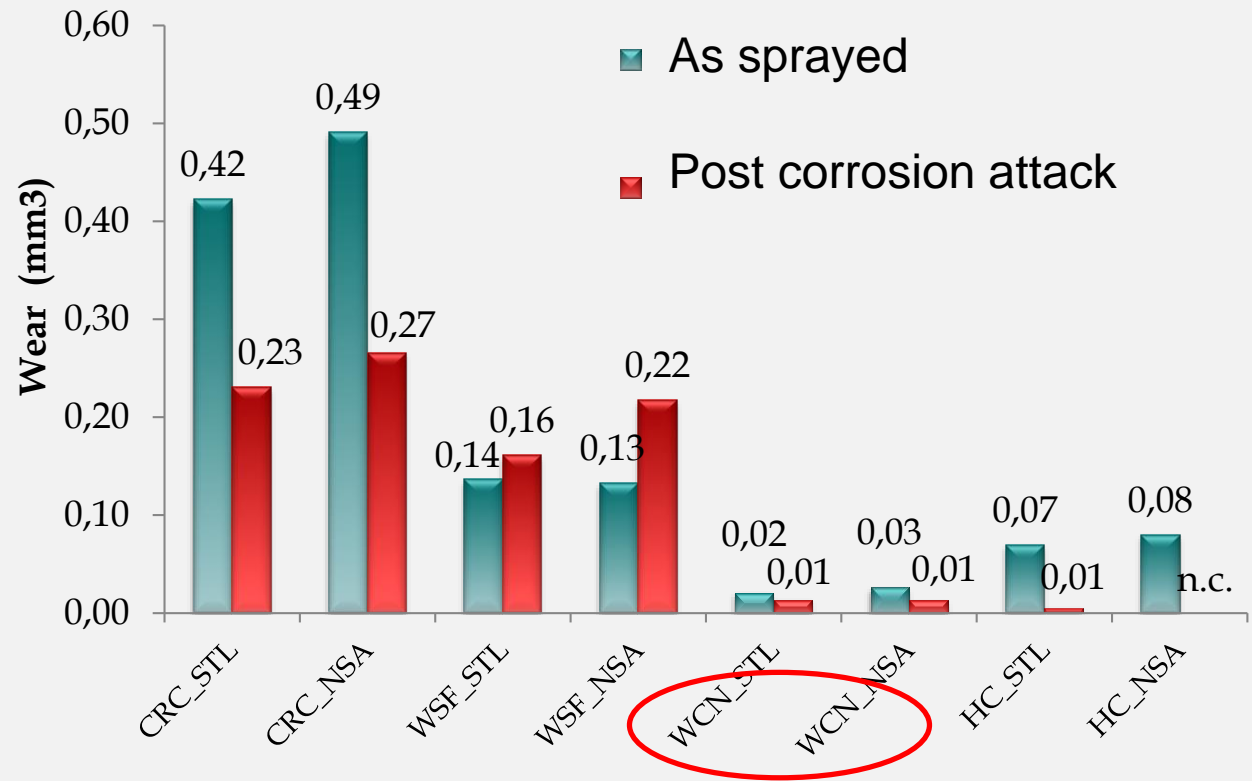
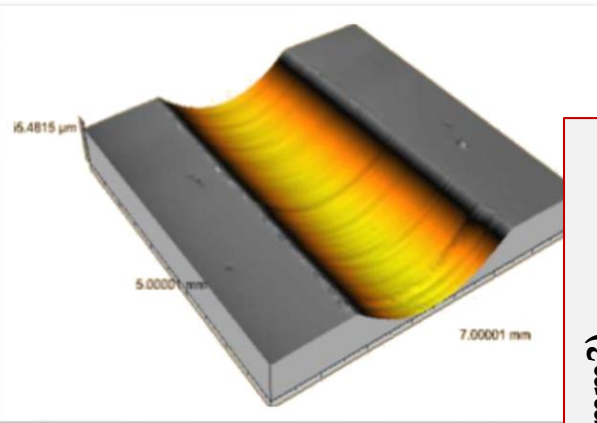
**No surface damage  
detectable**



## Phase 2- optimized coatings characterization

### Wear tests

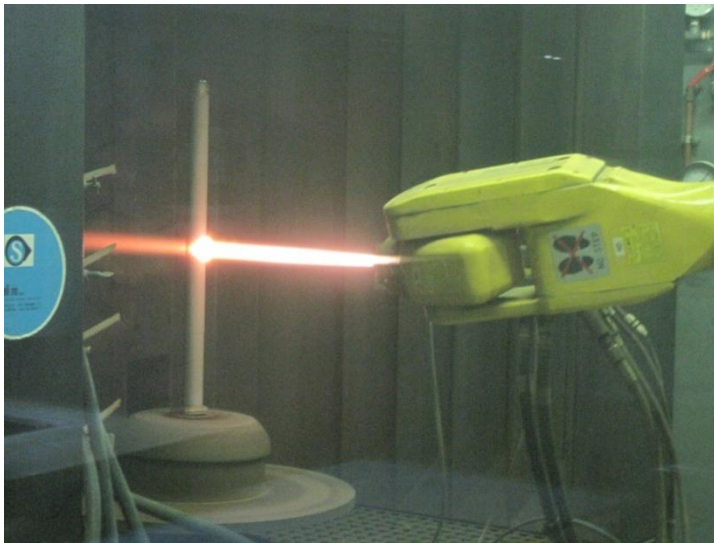
- Load: 91 N
- Sliding distance: 2000 m
- Speed: 1 m/s



## Phase 3- coating selection for valve deposition

### WC-CoCrNi:

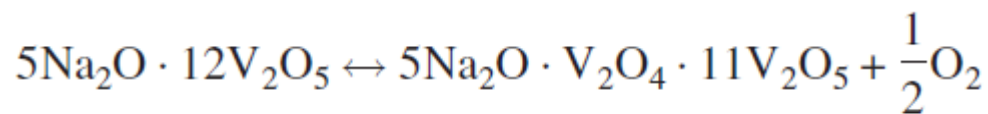
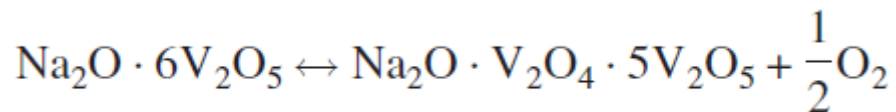
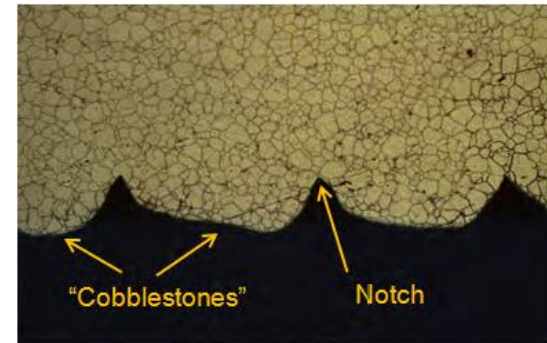
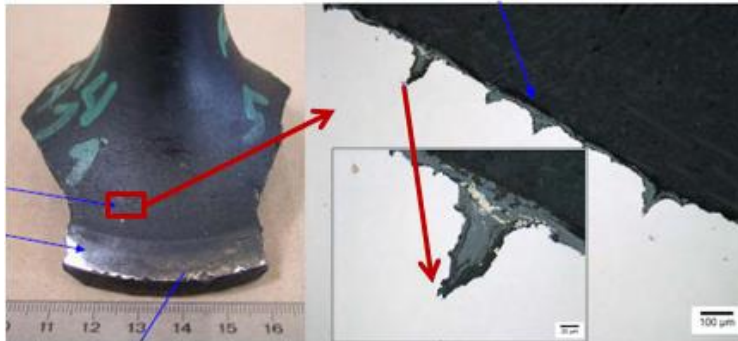
- Highest hardness
- Lowest wear rate
- Lowest corrosion rate
- Good density
- High deposition efficiency



- WC-CoCrNi coating was selected
- Coated valves tested for 3 years on a Wartsila test engine

# Innovative coatings for hot corrosion

## Case study: Hot corrosion on marine diesel engines valves



**Aggressive environment produced by high impurity contents within the fuel: V, Na, S.**

# Innovative coatings for hot corrosion

## Multiphase cermet coatings: CrystalCoat

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**VERLOTSKI**

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(54) **THERMALLY SPRAYED GASTIGHT  
PROTECTIVE LAYER FOR METAL  
SUBSTRATES**

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(DE)

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(57) **ABSTRACT**

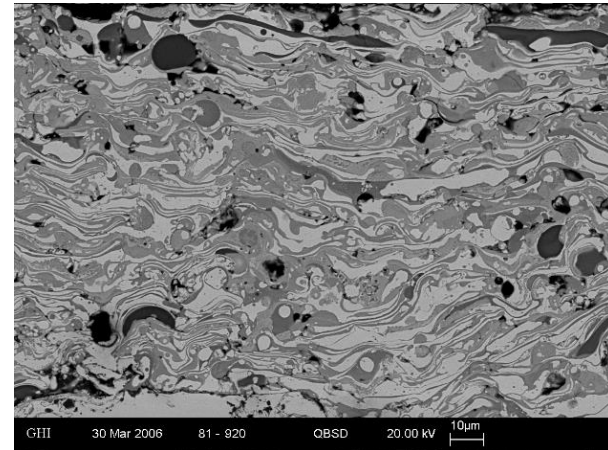
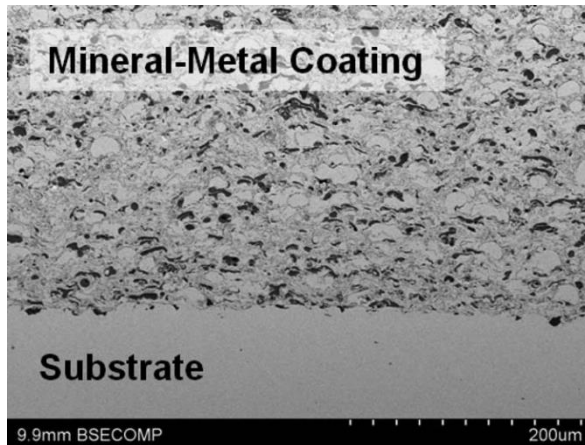
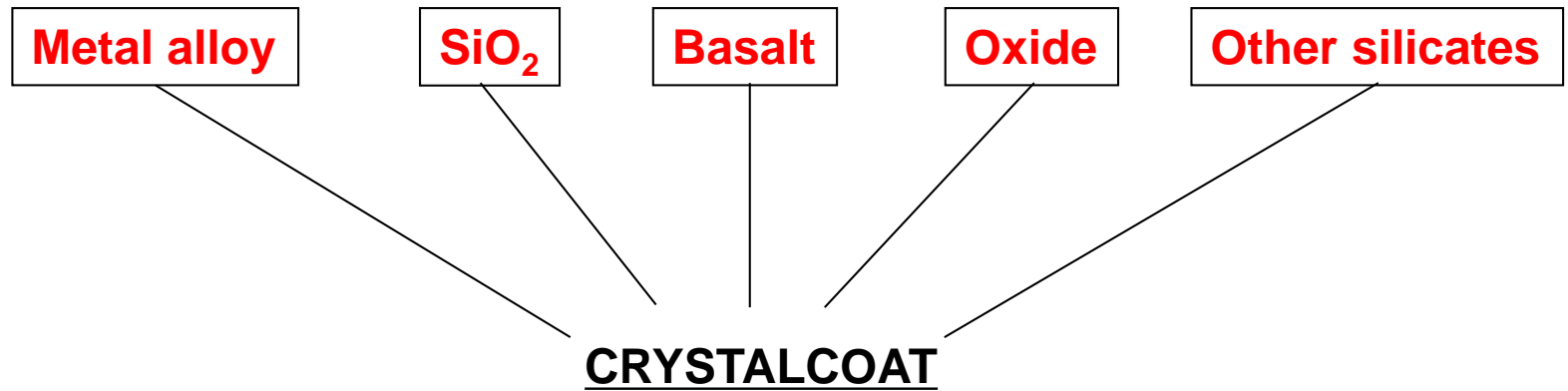
(73) Assignee: **MARKISCHES WERK GMBH**,  
Halver (DE)

In a thermally sprayed, gastight protective layer for metal substrates, especially those based on Fe, Ni, Al, Mg and/or Ti, wherein the spray powder for the purpose comprises at least



# Innovative coatings for hot corrosion

## Multiphase cermet coatings: CrystalCoat



# Innovative coatings for hot corrosion

Case study: Hot corrosion on marine diesel engines valves

**Aim: development of innovative thermal spray coatings alternative to Crystal Coat**



**Commercial  
(non standard)  
tested solutions**

- ✓  $\text{Cr}_3\text{C}_2 - \text{NiCr}$
- ✓  $\text{Cr}_3\text{C}_2 - \text{NiCrAlY}$
- ✓  $\text{Cr}_3\text{C}_2 - \text{CoNiCrAlY}$
- ✓  $\text{Cr}_3\text{C}_2 - \text{self fusing alloy}$

**Development of  
innovative  
solutions**

- ✓ **Mullite – nano  $\text{SiO}_2$  – NiCr**



# Innovative coatings for hot corrosion

## Phase 1- coatings optimization (DoE)

- Thickness higher than 150  $\mu\text{m}$
- Porosity as low as possible
- High deposition efficiency



Optimization of deposition  
parameters by  
*Design of Experiment (DoE)*

### 2 factors

1) Kerosene - O<sub>2</sub> flow

2) Spray distance

### 3 levels

1) Low

2) Medium

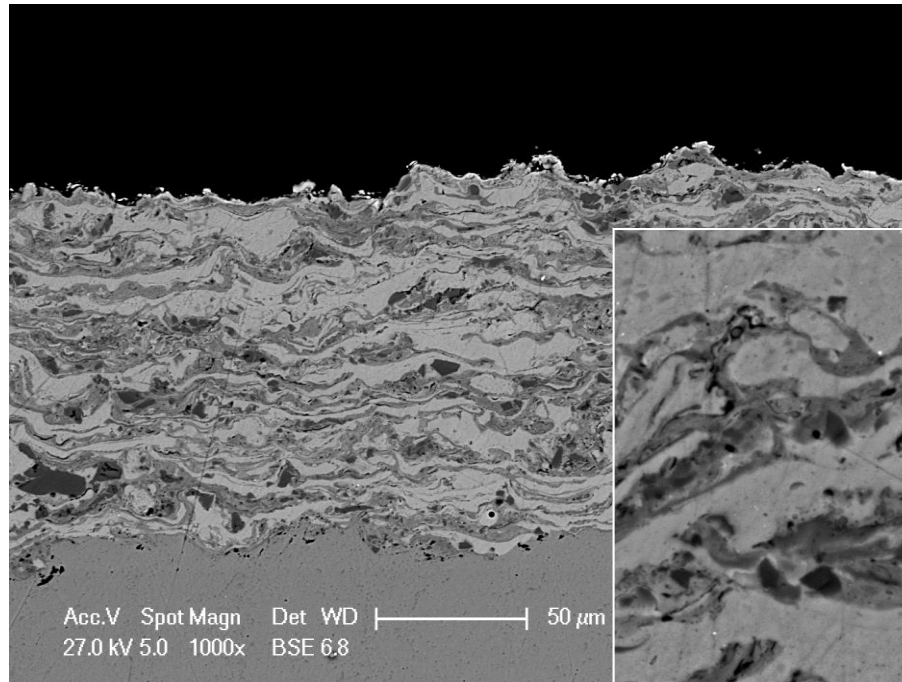
3) High

### 3<sup>2</sup> parameter sets

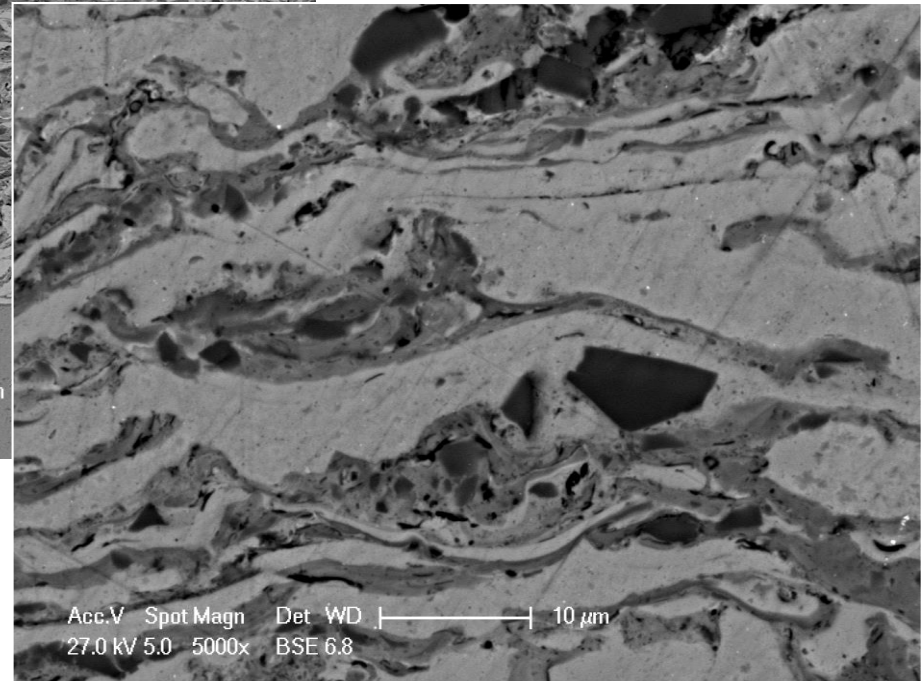
		Keros. (gph) – O <sub>2</sub> (scfh)		
		5,5-1700	6-1850	6,5-2000
Distance (mm)	320			
	355			
	380			

## Example: optimization of $\text{Cr}_3\text{C}_2$ - self fusing alloy

### Optimized coating



Very low porosity (< 1 %)

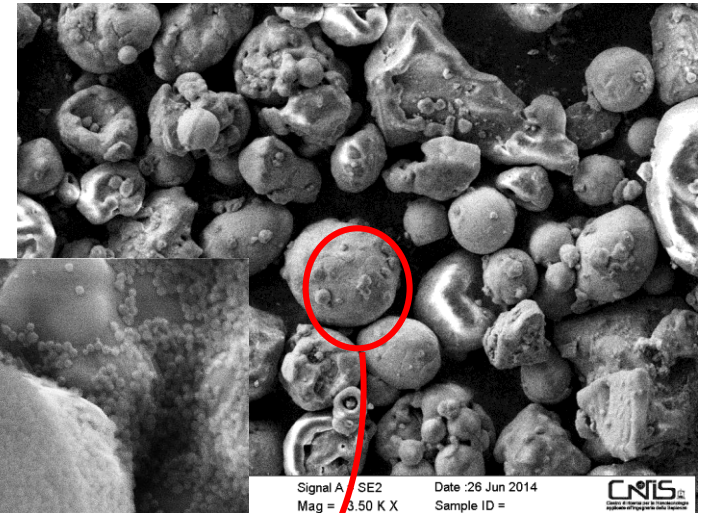
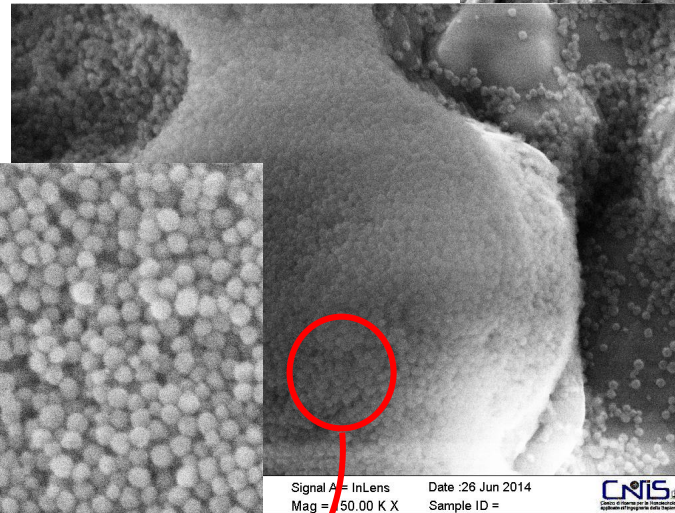
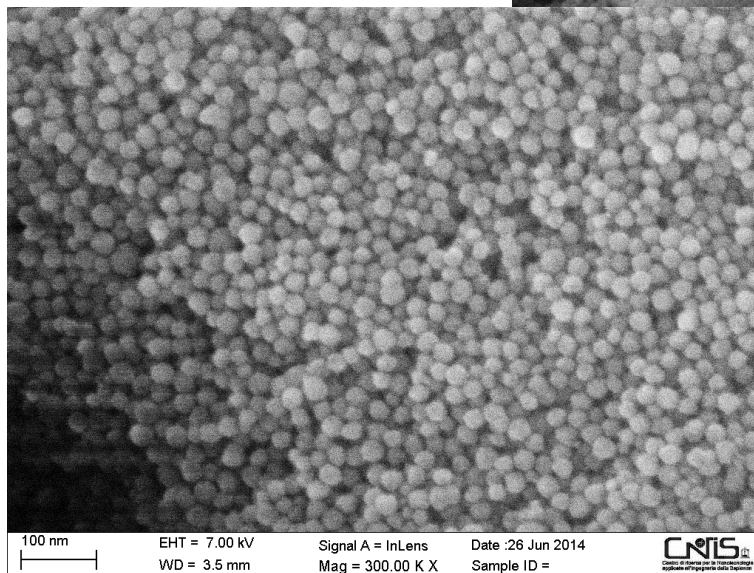


**Ceramic phase dispersed  
in multiphase metallic matrix**

# Innovative coatings for hot corrosion

✓ Mullite – nano  $\text{SiO}_2$  – NiCr

Powder agglomeration – spray drying



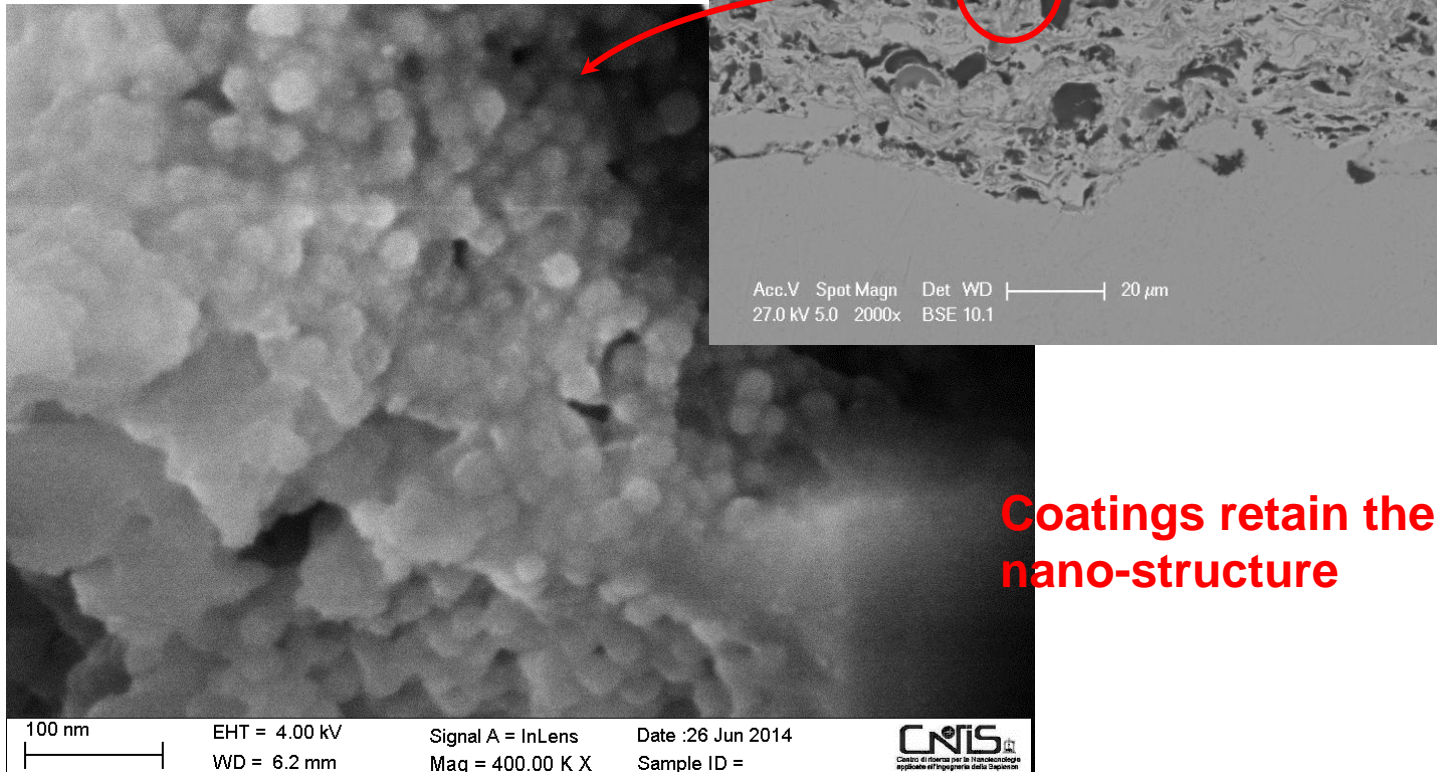
**Nano-SiO<sub>2</sub> cover completely  
the spray dried agglomerates**



# Innovative coatings for hot corrosion

✓ Mullite – nano  $\text{SiO}_2$  – NiCr

Coating deposition

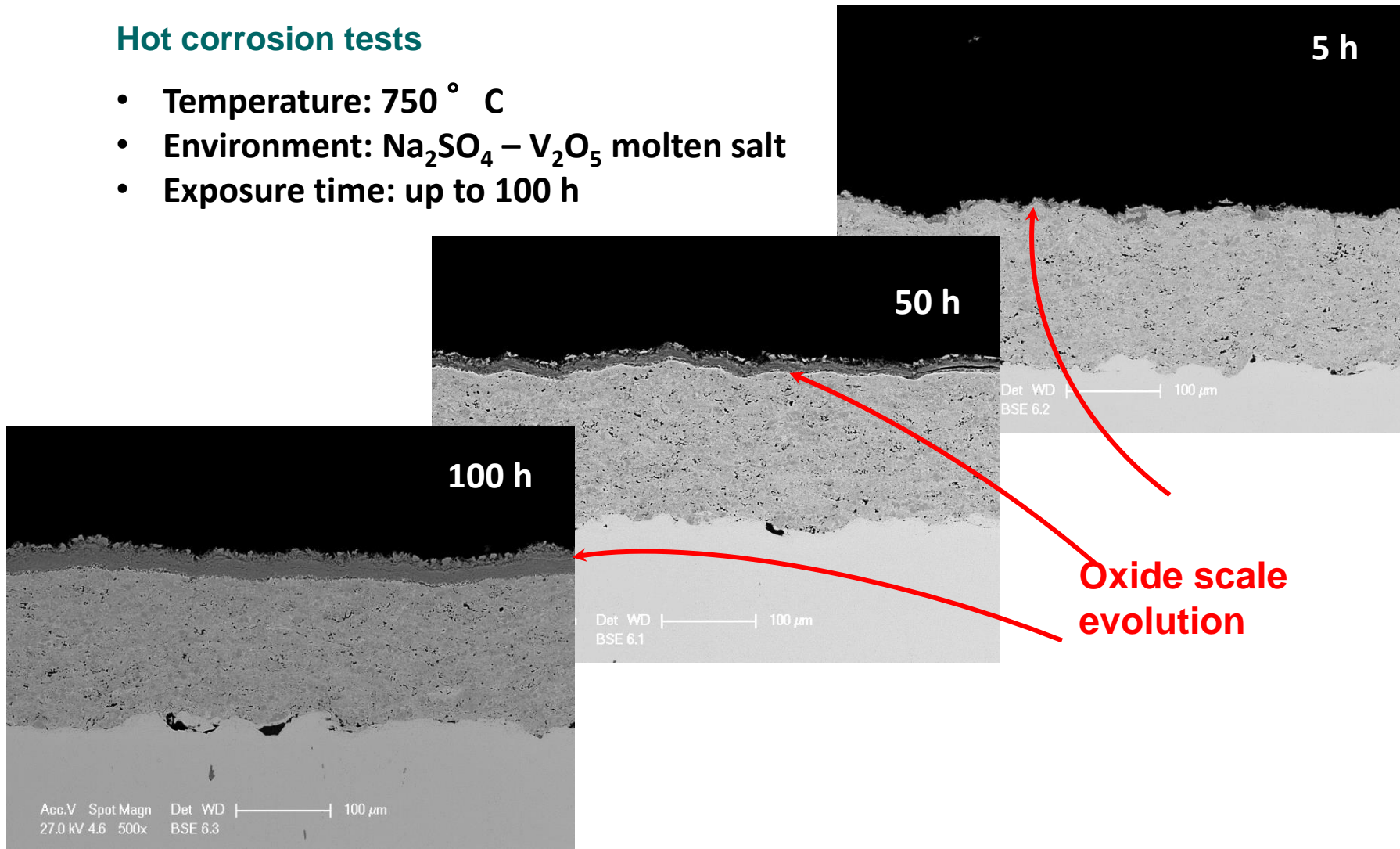


**Coatings retain the  
nano-structure**

# Innovative coatings for hot corrosion

## Hot corrosion tests

- Temperature: 750 ° C
- Environment:  $\text{Na}_2\text{SO}_4 - \text{V}_2\text{O}_5$  molten salt
- Exposure time: up to 100 h

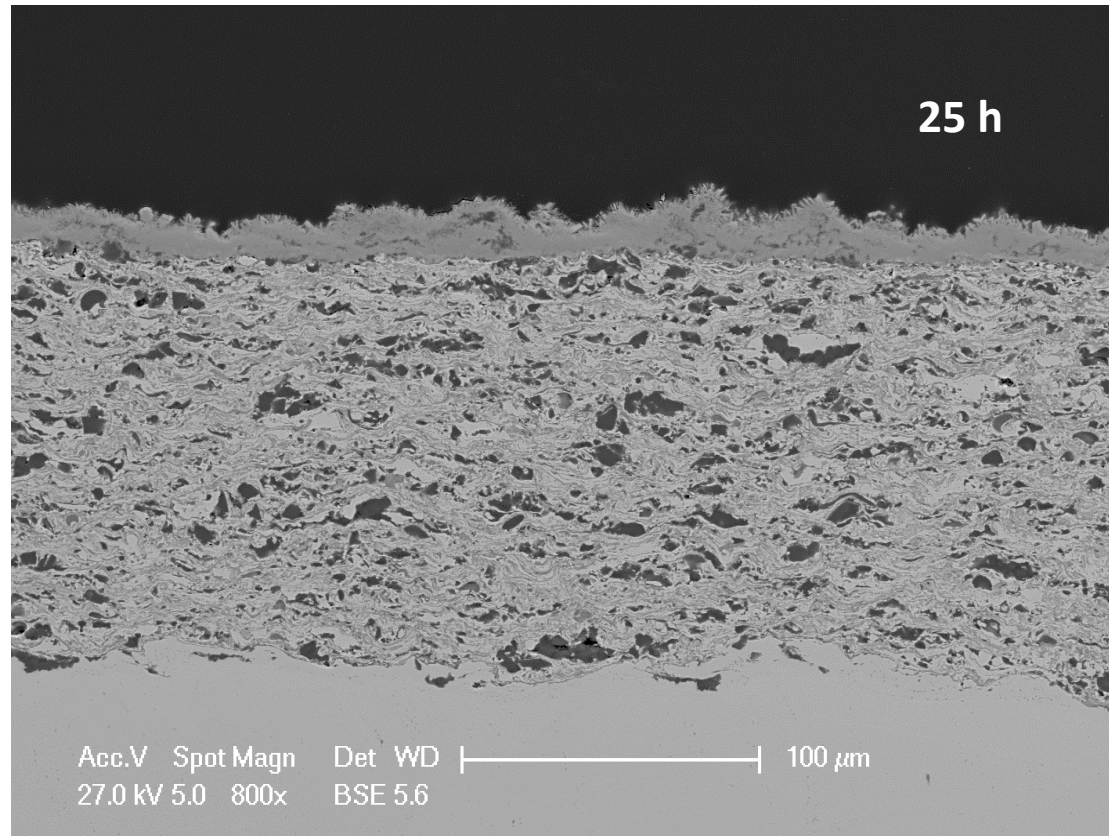




# Innovative coatings for hot corrosion

## Hot corrosion tests

### Nanostructured coating

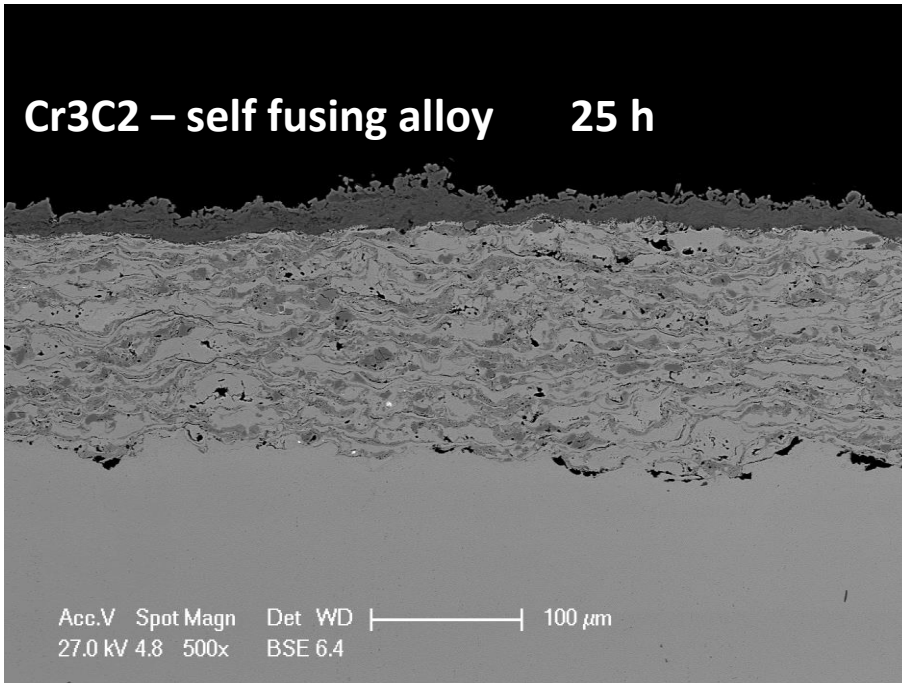


# Innovative coatings for hot corrosion

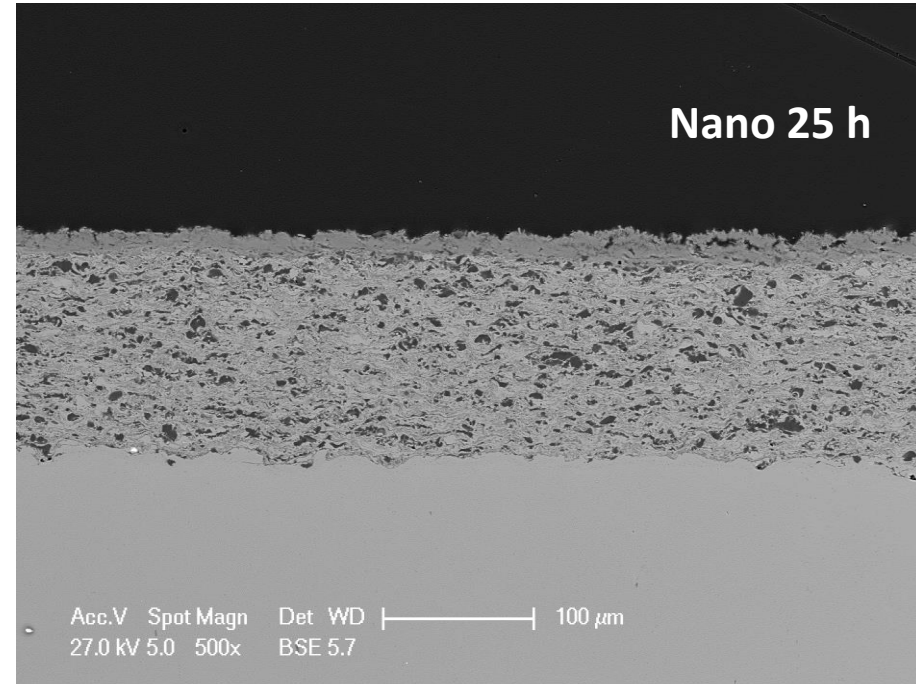
## Hot corrosion tests

Comparison between nanostructured and self fusing coatings

Cr<sub>3</sub>C<sub>2</sub> – self fusing alloy 25 h

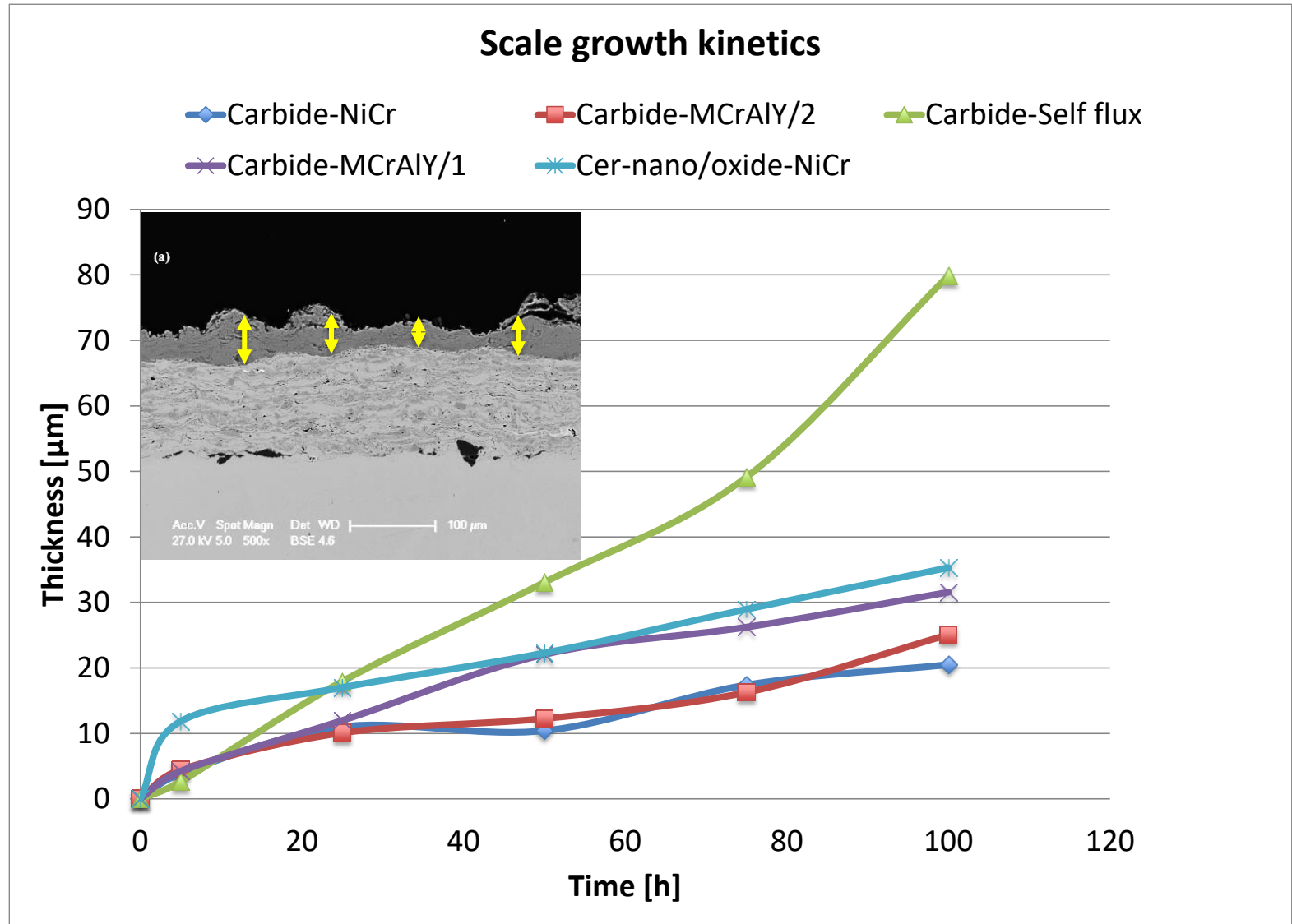


Nano 25 h



**Thinner oxide scale in nano-coating**

# Corrosion kinetics







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